

## SAFETY DATA SHEET

### 1. SUBSTANCE AND SOURCE IDENTIFICATION

**Product Identifier**

**SRM Number:** 3114  
**SRM Name:** Copper (Cu) Standard Solution  
**Other Means of Identification:** Not applicable.

**Recommended Use of This Material and Restrictions of Use**

This Standard Reference Material (SRM) is intended for use as a primary calibration standard for the quantitative determination of copper. A unit of SRM 3114 consists of five 10 mL sealed borosilicate glass ampoules of an acidified aqueous solution prepared gravimetrically to contain a known mass fraction of copper. The solution contains nitric acid at a volume fraction of approximately 10 %, equivalent to an amount-of-substance concentration (molarity) of approximately 1.6 mol/L.

**Company Information**

National Institute of Standards and Technology  
 Standard Reference Materials Program  
 100 Bureau Drive, Stop 2300  
 Gaithersburg, Maryland 20899-2300

Telephone: 301-975-2200  
 E-mail: SRMMSDS@nist.gov  
 Website: <https://www.nist.gov/srm>

Emergency Telephone ChemTrec:  
 1-800-424-9300 (North America)  
 +1-703-527-3887 (International)

### 2. HAZARDS IDENTIFICATION

**Classification**

**Physical Hazard:** Not classified.  
**Health Hazard:** Skin Corrosion/Irritation Category 1B  
 Serious Eye Damage/Eye Irritation Category 1

**Label Elements**

**Symbol**



**Signal Word**

DANGER

**Hazard Statement(s)**

H314 Causes severe skin burns and eye damage.

**Precautionary Statement(s)**

P260 Do not breathe mists, vapors, or spray.  
 P264 Wash hands thoroughly after handling.  
 P280 Wear protective gloves, protective clothing, face protection, and eye protection.  
 P301 + P330 + P331 If swallowed: Rinse mouth. Do NOT induce vomiting.  
 P303 + P361 + P353 If on skin (or hair): Remove immediately all contaminated clothing. Rinse skin with water/shower.  
 P305 + P351 + P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.  
 P363 Wash contaminated clothing before reuse.  
 P304 + P340 If inhaled: Remove person to fresh air and keep comfortable for breathing.  
 P310 Immediately call a poison center or doctor.  
 P405 Store locked up.

P501

Dispose of contents according to local regulations.

**Hazards Not Otherwise Classified:** Not applicable.

**Ingredients(s) with Unknown Acute Toxicity:** Not applicable.

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### 3. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

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**Substance:** Nitric Acid/Copper Nitrate Solution

**Other Designations:**

Nitric Acid (Aqua fortis; hydrogen nitrate; azotic acid; engraver's acid)

Copper Nitrate (Nitric acid, copper salt; copper dinitrate; cupric nitrate)

**NOTE:** Copper in nitric acid solution forms a solvated copper nitrate salt.

Components are listed in compliance with OSHA's 29 CFR 1910.1200; for the actual values see the Certificate of Analysis.

Hazardous Component(s)	CAS Number	EC Number (EINECS)	Nominal Mass Concentration (%)
Nitric Acid	7697-37-2	231-714-2	<20
Copper Nitrate	3251-23-8	221-838-5	2.9
Non-Hazardous Component(s)			
Water	7732-18-5	231-791-2	>80

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### 4. FIRST AID MEASURES

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**Description of First Aid Measures:**

**Inhalation:** If adverse effects occur, remove to uncontaminated area. If not breathing, give artificial respiration or oxygen by qualified personnel. Seek immediate medical attention.

**Skin Contact:** Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get immediate medical attention. Thoroughly clean and dry contaminated clothing before reuse. Destroy contaminated shoes.

**Eye Contact:** Immediately flush eyes, including under the eyelids with copious amounts of water for at least 15 minutes. Seek immediate medical attention.

**Ingestion:** Contact a poison control center immediately for instructions. Give water to rinse out mouth. Never give liquids to a person with reduced awareness or becoming unconscious. If vomiting occurs, keep head lower than hips to prevent aspiration. If not breathing, give artificial respiration by qualified personnel. Seek immediate medical attention.

**Most Important Symptoms/Effects, Acute and Delayed:** Acid burns to skin, eyes, and lungs.

**Indication of any immediate medical attention and special treatment needed, if necessary:** If any of the above symptoms are present, seek immediate medical attention.

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### 5. FIRE FIGHTING MEASURES

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**Fire and Explosion Hazards:** Negligible fire hazard. See Section 9, "Physical and Chemical Properties" for flammability properties.

**Extinguishing Media:**

Suitable: Use extinguishing media appropriate to the surrounding fire.

Unsuitable: None listed.

**Specific Hazards Arising from the Chemical:** Thermal decomposition will form oxides of nitrogen.

**Special Protective Equipment and Precautions for Fire-Fighters:** Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH approved self-contained breathing apparatus (SCBA).

**NFPA Ratings** (0 = Minimal; 1 = Slight; 2 = Moderate; 3 = Serious; 4 = Severe)

Health = 3

Fire = 0

Reactivity = 0

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## 6. ACCIDENTAL RELEASE MEASURES

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**Personal Precautions, Protective Equipment and Emergency Procedures:** Immediately contact emergency personnel. Keep unnecessary personnel away. Use suitable protective equipment; see Section 8, “Exposure Controls and Personal Protection”.

**Methods and Materials for Containment and Clean up:** Do not touch spilled material. Notify safety personnel of spills. Absorb with sand or other non-combustible material. Collect spilled material in appropriate container for disposal. Isolate hazard area and deny entry.

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## 7. HANDLING AND STORAGE

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**Safe Handling Precautions:** See Section 8, “Exposure Controls and Personal Protection”. Handle glass ampoules with care.

**Storage:** Store and handling in accordance with all current regulations and standards. Keep separated from incompatible substances (acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals, halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides).

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## 8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

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### Exposure Limits:

**Component:** Nitric Acid

NIOSH (REL):	5 mg/m <sup>3</sup> (2 ppm; TWA)
	10 mg/m <sup>3</sup> (4 ppm; STEL)
	65 mg/m <sup>3</sup> (25 ppm; IDLH)
ACGIH (TLV):	5 mg/m <sup>3</sup> (2 ppm; TWA)
	10 mg/m <sup>3</sup> (4 ppm; STEL)
OSHA (PEL):	5 mg/m <sup>3</sup>

**Component:** Copper Nitrate (as Cu, dust and mist, related to copper compounds).

NIOSH (REL):	1 mg/m <sup>3</sup> (TWA)
	100 mg/m <sup>3</sup> (IDLH)
ACGIH (TLV):	1 mg/m <sup>3</sup> (TWA)
OSHA (PEL):	No occupational limits established.

**Engineering Controls:** Provide local exhaust or process enclosure ventilation system. Ensure compliance with applicable exposure limits.

**Personal Protection:** In accordance with OSHA 29 CFR 1910.132, subpart I, wear appropriate Personal Protective Equipment (PPE) to minimize exposure to this material.

**Respiratory Protection:** If workplace conditions warrant a respirator, a respiratory protection program that meets OSHA 29CFR 1910.134 must be followed. Refer to NIOSH 42 CFR 84 for applicable certified respirators.

**Eye/Face Protection:** Wear splash resistant safety goggles with a face shield. An eye wash station should be readily available near areas of use.

**Skin and Body Protection:** Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product. Chemical-resistant gloves should be worn at all times when handling chemicals.

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## 9. PHYSICAL AND CHEMICAL PROPERTIES

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### Descriptive Properties:

Appearance (physical state, color, etc.):	not available
Molecular Formula:	not applicable
Molar Mass (g/mol):	not applicable
Odor:	irritating odor
Odor threshold:	not available
pH:	acidic
Evaporation rate:	not available
Melting point/freezing point (°C):	not available
Relative Density (g/L):	not available
Vapor Pressure (mmHg):	not available
Vapor Density (air = 1):	not available
Viscosity (cP):	not available
Solubility(ies):	miscible with water
Partition coefficient (n-octanol/water):	not available

### Thermal Stability Properties:

Autoignition Temperature (°C):	not applicable
Thermal Decomposition (°C):	not applicable
Initial boiling point and boiling range (°C):	not available
Explosive Limits, LEL (Volume %):	not applicable
Explosive Limits, UEL (Volume %):	not applicable
Flash Point (°C):	not applicable
Flammability (solid, gas):	not applicable

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## 10. STABILITY AND REACTIVITY

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**Reactivity:** Stable at room temperatures and pressure.

**Stability:**  Stable  Unstable

**Possible Hazardous Reactions:** None listed.

**Conditions to Avoid:** Contact with combustible or incompatible materials.

**Incompatible Materials:** Acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals, halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides.

**Fire/Explosion Information:** See Section 5, "Fire Fighting Measures".

**Hazardous Decomposition:** Thermal decomposition will produce oxides of nitrogen.

**Hazardous Polymerization:**  Will Occur  Will Not Occur

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## 11. TOXICOLOGICAL INFORMATION

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**Route of Exposure:**  Inhalation  Skin  Ingestion

**Symptoms Related to the Physical, Chemical and Toxicological Characteristics:** Burning pain and severe skin corrosion, and eye damage.

### Potential Health Effects (Acute, Chronic and Delayed):

**Inhalation:** Inhalation of nitric acid can damage the mucous membranes and upper respiratory tract. Short term exposure may cause irritation and inflammation of the upper respiratory tract, coughing, choking, sore throat, shortness of breath, headache, dizziness, and nausea. Long term exposure to acid fumes may cause damage to

teeth, bronchial irritation, chronic cough, bronchial pneumonia, and gastrointestinal disturbances. Mists of copper salts may cause irritation of the upper respiratory tract. Chronic exposure to mists of copper salts may cause rhinitis, sneezing, coughing, fever, metallic tastes and digestive disorders.

**Skin Contact:** Nitric acid can cause severe skin burns. Severity of the damage depends on the concentration and duration of exposure. Effects of acid burns may be delayed. Copper nitrate may cause severe irritation, redness, pain, and possibly burns.

**Eye Contact:** Nitric acid can cause severe eye irritation, corneal burns, permanent eye damage, or blindness. Severity of the damage depends on the concentration and duration of exposure. Copper nitrate acute and chronic exposure may cause severe irritation and conjunctivitis.

**Ingestion:** If ingested, nitric acid can cause severe burns and damage to the gastrointestinal tract. Copper salts may cause an immediate metallic taste and chronic ingestion may cause anemia.

#### Numerical Measures of Toxicity:

**Acute Toxicity:** Not classified.

Nitric acid, Rat, Inhalation LC50: 130 mg/m<sup>3</sup> (4 h)

Copper Nitrate, Rat, Oral LD50: 794 mg/kg

**Skin Corrosion/Irritation:** This SRM contains >1 % of nitric acid and it is classified as Category 1B.

**Serious Eye damage/ Eye irritation:** This SRM contains >1 % nitric acid and it is classified as Category 1.

Copper Nitrate, Rabbit, Eyes (severe): 100 mg (4 s)

**Respiratory Sensitization:** No data available.

**Skin Sensitization:** No data available.

**Germ Cell Mutagenicity:** No data available.

**Carcinogenicity:** Not classified.

**Listed as a Carcinogen/Potential Carcinogen** \_\_\_\_\_ Yes                        X   No

Nitric acid is not listed in NTP, IARC or OSHA as a carcinogen.

Copper nitrate is not listed in NTP, IARC or OSHA as a carcinogen.

**Reproductive Toxicity:** Not classified.

Nitric acid, Rat, Oral TDLo: 21 150 mg/kg (pregnant 1-21 days)

Nitric acid, Rat, Oral TDLo: 2345 mg/kg (pregnant 18 days)

**Specific Target Organ Toxicity, Single Exposure:** Not classified.

**Specific Target Organ Toxicity, Repeated Exposure:** Not classified.

**Aspiration Hazard:** No data available.

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## 12. ECOLOGICAL INFORMATION

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#### Ecotoxicity Data:

Nitric Acid: Mosquitofish (*Gambusia affinis*), LC50: 72 mg/L (96 h)

Copper Nitrate: Coho salmon (*Oncorhynchus kisutch*), LC50: 286 µg/L (96 h).

**Persistence and Degradability:** No data available.

**Bioaccumulative Potential:** No data available.

**Mobility in Soil:** No data available.

**Other Adverse effects:** No data available.

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## 13. DISPOSAL CONSIDERATIONS

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**Waste Disposal:** Dispose of waste in accordance with all applicable federal, state, and local regulations. Nitric acid waste is subject to disposal regulations: U.S. EPA 40 CFR 262, Hazardous Waste Numbers: D001, D002.

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## 14. TRANSPORTATION INFORMATION

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**U.S. DOT and IATA:** UN1760, Corrosive liquid, n.o.s. (contains nitric acid), Hazard Class 8, Packing Group II, Excepted Quantities E2.

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## 15. REGULATORY INFORMATION

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### U.S. Regulations:

CERCLA Sections 102a/103 (40 CFR 302.4):

Nitric Acid, 1000 lbs; 454 kg RQ. Copper Nitrate, 100 lbs; 45.4 kg RQ.

SARA Title III Section 302 (40 CFR 355.30): Nitric Acid, 1000 lbs TPQ.

SARA Title III Section 304 (40 CFR 355.40): Nitric Acid, 1000 lbs EPCRA RQ.

SARA Title III Section 313 (40 CFR 372.65):

Nitric Acid: 1 % de minimis concentration. Copper Nitrate: 1 % de minimis concentration.

OSHA Process Safety (29 CFR 1910.119):

Regulated for nitric acid at higher concentrations 500 lb TQ ( $\geq 94.5$  % by weight).

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE HEALTH: Yes.

CHRONIC HEALTH: No.

FIRE: No.

REACTIVE: No.

PRESSURE: No.

### State Regulations:

California Proposition 65: Not listed.

**U.S. TSCA Inventory:** Nitric acid and copper nitrate are listed.

**TSCA 12(b), Export Notification:** Not listed.

**Canadian Regulations:** WHMIS Information is not provided for this material.

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## 16. OTHER INFORMATION

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**Issue Date:** 10 February 2021

**Sources:** ChemAdvisor, Inc., MSDS *Nitric Acid*, 09 December 2015.

ChemAdvisor, Inc., MSDS *Cupric Nitrate*, 09 December 2015.

CDC; NIOSH; *NIOSH Pocket Guide to Chemical Hazards*; Department of Health and Human Services (DHHS), Centers for Disease Control and Prevention (CDC), National Institute for Safety and Health; *Nitric Acid*, 30 October 2019; available at <https://www.cdc.gov/niosh/npg/npgd0447.html> (accessed Feb 2021).

PubChem, National Library of Medicine, *Copper Nitrate CAS 3251-23-8*, available at <https://pubchem.ncbi.nlm.nih.gov/> (accessed Feb 2021).

## Key of Acronyms:

ACGIH	American Conference of Governmental Industrial Hygienists	NRC	Nuclear Regulatory Commission
ALI	Annual Limit on Intake	NTP	National Toxicology Program
CAS	Chemical Abstracts Service	OSHA	Occupational Safety and Health Administration
CERCLA	Comprehensive Environmental Response, Compensation, and Liability Act	PEL	Permissible Exposure Level
CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
DOT	Department of Transportation	REL	Recommended Exposure Limit
EC50	Effective Concentration, 50%	RM	Reference Material
EINECS	European Inventory of Existing Commercial Chemical Substances	RQ	Reportable Quantity
EPCRA	Emergency Planning and Community Right-to-Know Act	RTECS	Registry of Toxic Effects of Chemical Substances
IARC	International Agency for Research on Cancer	SARA	Superfund Amendments and Reauthorization Act
IATA	International Air Transport Association	SCBA	Self-Contained Breathing Apparatus
IDLH	Immediately Dangerous to Life and Health	SRM	Standard Reference Material
LC50	Lethal Concentration, 50 %	STEL	Short Term Exposure Limit
LD50	Lethal Dose, 50 %	TLV	Threshold Limit Value
LEL	Lower Explosive Limit	TPQ	Threshold Planning Quantity
MSDS	Material Safety Data Sheet	TSCA	Toxic Substances Control Act
NFPA	National Fire Protection Association	TWA	Time Weighted Average
NIOSH	National Institute for Occupational Safety and Health	UEL	Upper Explosive Limit
NIST	National Institute of Standards and Technology	WHMIS	Workplace Hazardous Materials Information System
n.o.s.	Not Otherwise Specified		

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