

# SAFETY DATA SHEET

# **1. SUBSTANCE AND SOURCE IDENTIFICATION**

#### **Product Identifier**

SRM Number:3109aSRM Name:Calcium (Ca) Standard SolutionOther Means of Identification:Not applicable.

#### **Recommended Use of This Material and Restrictions of Use**

This Standard Reference Material (SRM) is intended for use as a primary calibration standard for the quantitative determination of calcium. A unit of SRM 3109a consists of five 10 mL sealed borosilicate glass ampoules of an acidified aqueous solution prepared gravimetrically to contain a known mass fraction of calcium.

#### **Company Information**

National Institute of Standards and Technology Standard Reference Materials Program 100 Bureau Drive, Stop 2300 Gaithersburg, Maryland 20899-2300

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# 2. HAZARDS IDENTIFICATION

#### Classification

Physical Hazard:	Not classified.	
Health Hazard:	Skin Corrosion/Irritation	Category
	Serious Eye Damage/Eye Irritation	Category

Label Elements Symbol



Signal Word DANGER

# Hazard Statement(s)H314Causes severe skin burns and eye damage.

ient(s)
Do not breathe fume, mists, vapors, spray.
Wash hands thoroughly after handling.
Wear protective gloves, protective clothing, eye protection.
If swallowed: Rinse mouth. Do NOT induce vomiting.
If on skin (or hair): Remove immediately all contaminated clothing. Rinse skin with
water.
If inhaled: Remove person to fresh air and keep comfortable for breathing.
If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
Immediately call a poison center or doctor.
Wash contaminated clothing before reuse.

P390 Absorb spillage to prevent material damage.

P405 Store locked up.

P501 Dispose of contents and container according to local regulations.

Hazards Not Otherwise Classified: Not applicable.

Ingredients(s) with Unknown Acute Toxicity: Not applicable.

## 3. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Substance: Calcium in nitric acid solution

#### **Other Designations:**

Nitric acid (aqua fortis; hydrogen nitrate; azotic acid; engraver's acid) Calcium nitrate (nitric acid, calcium salt; Norway saltpeter; lime nitrate)

**NOTE:** Calcium in nitric acid solution forms a solvated calcium nitrate salt. The health and physical hazard information provided in this SDS are for nitric acid and calcium nitrate. No physical or chemical data are listed for this solution. The actual effects of the solution may differ from the individual components.

Components are listed in compliance with OSHA 29 CFR 1910.1200; for the actual values see the NIST Certificate of Analysis.

Hazardous Component(s)	CAS Number	EC Number (EINECS)	Nominal Mass Concentration (%)
Nitric acid	7697-37-2	231-714-2	<20
Calcium nitrate	10124-37-5	233-332-1	1
<b>Non-Hazardous Component(s)</b> Water	7732-18-5	231-791-2	>79

# 4. FIRST AID MEASURES

#### **Description of First Aid Measures:**

**Inhalation:** If adverse effects occur, remove to uncontaminated area. If not breathing, give artificial respiration or oxygen by qualified personnel. Seek immediate medical attention.

**Skin Contact:** Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get immediate medical attention. Thoroughly clean and dry contaminated clothing before reuse. Destroy contaminated shoes.

**Eye Contact:** Immediately flush eyes, including under the eyelids with copious amounts of water for at least 15 minutes. Seek immediate medical attention.

**Ingestion:** Contact a poison control center immediately for instructions. Do not induce vomiting. Give water to rinse out mouth. Never give liquids to a person with reduced awareness or becoming unconscious. If vomiting occurs, keep head lower than hips to prevent aspiration. If not breathing, give artificial respiration by qualified personnel. Seek immediate medical attention.

Most Important Symptoms/Effects, Acute and Delayed: Acid burns to skin, eyes, and lungs.

**Indication of any immediate medical attention and special treatment needed, if necessary:** If any of the above symptoms are present, seek immediate medical attention.

# **5.** FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Negligible fire hazard. See Section 9, "Physical and Chemical Properties" for flammability properties.

#### **Extinguishing Media:**

Suitable: Use extinguishing media appropriate to the surrounding fire. Unsuitable: None listed.

Specific Hazards Arising from the Chemical: Thermal decomposition will form oxides of nitrogen and calcium.

**Special Protective Equipment and Precautions for Fire-Fighters:** Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH approved self-contained breathing apparatus (SCBA).

**NFPA Ratings** (0 = Minimal; 1 = Slight; 2 = Moderate; 3 = Serious; 4 = Severe)

Health = 3 Fire = 0 Reactivity = 0

# 6. ACCIDENTAL RELEASE MEASURES

**Personal Precautions, Protective Equipment and Emergency Procedures:** Immediately contact emergency personnel. Keep unnecessary personnel away. Use suitable protective equipment; see Section 8, "Exposure Controls and Personal Protection".

**Methods and Materials for Containment and Clean up:** Do not touch spilled material. Notify safety personnel of spills. Absorb with sand or other non-combustible material. Collect spilled material in appropriate container for disposal. Isolate hazard area and deny entry.

# 7. HANDLING AND STORAGE

**Safe Handling Precautions:** See Section 8, "Exposure Controls and Personal Protection". Handle glass ampoules with care.

**Storage:** Store and handling in accordance with all current regulations and standards. Keep separated from incompatible substances (acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals, halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides).

# 8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

**Exposure Limits** 

Nitric acid

NIOSH (REL): 5 mg/m<sup>3</sup> (2 ppm) TWA 10 mg/m<sup>3</sup> (4 ppm) STEL 65 mg/m<sup>3</sup> (25 ppm) IDLH ACGIH (TLV): 5 mg/m<sup>3</sup> (2 ppm) TWA 10 mg/m<sup>3</sup> (4 ppm) STEL

OSHA (PEL): 5 mg/m<sup>3</sup> (2 ppm) TWA

Calcium nitrate: No occupational exposure limits available.

**Engineering Controls:** Provide local exhaust or process enclosure ventilation system. Ensure compliance with applicable exposure limits.

**Personal Protection:** In accordance with OSHA 29 CFR 1910.132, subpart I, wear appropriate Personal Protective Equipment (PPE) to minimize exposure to this material.

**Respiratory Protection:** If workplace conditions warrant a respirator, a respiratory protection program that meets OSHA 29CFR 1910.134 must be followed. Refer to NIOSH 42 CFR 84 for applicable certified respirators.

**Eye/Face Protection:** Wear splash resistant safety goggles with a face shield. An eyewash station should be readily available near areas of use.

**Skin and Body Protection:** Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product. Chemical-resistant gloves should be worn at all times when handling chemicals.

# 9. PHYSICAL AND CHEMICAL PROPERTIES

Descriptive Properties:	
Appearance (physical state, color, etc.):	colorless to yellow liquid
Molecular Formula	not applicable
Molar Mass (g/mol)	not applicable
Odor	irritating odor
Odor threshold	not available
рН	acidic
Evaporation rate	not available
Melting point/freezing point	not available
<b>Relative Density (g/L)</b> as specific gravity (water = 1):	not available
Vapor Pressure (mmHg)	not available
Vapor Density (air = 1)	not available
Viscosity (cP)	not available
Solubility(ies)	soluble with water and some alcohols
Partition coefficient (n-octanol/water)	not available
Particle Size (if relevant)	not applicable
Thermal Stability Properties	
Autoignition Temperature	not applicable
Thermal Decomposition	not applicable
Initial boiling point and boiling range	not available
Explosive Limits, LEL (Volume %)	not applicable
Explosive Limits, UEL (Volume %)	not applicable
Flash Point	not applicable
Flammability (solid, gas)	not applicable

# **10. STABILITY AND REACTIVITY**

Reactivity: Stable at normal temperatures and pressure.

 Stability:
 X
 Stable
 Unstable

Possible Hazardous Reactions: None listed.

Conditions to Avoid: Contact with combustible or incompatible materials.

**Incompatible Materials:** Acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals, halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides.

Fire/Explosion Information: See Section 5, "Fire Fighting Measures".

Hazardous Decomposition: Thermal decomposition will produce oxides of nitrogen and calcium.

Hazardous Polymerization: \_\_\_\_\_ Will Occur \_\_\_\_ X \_\_\_ Will Not Occur

# **11. TOXICOLOGICAL INFORMATION**

Route of Exposure: X Inhalation X Skin X Ingestion

Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Burning pain and severe skin corrosion, eye and lung damage.

Potential Health Effects (Acute, Chronic and Delayed):

**Inhalation:** Inhalation of nitric acid can damage the mucous membranes and upper respiratory tract. Short term exposure may cause irritation and inflammation of the upper respiratory tract, coughing, choking, sore throat, shortness of breath, headache, dizziness, and nausea. Long term exposure to acid fumes may cause damage to teeth, bronchial irritation, chronic cough, bronchial pneumonia, and gastrointestinal disturbances. Calcium nitrate may cause respiratory tract irritation.

**Skin Contact:** Nitric acid can cause severe skin burns. Severity of the damage depends on the concentration and duration of exposure. Effects of acid burns may be delayed. Calcium nitrate may cause irritation.

**Eye Contact:** Nitric acid can cause severe eye irritation, corneal burns, permanent eye damage, or blindness. Severity of the damage depends on the concentration and duration of exposure. Contact with calcium nitrate may cause irritation and conjunctivitis.

**Ingestion:** If ingested, nitric acid can cause severe burns and damage to the gastrointestinal tract. Ingestion of large amounts of calcium nitrate may cause gastric irritation.

#### Numerical Measures of Toxicity:

Acute Toxicity: Not classified.

Nitric acid: Rat, Inhalation LC50: 130 mg/m<sup>3</sup> (4 h)

Calcium nitrate: Rat, Oral LD50: 302 mg/kg

- Skin Corrosion/Irritation: This SRM contains >1 % of nitric acid and it is classified as Category 1B. Calcium nitrate: No data available.
- Serious Eye damage/Eye irritation: This SRM contains >1 % nitric acid and it is classified as Category 1. Calcium nitrate: No data available.

Respiratory Sensitization: No data available.

Skin Sensitization: No data available. Calcium nitrate: No data available.

Germ Cell Mutagenicity: No data available.

Carcinogenicity: Not classified.

Listed as a Carcinogen/Potential Carcinogen Yes X No Nitric acid and calcium nitrate are not listed by NTP, IARC or OSHA as carcinogens.

Reproductive Toxicity: Not classified.

Nitric acid, Rat, Oral TDLo: 21 150 mg/kg (pregnant 1 d to 21 d)

Nitric acid, Rat, Oral TDLo: 2345 mg/kg (pregnant 18 d)

Calcium nitrate: No data available.

Specific Target Organ Toxicity, Single Exposure: No data available.

Specific Target Organ Toxicity, Repeated Exposure: No data available.

Aspiration Hazard: No data available.

# **12.** ECOLOGICAL INFORMATION

#### **Ecotoxicity Data**

Nitric acid: mosquitofish (*Gambusia affinis*) LC50: 72 mg/L (96 h) Calcium Nitrate: bluegill fish (*Lepomis macrochirus*) LC50: 10000 mg/L [static] (96 h)

Persistence and Degradability: No data available.

**Bioaccumulative Potential:** No data available.

Mobility in Soil: No data available.

Other Adverse effects: No data available.

# **13. DISPOSAL CONSIDERATIONS**

**Waste Disposal:** Dispose of waste in accordance with all applicable federal, state, and local regulations. Nitric acid subject to disposal regulations: U.S. EPA 40 CFR 262, Hazardous Waste Numbers: D001, D002.

# 14. TRANSPORTATION INFORMATION

**U.S. DOT and IATA:** UN1760, Corrosive liquid, n.o.s. (contains nitric acid), Hazard Class 8, Packing Group II, Excepted Quantities E2.

# **15. REGULATORY INFORMATION**

#### **U.S. Regulations:**

CERCLA Sections 102a/103 (40 CFR 302.4):	Nitric acid, 1000 lbs. (454 kg) RQ
SARA Title III Section 302 (40 CFR 355.30):	Nitric acid, 1000 lbs. (454 kg) TPQ
SARA Title III Section 304 (40 CFR 355.40):	Nitric acid, 1000 lbs. (454 kg) EPCRA RQ
SARA Title III Section 313 (40 CFR 372.65):	Nitric acid, 1 % de minimis concentration
OSHA Process Safety (29 CFR 1910.119):	Regulated for nitric acid at higher concentrations 500 lbs. TQ (≥94.5 % by weight).

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE HEALTH:	Yes
CHRONIC HEALTH:	No
FIRE:	No
REACTIVE:	No
PRESSURE:	No

State Regulations: Not listed under California Proposition 65.

U.S. TSCA Inventory: Nitric acid and calcium nitrate are listed.

TSCA 12(b), Export Notification: Not listed.

Canadian Regulations: WHMIS Information is not provided for this material.

# **16. OTHER INFORMATION**

Issue Date: 21 May 2021

Sources: ChemADVISOR, Inc., SDS Nitric Acid, 09 December 2015.

ChemADVISOR, Inc., SDS Calcium Nitrate, 11 September 2013.

CDC; NIOSH; *NIOSH Pocket Guide to Chemical Hazards*; Department of Health and Human Services (DHHS), Centers for Disease Control and Prevention (CDC), National Institute for Safety and Health; *Nitric Acid*, 30 October 2019; available at https://www.cdc.gov/niosh/npg/npgd0447.html (accessed May 2021).

PubChem, *Nitric Acid CAS No.* 7697-37-2; available at https://pubchem.ncbi.nlm.nih.gov/compound/944 (accessed May 2021).

PubChem, *Calcium Nitrate CAS No. 10124-37-5*; available at https://pubchem.ncbi.nlm.nih.gov/compound/24963 (accessed May 2021).

# Key of Acronyms:

ACGIH	American Conference of Governmental Industrial	NRC	Nuclear Regulatory Commission
	Hygienists		
ALI	Annual Limit on Intake	NTP	National Toxicology Program
CAS	Chemical Abstracts Service	OSHA	Occupational Safety and Health Administration
CERCLA	Comprehensive Environmental Response,	PEL	Permissible Exposure Limit
	Compensation, and Liability Act		-
CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
DOT	Department of Transportation	REL	Recommended Exposure Limit
EC50	Effective Concentration, 50 %	RM	Reference Material
EINECS	European Inventory of Existing Commercial Chemical	RO	Reportable Quantity
	Substances		
EPCRA	Emergency Planning and Community Right-to-Know	RTECS	Registry of Toxic Effects of Chemical Substances
	Act		
IARC	International Agency for Research on Cancer	SARA	Superfund Amendments and Reauthorization Act
IATA	International Air Transport Association	SCBA	Self-Contained Breathing Apparatus
IDLH	Immediately Dangerous to Life and Health	SRM	Standard Reference Material
LC50	Lethal Concentration, 50 %	STEL	Short Term Exposure Limit
LD50	Lethal Dose, 50 %	STOT	Specific Target Organ Toxicity
LEL	Lower Explosive Limit	TLm	Threshold Limit, median
MSDS	Material Safety Data Sheet	TLV	Threshold Limit Value
NFPA	National Fire Protection Association	TPO	Threshold Planning Quantity
NIOSH	National Institute for Occupational Safety and Health	TSCA	Toxic Substances Control Act
NIST	National Institute of Standards and Technology	TWA	Time Weighted Average
n.o.s.	Not Otherwise Specified	UEL	Upper Explosive Limit
	1	WHMIS	Workplace Hazardous Materials Information System
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