

SAFETY DATA SHEET

1. SUBSTANCE AND SOURCE IDENTIFICATION

Product Identifier

SRM Number: 3103a
SRM Name: Arsenic (As) Standard Solution
Other Means of Identification: Not applicable.

Recommended Use of This Material and Restrictions of Use

This Standard Reference Material (SRM) is intended for use as a primary calibration standard for the quantitative determination of arsenic. A unit of SRM 3103a consists of five 10 mL sealed borosilicate glass ampoules of an acidified aqueous solution prepared gravimetrically to contain a known mass fraction of arsenic. The solution contains nitric acid at a mass fraction of approximately 23 %.

Company Information

National Institute of Standards and Technology
Standard Reference Materials Program
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2. HAZARDS IDENTIFICATION

Classification

Physical Hazard:	Not classified.	
Health Hazard:	Skin Corrosion/Irritation	Category 1
	Serious Eye Damage/Irritation	Category 1
	Carcinogenicity	Category 1A

Label Elements

Symbol



Signal Word

DANGER

Hazard Statement(s)

H314 Causes severe skin burns and eye damage.
H350 May cause cancer through inhalation.

Precautionary Statement(s)

P201 Obtain special instructions before use.
P202 Do not handle until all safety precautions have been read and understood.
P260 Do not breathe fumes, mists, vapors, spray.
P264 Wash hands thoroughly after handling.
P270 Do not eat, drink or smoke when using this product.
P280 Wear protective gloves, protective clothing, eye protection.
P301 + P330 + P331 If swallowed: Rinse mouth. Do NOT induce vomiting.
P303 + P361 + P353 If on skin (or hair): Remove immediately all contaminated clothing. Rinse skin with water.
P304 + P340 If inhaled: Remove person to fresh air and keep comfortable for breathing.
P305 + P351 + P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P310	Immediately call a doctor.
P363	Wash contaminated clothing before reuse.
P405	Store locked up.
P501	Dispose of contents and container according to local regulations.

Hazards Not Otherwise Classified: Not applicable.

Ingredients(s) with Unknown Acute Toxicity: Not applicable.

3. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Substance: Nitric Acid/Arsenic Acid Solution

Other Designations:

Nitric acid (aqua fortis; hydrogen nitrate; azotic acid; engraver's acid)

Arsenic acid (Arsenate; Orthoarsenic acid; H_3AsO_4)

Components are listed in compliance with OSHA's 29 CFR 1910.1200; for the actual values see the Certificate of Analysis.

Hazardous Component(s)	CAS Number	EC Number (EINECS)	Nominal Mass Concentration (%)
Nitric acid	7697-37-2	231-714-2	≤ 23
Arsenic Acid	7778-39-4	231-901-9	1.9
Non-Hazardous Component(s)			
Water	7732-18-5	231-791-2	≥ 75

4. FIRST AID MEASURES

Description of First Aid Measures:

Inhalation: If adverse effects occur, remove to uncontaminated area. If not breathing, give artificial respiration or oxygen by qualified personnel. Seek immediate medical attention.

Skin Contact: Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get immediate medical attention. Thoroughly clean and dry contaminated clothing before reuse. Destroy contaminated shoes.

Eye Contact: Immediately flush eyes, including under the eyelids with copious amounts of water for at least 15 minutes. Seek immediate medical attention.

Ingestion: Contact a poison control center immediately for instructions. Do not induce vomiting. Give water to rinse out mouth. Never give liquids to a person with reduced awareness or becoming unconscious. If vomiting occurs, keep head lower than hips to prevent aspiration. If not breathing, give artificial respiration by qualified personnel. Seek immediate medical attention.

Most Important Symptoms/Effects, Acute and Delayed: Acid burns to skin, eyes, and lungs.

Indication of any immediate medical attention and special treatment needed, if necessary: If any of the above symptoms are present, seek immediate medical attention.

5. FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Negligible fire hazard. See Section 9, "Physical and Chemical Properties" for flammability properties.

Extinguishing Media:

Suitable: Use extinguishing media appropriate to the surrounding fire.

Unsuitable: None listed.

Specific Hazards Arising from the Chemical: Thermal decomposition will form oxides of nitrogen and arsine.

Special Protective Equipment and Precautions for Fire-Fighters: Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH approved self-contained breathing apparatus (SCBA).

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions, Protective Equipment and Emergency Procedures: Immediately contact emergency personnel. Keep unnecessary personnel away. Use suitable protective equipment; see Section 8, “Exposure Controls and Personal Protection”.

Methods and Materials for Containment and Clean up: Do not touch spilled material. Notify safety personnel of spills. Absorb with sand or other non-combustible material. Collect spilled material in appropriate container for disposal. Isolate hazard area and deny entry.

7. HANDLING AND STORAGE

Safe Handling Precautions: See Section 8, “Exposure Controls and Personal Protection”. Handle glass ampoules with care.

Storage: Store and handle in accordance with all current regulations and standards. Keep separated from incompatible substances (See Section 10, “Stability and Reactivity”).

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Exposure Limits:

Component: Nitric acid

NIOSH (REL):	5 mg/m ³ (2 ppm; TWA)
	10 mg/m ³ (4 ppm; STEL)
	65 mg/m ³ (25 ppm; IDLH)
ACGIH (TLV):	5 mg/m ³ (2 ppm; TWA)
	10 mg/m ³ (4 ppm; STEL)
OSHA (PEL):	5 mg/m ³ (2 ppm; TWA)

Component: Arsenic acid (as As, related to Arsenic, inorganic compounds)

NIOSH (REL):	0.002 mg/m ³ (15 min, Ceiling)
	5 mg/m ³ (15 min, Ceiling)
ACGIH (TLV):	0.01 mg/m ³ (TWA)
OSHA (PEL):	10 µg/m ³ (cancer hazard, see 29 CFR 1910.1018, except Arsine as As, TWA)
	5 µg/m ³ (Action Level)

Engineering Controls: Provide local exhaust or process enclosure ventilation system. Ensure compliance with applicable exposure limits.

Personal Protection: In accordance with OSHA 29 CFR 1910.132, subpart I, wear appropriate Personal Protective Equipment (PPE) to minimize exposure to this material.

Respiratory Protection: If workplace conditions warrant a respirator, a respiratory protection program that meets OSHA 29CFR 1910.134 must be followed. Refer to NIOSH 42 CFR 84 for applicable certified respirators.

Eye/Face Protection: Wear splash resistant safety goggles with a face shield. An eyewash station should be readily available near areas of use.

Skin and Body Protection: Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product. Chemical-resistant gloves should be worn at all times when handling chemicals.

9. PHYSICAL AND CHEMICAL PROPERTIES

Descriptive Properties:

Appearance (physical state, color, etc.):	liquid
Molecular Formula:	not applicable
Molar Mass (g/mol):	not applicable
Odor:	irritating odor
Odor threshold:	not available
pH:	acidic
Evaporation rate:	not available
Melting point/freezing point (°C):	not available
Relative Density (g/L) as specific gravity (water = 1):	not available
Vapor Pressure (mmHg):	not available

Descriptive Properties:

Vapor Density (air = 1):	not available
Viscosity (cP):	not available
Solubility(ies):	miscible with water and ether
Partition coefficient (n-octanol/water):	not available

Thermal Stability Properties:

Autoignition Temperature (°C):	not applicable
Thermal Decomposition (°C):	not applicable
Initial boiling point and boiling range (°C):	not available
Explosive Limits, LEL (Volume %):	not applicable
Explosive Limits, UEL (Volume %):	not applicable
Flash Point (°C):	not applicable
Flammability (solid, gas):	not applicable

10. STABILITY AND REACTIVITY

Reactivity: Stable at normal temperatures and pressure.

Stability: X Stable Unstable

Possible Hazardous Reactions: None listed.

Conditions to Avoid: Contact with combustible or incompatible materials.

Incompatible Materials: Incompatible with acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals, halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, and cyanides.

Fire/Explosion Information: See Section 5, "Fire Fighting Measures".

Hazardous Decomposition: Thermal decomposition will produce oxides of nitrogen and arsenic.

Hazardous Polymerization: Will Occur X Will Not Occur

11. TOXICOLOGICAL INFORMATION

Route of Exposure: X Inhalation X Skin X Ingestion

Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Burning pain and severe skin corrosion, eye, lung, and blood damage, and cancer.

Potential Health Effects (Acute, Chronic and Delayed):

Inhalation: Inhalation of nitric acid can damage the mucous membranes and upper respiratory tract. Short term exposure may cause irritation and inflammation of the upper respiratory tract, coughing, choking, sore throat, shortness of breath, headache, dizziness, and nausea. Long term exposure to acid fumes may cause damage to teeth, bronchial irritation, chronic cough, bronchial pneumonia, and gastrointestinal disturbances. Arsenic inorganic compounds may cause foamy sputum.

Skin Contact: Nitric acid can cause severe skin burns. Severity of the damage depends on the concentration and duration of exposure. Effects of acid burns may be delayed. Short term contact with arsenic inorganic compounds can cause irritation and may cause sensitization.

Eye Contact: Nitric acid and arsenic inorganic compounds can cause severe eye irritation, corneal burns, permanent eye damage, or blindness. Severity of the damage depends on the concentration and duration of exposure.

Ingestion: Ingestion of this material is unlikely under normal conditions of use. If ingested, nitric acid can cause severe burns and damage to the gastrointestinal tract. Acute ingestion of low levels of arsenic inorganic compounds can cause tearing, diarrhea, bluish skin color, kidney damage, liver damage, and death. Chronic ingestion may have the same effects and may also cause cancer.

Numerical Measures of Toxicity:

Acute Toxicity: Not classified.

Nitric acid, Rat, Inhalation LC50: 130 mg/m³ (4 h)

Arsenic acid, Rat, Oral LD50: 48 mg/kg

Skin Corrosion/Irritation: This SRM contains >1 % of nitric acid and it is classified as Category 1.

Serious Eye Damage/Irritation: This SRM contains >1 % nitric acid and it is classified as Category 1.

Respiratory Sensitization: No data available; not classified.

Skin Sensitization: No data available; not classified.

Germ Cell Mutagenicity: No data available; not classified.

Carcinogenicity: Category 1A.

Listed as a Carcinogen/Potential Carcinogen X Yes No

Nitric acid is not listed by NTP, IARC or OSHA as a carcinogen.

NTP lists Arsenic (inorganic compounds) as known human carcinogen. IARC Monograph 84 (2004) lists arsenic in Group 1 (carcinogenic to humans). OSHA lists inorganic arsenic as a designated carcinogen.

Reproductive Toxicity: Not classified.

Nitric acid, Rat, Oral TDLo: 21 150 mg/kg (pregnant 1 d to 21 d)

Nitric acid, Rat, Oral TDLo: 2345 mg/kg (pregnant 18 d)

Arsenic acid: Rat, Oral TDLo: 120 mg/kg (pregnant 7 to 15 d)

Specific Target Organ Toxicity, Single Exposure: No data available; not classified.

Specific Target Organ Toxicity, Repeated Exposure: No data available; not classified.

Aspiration Hazard: No data available; not classified.

12. ECOLOGICAL INFORMATION

Ecotoxicity Data:

Nitric acid: Starfish (*Asterias rubens*) LC50: 100 mg/L – 300 mg/L (renewal/aerated water, 48 h)

Arsenic acid: Fathead minnow (*Pimephales promelas*) LC50: 25.6 mg/L (96 h)

Bluegill (*Lepomis macrochirus*) LC50: 39 mg/L – 110 mg/L (static, 96 h)

Bluegill (*Lepomis macrochirus*) LC50: 43 mg/L – 59 mg/L (flow-through, 96 h)

Persistence and Degradability: No data available.

Bioaccumulative Potential: No data available.

Mobility in Soil: No data available.

Other Adverse effects: No data available.

13. DISPOSAL CONSIDERATIONS

Waste Disposal: Dispose of waste in accordance with all applicable federal, state, and local regulations.

Nitric acid and arsenic acid subject to disposal regulations: U.S. EPA 40 CFR 262.

Nitric acid Hazardous Waste Numbers: D001, D002.

Arsenic acid Hazardous Waste Numbers: P010, D004. Dispose of in accordance with U.S. EPA 40 CFR 262 for concentrations at or above the Regulatory level (5.0 mg/L).

14. TRANSPORTATION INFORMATION

U.S. DOT and IATA: UN1760, Corrosive liquid, n.o.s. (contains nitric acid), Hazard Class 8, Packing Group II, Excepted Quantities E2.

15. REGULATORY INFORMATION

U.S. Regulations:

CERCLA Sections 102a/103 (40 CFR 302.4): Nitric acid, 1000 lbs (454 kg) RQ;

Arsenic acid, 1 lb (0.454 kg) RQ

SARA Title III Section 302 (40 CFR 355.30): Nitric acid, 1000 lbs (454 kg) TPQ

SARA Title III Section 304 (40 CFR 355.40): Nitric acid, 1000 lbs (454 kg) EPCRA RQ

SARA Title III Section 313 (40 CFR 372.65): Nitric acid, 1 % de minimis concentration;

Arsenic acid, 0.1 % de minimis concentration (related to Arsenic inorganic compounds)

OSHA Process Safety (29 CFR 1910.119): Nitric acid, higher concentrations 500 lbs TQ (≥94.5 % by weight)

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE HEALTH: Yes.

CHRONIC HEALTH: Yes.

FIRE: No.

REACTIVE: No.

PRESSURE: No.

State Regulations:

California Proposition 65: WARNING! This product contains a chemical known (arsenic inorganic compounds) to the state of California to cause cancer.

U.S. TSCA Inventory: Nitric acid and arsenic acid are listed.

TSCA 12(b), Export Notification: Not listed.

Canadian Regulations: WHMIS Information is not provided for this material.

16. OTHER INFORMATION

Issue Date: 16 October 2023

Sources: ChemAdvisor, Inc., SDS *Nitric Acid*, 09 December 2015.

ChemAdvisor, Inc., SDS *Arsenic Acid*, 09 December 2015.

CDC; NIOSH; *NIOSH Pocket Guide to Chemical Hazards*; Department of Health and Human Services (DHHS), Centers for Disease Control and Prevention (CDC), National Institute for Safety and Health; *Nitric Acid*, 30 October 2019; available at <https://www.cdc.gov/niosh/npg/npgd0447.html> (accessed Oct 2023).

National Library of Medicine's PubChem Database, *Nitric Acid CAS No. 7697-37-2*; available at <https://pubchem.ncbi.nlm.nih.gov/> (accessed Oct 2023).

Key of Acronyms:

ACGIH	American Conference of Governmental Industrial Hygienists	NRC	Nuclear Regulatory Commission
ALI	Annual Limit on Intake	NTP	National Toxicology Program
CAS	Chemical Abstracts Service	OSHA	Occupational Safety and Health Administration
CERCLA	Comprehensive Environmental Response, Compensation, and Liability Act	PEL	Permissible Exposure Limit
CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
DOT	Department of Transportation	REL	Recommended Exposure Limit
EC50	Effective Concentration, 50%	RM	Reference Material
EINECS	European Inventory of Existing Commercial Chemical Substances	RQ	Reportable Quantity
EPCRA	Emergency Planning and Community Right-to-Know Act	RTECS	Registry of Toxic Effects of Chemical Substances
IARC	International Agency for Research on Cancer	SARA	Superfund Amendments and Reauthorization Act
IATA	International Air Transport Association	SCBA	Self-Contained Breathing Apparatus
IDLH	Immediately Dangerous to Life and Health	SRM	Standard Reference Material
LC50	Lethal Concentration, 50 %	STEL	Short Term Exposure Limit
LD50	Lethal Dose, 50 %	TLV	Threshold Limit Value
LEL	Lower Explosive Limit	TPQ	Threshold Planning Quantity
MSDS	Material Safety Data Sheet	TSCA	Toxic Substances Control Act
NFPA	National Fire Protection Association	TWA	Time Weighted Average
NIOSH	National Institute for Occupational Safety and Health	UEL	Upper Explosive Limit
NIST	National Institute of Standards and Technology	WHMIS	Workplace Hazardous Materials Information System
n.o.s.	Not Otherwise Specified		

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