

DEPARTMENT OF COMMERCE

Bureau of Standards

SS  
RECORD FILE

Certificate of Analyses

OF

STANDARD SAMPLE No. 50

CHROME-TUNGSTEN-VANADIUM STEEL

ANALYST.	CARBON.		SILICON.		PHOSPHORUS.		SULPHUR.		MANGANESE.		CHROMIUM.		TUNGSTEN.		VANADIUM.	COPPER.	Molybdenum.
	Recommended Method.	Analyst's Method.	Recommended Method.	Analyst's Method.	Recommended Method.	Analyst's Method.	Recommended Method.	Analyst's Method.	Recommended Method.	Analyst's Method.	Recommended Method.	Analyst's Method.	Recommended Method.	Analyst's Method.	Analyst's Method.	H <sub>2</sub> S—CuS—CuO.	
1	.055		.152	.164 <sup>a</sup>	.026		.031		.201		3.60		17.55	17.51 <sup>b</sup>	.757 <sup>c</sup>	.044	.01
2	.667		.175		.025		.031		.199	.200	3.52		17.50		.803 <sup>d</sup>		
3		.660 <sup>e</sup>		.164	.031	.031	.030	.030	.201	.196 <sup>f</sup>		3.62 <sup>f</sup>	17.53	17.53	.759 <sup>f</sup>	{.045 .049 <sup>z</sup> }	
4		.650 <sup>g</sup>		.164		.025 <sup>h</sup>		{.030 <sup>i</sup> .031 <sup>j</sup> }		.200 <sup>k</sup>		3.63 <sup>l</sup>		17.61	.762 <sup>m</sup>		
5	.639	.635	.140	.135			.033	.023 <sup>n</sup>		.228 <sup>o</sup>	3.63	3.64	17.53	17.58 <sup>p</sup>		trace	
6		.630 <sup>e</sup>	.130	.152 <sup>u</sup>	.026	.026 <sup>q</sup>		{.026 <sup>r</sup> .023 <sup>n</sup> }		.21 <sup>s</sup>		3.54 <sup>l</sup>		17.59		trace	
7	.682		.159		.026		.032		.190			3.57 <sup>f</sup>	17.38	17.38	.736 <sup>f</sup>		
8	.009		.13	.106	.027	.024 <sup>q</sup>	.031	.028 <sup>r</sup>	.215	.21 <sup>t</sup>	3.68	3.64 <sup>u</sup>	17.65	17.65 <sup>v</sup>	.755 <sup>w</sup>	trace	
9	.67		.15		.025						3.52		17.48				
10	.641			.135								3.64 <sup>x</sup>	17.49		.705 <sup>x</sup>		
11	.665	.668 <sup>y</sup>	.163	.162			.034 <sup>r</sup>		.194 <sup>k</sup>		3.64 <sup>l</sup>		17.70	17.79	.774 <sup>m</sup>	.063	
Aver.	.661	.649	.150	.155	.027	.027	.031	.030	.201	.205	3.59	3.62	17.53	17.58	.756	.050	.01
Gen. Av.	.656		.153 <sup>*</sup>		.027		.031		.204		3.61		17.56		.756	.050	.01

\* Recommended value, 0.16.

NOTE.—When analyst's method differs only in detail from the recommended method, no footnote is given.

- <sup>a</sup> H<sub>2</sub>SO<sub>4</sub> dehydration.
- <sup>b</sup> Solution in HCl, precipitation with HNO<sub>3</sub>+cinchonine, solution in NH<sub>4</sub>OH and reprecipitation with acid and cinchonine.
- <sup>c</sup> Reduction with SO<sub>2</sub> after removal of WO<sub>3</sub> with HNO<sub>3</sub> digestion, precipitation with NaHCO<sub>3</sub>, then B. S. fusion method.
- <sup>d</sup> Same as <sup>c</sup>, but reduced with zinc.
- <sup>e</sup> Absorbed in soda-asbestos.
- <sup>f</sup> Titrated electrometrically; J. Ind. Eng. Chem., 10, 19 (1918).
- <sup>g</sup> Absorbed in KOH.
- <sup>h</sup> Precipitated with a water solution of (NH<sub>4</sub>)<sub>2</sub>MoO<sub>4</sub> in an oxidized solution; J. Ind. Eng. Chem., 11, 113 (1919).
- <sup>i</sup> Last traces of WO<sub>3</sub> removed with cinchonine before precipitation with BaCl<sub>2</sub>.
- <sup>j</sup> Volatilization of sulphur as H<sub>2</sub>S by passing H<sub>2</sub>+HCl over drillings at 950° C.—1000° C.
- <sup>k</sup> PbO<sub>2</sub>—arsenite after removal of chromium with ZnO.
- <sup>l</sup> KMnO<sub>4</sub> oxidation.

- <sup>m</sup> KMnO<sub>4</sub> oxidation and FeSO<sub>4</sub> titration using K<sub>2</sub>Fe(CN)<sub>6</sub> as internal indicator.
- <sup>n</sup> Volumetric and not included in average.
- <sup>o</sup> Recommended method, finishing with arsenite.
- <sup>p</sup> Used quinine hydrochloride instead of cinchonine in recommended method.
- <sup>q</sup> Same as recommended but dissolved in aqua regia.
- <sup>r</sup> Same as recommended, without zinc reduction.
- <sup>s</sup> Bismuthate—arsenite.
- <sup>t</sup> Ford-Williams' method.
- <sup>u</sup> KClO<sub>3</sub> oxidation and titration with KMnO<sub>4</sub> and FeSO<sub>4</sub>.
- <sup>v</sup> Same as recommended, but without cinchonine.
- <sup>w</sup> KClO<sub>3</sub> oxidation and titration with FeSO<sub>4</sub> using K<sub>2</sub>Fe(CN)<sub>6</sub> as internal indicator.
- <sup>x</sup> Cain & Hostetter method.
- <sup>y</sup> Chips mixed with red lead before combustion.
- <sup>z</sup> Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>—CuS—CuO.

INDEX TO ANALYSTS

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This standard is not recommended for colorimetric carbon determinations, because of uncertainty as to the condition of the carbon.

Washington, D. C., February 1, 1921.

S. W. STRATTON,  
Director.