## United States Department of Commerce Washington

## National Bureau of Standards

## Certificate

FOR

STANDARD SAMPLE NO. 41

DEXTROSE

Ash......0.003% Moisture.......01%

Each 100 ml of a normal dextrose solution contains 32.231 g of the dried substance weighed in air with brass weights. At 20°C this solution in a 200-mm polariscope tube reads 100°S on the International Sugar Scale as defined at the Eighth Session of the International Commission for Uniform Methods of Sugar Analysis, Amsterdam, 1932 [Int. Sugar Jour. 35, 17, 62 (1933), N.B.S. Circular C440, 79, 775 (1942)]. The illumination is white light filtered through a 15-mm layer of a 6-percent solution of potassium dichromate.

The rotation in circular degrees of the normal dextrose solution observed in a 200-mm polariscope tube, for wave length 5461 A is 40.897°.

The specific rotation of dextrose for wave length 5461 A is represented by the formula:

 $[a]_{5461A}^{20^{\circ}C} = 62.032^{\circ} + 0.04257^{\circ}c$ 

where c is grams of anhydrous dextrose weighed in vacuo and contained in 100 ml of solution. This formula is valid for values of c between 6 g and 32 g. [Bull. B. S. 13, 633 (1916), S293; N.B.S. Circular C440, 83 (1942)].

For the Director

Edward Wichers, Chief, Chemistry Division

Lot No. 4854 Washington, D. C.