

U. S. DEPARTMENT OF COMMERCE
WASHINGTON

National Bureau of Standards
Certificate of Analyses
Standard Sample 171
Magnesium-base Alloy

ANALYST	ALUMINUM	ZINC	MANGANESE	SILICON	COPPER	LEAD	IRON	NICKEL
1.	*2.97	^b 1.05	{ ^a 0.45} ^a 0.46	*0.011 _s	{ ^a 0.011 _s } ^a 0.012 _s	^b 0.0038	^a 0.0018	^a 0.0007
2.	{ ^a 2.96} ^a 2.98	^b 1.05	.45	*0.011 _s	{ ^a 0.010 _s } ^a 0.011 _s	^b 0.0031	^a 0.0017	^a 0.0008
3.	*2.97	^b 1.04	.44	*0.012	*0.010	^b 0.003	^a 0.002	^a 0.0014
4.	*3.00	^b 1.07	.45	*0.010 _s	*0.012	^b 0.0028		
5.	*2.98	^b 1.04	.45	*0.012	*0.011 _s	^b 0.0033	*0.0015	^a 0.0008
6.	{ ^a 2.98} ^a 2.99	^b 1.06	{ ^a .45} ^a .46	*0.011	{ ^a .011} ^a .011	{ ^a .0031} ^a .0034	*0.0018	^a 0.001
7.	*3.00	^b 1.03	.45	*0.013	*0.011	^b 0.004	{ ^a .0017} ^a .002	^a 0.001
8.	*2.96	^b 1.04	.46	*0.013	*0.011	^b 0.0031	^a 0.0018	^a 0.0008
Average	2.98	1.05	0.45	0.011 _s	0.011 _s	0.0033	0.0018	0.0009

* Benzoate-8-hydroxyquinoline method. See ASTM method E35-49. Methods for Chemical Analysis of Metals, p. 341 (1950). American Society for Testing Materials, Philadelphia, Pa.

^a ZnS-ZnO method.

^b KIO₄-photometric method. See ASTM method E35-50T.

^c Persulfate-arsenite method using potentiometric end point.

^d Molybdate-blue acid-photometric method. See ASTM method E35-50T.

^e Electrolytic deposition.

^f HBr-photometric method. See ASTM method E35-50T.

^g Dithizone method. See ASTM method E35-50T.

^a Bipyridyl-photometric method. See ASTM method E35-50T.

^b Dimethylglyoxime-photometric method. See ASTM method E35-50T.

^c Alizarin red S-photometric method.

^d Potassium ferrocyanide method. See ASTM method E35-49.

^e Copper separated with H₂S and determined by the iodide-thiosulfate method.

^f Molybdenum blue-photometric method. See Anal. Chem. 20, 630 (1948).

^g Diethylidithiocarbamate-photometric method.

^h Thiocyanate-photometric method.

ⁱ Double ammonium hydroxide precipitation with intervening oxine separation.

^a Internal electrolytic method.

^b Orthophenanthroline-photometric method.

^c Persulfate-arsenite method.

^d Lead deposited electrolytically as PbO₂.

^e Mercury cathode-aluminum oxide method.

^f ZnHg (ONS) method.

^g Perchloric acid method.

^h Lead separated electrolytically and determined by iodometric titration.

ⁱ Aluminon-photometric method.

^j Potassium ferrocyanide method using potentiometric end point.

Analyst 7 reported 0.002 percent calcium by the flame photometer method.

List of Analysts

1. R. K. Bell, K. M. Sappenfield, and E. E. Maczkowske, National Bureau of Standards, Washington 25, D. C.
2. V. A. Stenger and Walter R. Kramer, The Dow Chemical Co., Midland, Mich.
3. Sydney Abbey, Dominion Magnesium Limited, Haley, Ontario, Canada.
4. U. S. Naval Engineering Experiment Station, Annapolis, Md.
5. R. G. Ernst, United States Metals Refining Co., Carteret, N. J.
6. J. J. Aldrich, Jacob Nitz, and Robert Vitek, Apex Smelting Co., Cleveland, Ohio.
7. H. V. Churchill and M. L. Moss, Aluminum Co. of America, New Kensington, Pa.
8. K. C. Braun, American Smelting and Refining Co., Barber, N. J.

The metal for the preparation of this standard was furnished by the Dow Chemical Co.

WASHINGTON 25, D. C., November 27, 1951.

A. V. ASTIN, Acting Director.