

National Institute of Standards & Technology

Certificate of Analysis

Standard Reference Material® 158a

Silicon Bronze (chip form)

This Standard Reference Material (SRM) is intended primarily for use in validation of chemical and instrumental methods of analysis for element contents of bronze alloys and materials of similar matrix. It can be used to validate value assignment of a laboratory's in-house reference materials. A unit of SRM 158a consists of one bottle containing approximately 150 g of chips produced by a milling machine.

Certified Mass Fraction Values: Certified values for constituents of SRM 158a are reported in Table 1 as mass fractions of the elements in a bronze alloy matrix [1]. A NIST certified value is a value for which NIST has the highest confidence in its accuracy in that all known or suspected sources of bias have been taken into account [2]. A certified value is the present best estimate of the true value. The certified values are metrologically traceable to the SI derived unit of mass fraction (expressed as percent). The expanded uncertainty estimates are expressed at a confidence level of approximately 95 %, calculated following the ISO/JCGM Guide [3–10].

Table 1. Certified Mass Fraction Values for SRM 158a Silicon Bronze (chip form)

Constituent	Mass Fraction (%)	Expanded Uncertainty (%)
Aluminum (Al)	0.4580	0.0094
Copper (Cu)	90.93	0.29
Iron (Fe)	1.228	0.015
Lead (Pb)	0.0973	0.0048
Manganese (Mn)	1.112	0.014
Phosphorus (P)	0.0263	0.0028
Silicon (Si)	3.026	0.021
Tin (Sn)	0.960	0.013
Zinc (Zn)	2.076	0.019

Expiration of Certification: The certification of **SRM 158a** is valid indefinitely within the measurement uncertainty specified, provided the SRM is handled and stored in accordance with instructions given in this certificate (see "Instructions for Handling, Storage, and Use"). Periodic recertification of this SRM is not required. The certification is nullified if the SRM is damaged, contaminated, or otherwise modified.

Maintenance of SRM Certification: NIST will monitor this SRM over the period of its certification. If substantive technical changes occur that affect the certification before the expiration of this certificate, NIST will notify the purchaser. Registration (see attached sheet or register online) will facilitate notification.

Coordination of technical measurements for certification was performed by R.K. Bell, formerly of NIST. Review and revision of values and uncertainty estimates was coordinated by J.R. Sieber of the NIST Chemical Sciences Division.

Statistical consultation for this SRM was provided by A. Possolo of the NIST Statistical Engineering Division.

Carlos A. Gonzalez, Chief Chemical Sciences Division

Steven J. Choquette, Director Office of Reference Materials

Gaithersburg, MD 20899 Certificate Issue Date: 23 August 2018 Certificate Revision History on Last Page

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Support aspects involved in the issuance of this SRM were coordinated through the NIST Office of Reference Materials.

INSTRUCTIONS FOR HANDLING, STORAGE, AND USE

Silicon bronze chips may be analyzed in the as-received form. Test methods used to characterize the material used sample quantities of 2 g or more. While the material is believed to be homogeneous at lower quantities, it has not been tested in amounts < 2 g. Before sampling, it is recommended to mix bottle contents by inverting and rotating the bottle by hand for at least one minute. A bottle containing unused material should be recapped immediately and stored at room temperature away from light.

To use the uncertainty estimates given in this certificate, divide the expanded uncertainty by a coverage factor of k = 2 to obtain the combined standard uncertainty. The effective degrees of freedom of the combined standard uncertainty are ≥ 60 .

PREPARATION AND ANALYSIS(1)

The material was sieved, blended and bottled at NIST. Homogeneity testing of samples from the final lot was performed J.E. Cumbo, Airco Alloys (Niagara Falls, NY) and J.C. Cline, Interlake, Inc. (Beverly, OH). Analyses for certification were performed using the test methods listed in Table 2.

Analyses leading to the certification of this SRM were performed at NIST by R.K. Bell and E.E. Maczkowske, formerly of NIST. Analytical determinations were also performed by A.W. Young, Bridgeport Brass (Bridgeport, CT); E.L. Smith, H.J. Smith, R.C. Burnham, and A.B. Feest, Chase Brass and Copper Co. (Waterbury, CT); O.P. Case, G.A. Reihl, W.T. Taras, and K.M. O'Brien, The American Brass Co. (Waterbury, CT); and K.H. Storks, E. Bloom, Jr., and E.K. Jaycox, Bell Telephone Laboratories (Murray Hill, NJ).

Table 2. Test Methods Employed in the Certification of SRM 158a Silicon Bronze (chip form)

Element	Test Methods Used at NIST and Collaborating Laboratories	
Aluminum	 Cu in a 2 g sample removed by electrolysis of a HNO₃-HF solution. Mercury cathode separation then made in sulfate solution followed by H₂S separation in 0.01 N acidity. Sulfides filtered off, and MnO₂ removed with persulfate in dilute acid solution. All precipitated twice with NH₄OH and ignited to Al₂O₃.; Mercury cathode-Al₂O₃ method (see ASTM International method E54-49); Mercury cathode 8-hydroxyquinoline gravimetric method; Cu removed by electrolysis of a HNO₃-HF solution. Si in the electrolyte removed with H₂SO₄-HF treatment, Sn with HBr, and Mn with (NH₄)₂S₂O₈. Filtrate electrolyzed in a mercury cathode. All precipitated with NH₄OH, ignited to Al₂O₃, and corrected for P. 	
Copper	 5.0 g sample dissolved in 40 mL HNO₃ (1+1). Solution digested on steam bath overnight, filtered and the precipitate washed with hot HNO₃ (1+99). Precipitate treated with HNO₃-HClO₄-HF-HBr and residual solution combined with first filtrate. Two drops 0.1 N HCl added, solution diluted to 300 mL and electrolyzed overnight, using a current density of 0.5 A/dm². Residual Cu and Pb in electrolyte precipitated with H₂S and determined by electrolysis.; ASTM International method E36-45; Direct electrolysis of a 2 g sample in HNO₃-HF solution containing small amount of added Pb; Direct electrolysis of a 2 g sample in HNO₃-HF solution. 	
Iron	 SnCl₂-K₂Cr₂O₇ method; Fe titrated with TiCl₃; Fe reduced in a silver redactor and titrated with Ce(SO₄)₂. 	
Lead	 Weighed as PbO₂; Dithizone-photometric method; Pb separated as PbSO₄ and weighed as PbCrO₄. 	

⁽¹⁾ Certain commercial equipment, instruments or materials are identified in this certificate to adequately specify the experimental procedure. Such identification does not imply recommendation or endorsement by the National Institute of Standards and Technology, nor does it imply that the materials or equipment identified are necessarily the best available for the purpose.

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Element	Test Methods Used at NIST and Collaborating Laboratories
Manganese	 Persulfate-arsenite method with potentiometric titration; KIO₄ photometric method (see ASTM International method E62-56); NaBiO₃ method.
Phosphorus	Phosphomolybdenum blue method;Molybdivanadophosphoric acid method.
Silicon	 Double dehydration with HClO₄, and a Na₂CO₃ fusion followed by double dehydration with H₂SO₄; Photometric method; H₂SO₄ dehydration (see ASTM International method E54-49); Double dehydration with H₂SO₄, and a Na₂CO₃ fusion followed by double dehydration with H₂SO₄; Molybdisilicic acid-photometric method; HClO₄ dehydration.
Tin	 Sn reduced with Pb and titrated with KIO₃ standardized with high-purity Sn; Sn reduced with sodium hypophosphite and titrated with iodine; Sn reduced with Al and titrated with KIO₃ standardized with high-purity Sn; Sn reduced with Pb and titrated with KIO3 (see ASTM International method 54-49).
Zinc	 Cu removed by electrolysis and electrolyte evaporated to fumes of H₂SO₄. Solution treated with an excess of NaOH and filtered. Zn determined in filtrate by electrolysis.; ZnHg(CNS)₄ method; Zn extracted as thiocyanate and titrated with sodium ethylenediaminetetraacetate; Zn in the electrolyte from the Cu determination precipitated twice with H₂S in formic acid solution and ignited to ZnO.

ADDITIONAL CONSTITUENTS: Noncertified values are provided for the following additional constituents in SRM 158a.

Information Mass Fraction Values: Information values for constituents in SRM 158a are reported as mass fractions in Table 3. An information value is a value that may be of interest to the SRM user, but insufficient information is available to assess the uncertainty associated with the value [2]. Information values cannot be used to establish metrological traceability.

Table 3. Information Mass Fraction Values for SRM 158a Silicon Bronze (chip form)

Constituent	Mass Fraction	
	(%)	
Chromium (Cr)	0.001	
Nickel (Ni)	0.001	
Silver (Ag)	0.001	

NOTICE TO USERS

NIST strives to maintain the SRM inventory supply, but NIST cannot guarantee the continued or continuous supply of any specific SRM. Accordingly, NIST encourages the use of this SRM as a primary benchmark for the quality and accuracy of the user's in-house reference materials and working standards. As such, the SRM should be used to validate the more routinely used reference materials in a laboratory. Comparisons between the SRM and a laboratory's in-house reference materials or working measurement standards should take place at intervals appropriate to the conservation of the SRM and the stability of relevant in-house materials. For further guidance on how this approach can be implemented, contact NIST by email at srms@nist.gov.

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REFERENCES

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Certificate Revision History: 23 August 2018 (Revised values and uncertainties for certified constituents; update of the certificate to current NIST standards; editorial changes); 8 August 1961 (Original certificate date).

Users of this SRM should ensure that the Certificate of Analysis in their possession is current. This can be accomplished by contacting the SRM Program: telephone (301) 975-2200; fax (301) 948-3730; e-mail srminfo@nist.gov; or via the Internet at https://www.nist.gov/srm.

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