

SAFETY DATA SHEET

1. Substance and Source Identification

Product Identifier

SRM Number: 3155

SRM Name: Tantalum (Ta) Standard Solution **Other Means of Identification:** Not applicable.

Recommended Use of This Material and Restrictions of Use

This Standard Reference Material (SRM) is intended for use as a primary calibration standard for the quantitative determination of tantalum. A unit of SRM 3155 consists a high-density polyethylene bottle sealed in an aluminized bag, which contains 50 mL of an aqueous solution acidified with nitric and hydrofluoric acids, and a known mass fraction of tantalum.

Company Information

National Institute of Standards and Technology Standard Reference Materials Program 100 Bureau Drive, Stop 2300 Gaithersburg, Maryland 20899-2300

 Telephone:
 301-975-2200
 Emergency Telephone ChemTrec:

 FAX:
 301-948-3730
 1-800-424-9300 (North America)

 E-mail:
 SRMMSDS@nist.gov
 +1-703-527-3887 (International)

Website: https://www.nist.gov/srm

2. HAZARDS IDENTIFICATION

Classification

Physical Hazard: Not classified.

Health Hazard: Skin Corrosion/Irritation Category 1B Serious Eye Damage/Eye Irritation Category 1

Label Elements Symbol



Signal Word DANGER

Hazard Statement(s)

H314 Causes severe skin burns and eye damage.

Precautionary Statement(s)

P260 Do not breathe fumes, mists, vapors, or spray.
P264 Wash hands thoroughly after handling.

P280 Wear protective gloves, protective clothing, and eye protection.

P301+P330+P331 If swallowed: Rinse mouth. Do NOT induce vomiting.

P303+P361+P353 If on skin (or hair): Remove immediately all contaminated clothing. Rinse skin with

water.

P304+P340 If inhaled: Remove person to fresh air and keep comfortable for breathing.

P305+P351+P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if

present and easy to do. Continue rinsing.

P310 Immediately call a doctor.

P363 Wash contaminated clothing before reuse.

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P405 Store locked up.

P501 Dispose of contents and container according to local regulations.

Hazards Not Otherwise Classified: Not applicable.

Ingredients(s) with Unknown Acute Toxicity: Not applicable.

3. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Substance: Tantalum in nitric/hydrofluoric acid solution

Other Designations:

Nitric acid (aqua fortis; hydrogen nitrate; azotic acid; engraver's acid)

Hydrofluoric acid (hydrogen fluoride; fluorhydric acid)

Tantalum nitrate [tantalum(V) nitrate]

NOTE: Tantalum in nitric/hydrofluoric acid solution forms a solvated tantalum nitrate salt. The health and physical hazard information provided in this SDS are for nitric/hydrofluoric acid and tantalum nitrate, a rare earth salt. No physical or chemical data are listed for this solution. The actual effects of the solution may differ from the individual components.

Components are listed in compliance with OSHA's 29 CFR 1910.1200; for the actual values see the NIST Certificate of Analysis.

Hazardous Component(s)	CAS Number	EC Number (EINECS)	Nominal Mass Concentration (%)
Nitric acid	7697-37-2	231-714-2	<20
Hydrofluoric acid	7664-39-3	231-634-8	< 2
Tantalum nitrate	14985-19-4	239-069-9	< 3
Non-Hazardous Component(s) Water	7732-18-5	231-791-2	balance

4. FIRST AID MEASURES

Description of First Aid Measures:

Inhalation: If adverse effects occur, remove to uncontaminated area. If not breathing, give artificial respiration or oxygen by qualified personnel. Seek immediate medical attention.

Skin Contact: Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get immediate medical attention. Thoroughly clean and dry contaminated clothing before reuse. Destroy contaminated shoes.

Eye Contact: Immediately flush eyes, including under the eyelids with copious amounts of water for at least 15 minutes. Seek immediate medical attention.

Ingestion: Contact a poison control center immediately for instructions. Do not induce vomiting. Give water to rinse out mouth. Never give liquids to a person with reduced awareness or becoming unconscious. If vomiting occurs, keep head lower than hips to prevent aspiration. If not breathing, give artificial respiration by qualified personnel. Seek immediate medical attention.

Most Important Symptoms/Effects, Acute and Delayed: Acid burns to skin, eyes, and lungs.

Indication of any immediate medical attention and special treatment needed, if necessary: If any of the above symptoms are present, seek immediate medical attention.

5. FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Negligible fire hazard. See Section 9, "Physical and Chemical Properties" for flammability properties.

Extinguishing Media:

Suitable: Use extinguishing media appropriate to the surrounding fire.

Unsuitable: None listed.

Specific Hazards Arising from the Chemical: Miscellaneous decomposition products.

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Special Protective Equipment and Precautions for Fire-Fighters: Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH approved self-contained breathing apparatus (SCBA).

Health = 3

Fire = 0

Reactivity = 0

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions, Protective Equipment and Emergency Procedures: Immediately contact emergency personnel. Keep unnecessary personnel away. Use suitable protective equipment; see Section 8, "Exposure Controls and Personal Protection".

Methods and Materials for Containment and Clean up: Do not touch spilled material. Notify safety personnel of spills. Absorb with sand or other non-combustible material. Collect spilled material in appropriate container for disposal. Isolate hazard area and deny entry.

7. HANDLING AND STORAGE

Safe Handling Precautions: See Section 8, "Exposure Controls and Personal Protection".

Storage: Store and handle in accordance with all current regulations and standards. Keep separated from incompatible substances (see Section 10, "Stability and Reactivity").

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Exposure Limits				
Component(s)	OSHA (PEL)	ACGIH (TLV)	NIOSH (REL)	
Nitric acid	5 mg/m³ (2 ppm) TWA	5 mg/m³ (2 ppm) TWA 10 mg/m³ (4 ppm) STEL	5 mg/m³ (2 ppm) TWA 10 mg/m³ (4 ppm) STEL 65 mg/m³ (25 ppm) IDLH	
Hydrofluoric acid	2.5 mg/m ³ (3 ppm) TWA (as F)	0.5 ppm TWA (as F) 2 ppm Ceiling (as F) Skin - potential significant contribution to overall exposure by the cutaneous route	2.5 mg/m³ (3 ppm) TWA 5 mg/m³ (6 ppm) Ceiling 15 min (30 ppm) IDLH	
Tantalum nitrate	No occupational limits established.			

Engineering Controls: Provide local exhaust or process enclosure ventilation system. Ensure compliance with applicable exposure limits.

Personal Protection: In accordance with OSHA 29 CFR 1910.132, subpart I, wear appropriate Personal Protective Equipment (PPE) to minimize exposure to this material.

Respiratory Protection: If workplace conditions warrant a respirator, a respiratory protection program that meets OSHA 29CFR 1910.134 must be followed. Refer to NIOSH 42 CFR 84 for applicable certified respirators.

Eye/Face Protection: Wear splash resistant safety goggles with a face shield. An eyewash station should be readily available near areas of use.

Skin and Body Protection: Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product. Chemical-resistant gloves should be worn at all times when handling chemicals.

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9. PHYSICAL AND CHEMICAL PROPERTIES

Stability:X Stable Unstable Possible Hazardous Reactions: None listed. Conditions to Avoid: Contact with combustible or incompatible materials. Incompatible Materials: Acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides. Fire/Explosion Information: See Section 5, "Fire Fighting Measures". Hazardous Decomposition: Thermal decomposition will produce oxides of nitrogen, hydrogen fluoride, tantalum and other miscellaneous decomposition products. Hazardous Polymerization: Will Occur X Will Not Occur 11. TOXICOLOGICAL INFORMATION Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and						
Molar Mass (g/mol): Molar Mass (g/mol): Odor: irritating odor Odor threshold (mg/m³): pH: acidic Evaporation rate: Netting point/freezing point: Relative Density as specific gravity (water = 1): Native Density as specific gravity (water = 1): Not available Not available Not available Not available Vapor Pressure (mmHg): Vapor Density (air = 1): Not available Not available Solubility(ies): Solubility(ies): Solubility(ies): Partition coefficient (n-octanol/water): Thermal Stability Properties Autoignition Temperature: Thermal Decomposition: Initial boiling point and boiling range: Explosive Limits, LEL (Volume %): Explosive Limits, LEL (Volume %): Flash Point: Flammability (solid, gas): 10. STABILITY AND REACTIVITY Reactivity: Stable at normal temperatures and pressure. Stability: X Stable Unstable Possible Hazardous Reactions: None listed. Conditions to Avoid: Contact with combustible or incompatible materials. Incompatible Materials: Acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides. Fire/Explosion Information: See Section 5, "Fire Fighting Measures". Hazardous Decomposition: Thermal decomposition will produce oxides of nitrogen, hydrogen fluoride, tantalum and other miscellaneous decomposition products. Hazardous Polymerization: Will Occur X Will Not Occur 11. TOXICOLOGICAL INFORMATION Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and sting and contermine and to the physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and sting and contermines and position of the physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and the physical chemical and Toxicological Characteristics: Acid burns to skin, eyes, and the physical chemical and Toxicological Characteristics: Acid burns to skin, eyes	Descriptive Properties					
Molar Mass (g/mol):	Appearance (physical state, color, etc.):	colorless to yellow liquid				
Odor: irritating odor Odor threshold (mg/m³): not available acidic pH: acidic Evaporation rate: not available Melting point/freezing point: not available Relative Density as specific gravity (water = 1): not available Napor Pressure (mmlfg): not available Vapor Density (air = 1): not available Viscosity: not available Solubility(ies): soluble with water and alcohols Partition coefficient (n-octanol/water): not available Thermal Stability Properties Autoignition Temperature: not applicable Initial boiling point and boiling range: not available Explosive Limits, LEL (Volume %): not applicable Initial boiling point and boiling range: not applicable Explosive Limits, UEL (Volume %): not applicable Flash Point: not applicable Flammability (solid, gas): not applicable Flammability (solid, gas): not applicable Inot applicable Flammability (solid, gas): not applicable Flammability (solid, gas): not applicable Inot applicable Flammability (solid, gas): not applicable Flammability (solid, ga	Molecular Formula:	not applicable				
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Incompatible Materials: Acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals halogens, metal salts, metal oxides, reducing agents, peroxides, metal carbide, cyanides. Fire/Explosion Information: See Section 5, "Fire Fighting Measures". Hazardous Decomposition: Thermal decomposition will produce oxides of nitrogen, hydrogen fluoride, tantalum and other miscellaneous decomposition products. Hazardous Polymerization: Will Occur X Will Not Occur 11. TOXICOLOGICAL INFORMATION Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	Possible Hazardous Reactions: None listed.					
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Hazardous Decomposition: Thermal decomposition will produce oxides of nitrogen, hydrogen fluoride, tantalum and other miscellaneous decomposition products. Hazardous Polymerization: Will Occur X Will Not Occur 11. TOXICOLOGICAL INFORMATION Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	Incompatible Materials: Acids, combustible materials, halo carbons, amines, bases, oxidizing materials, metals,					
Hazardous Polymerization: Will Occur X Will Not Occur 11. TOXICOLOGICAL INFORMATION Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	Fire/Explosion Information: See Section 5, "Fire Fighting Mea	asures".				
11. TOXICOLOGICAL INFORMATION Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	Hazardous Decomposition: Thermal decomposition will produce and other miscellaneous decomposition products.	uce oxides of nitrogen, hydrogen fluoride, tantalum,				
Route of Exposure: X Inhalation X Skin X Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	Hazardous Polymerization: Will Occur X	Will Not Occur				
Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	11. TOXICOLOGICAL INFORMATION					
Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Acid burns to skin, eyes, and	Route of Exposure: X Inhalation X Skin	X Ingestion				

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Potential Health Effects (Acute, Chronic and Delayed):

Inhalation: Inhalation of acid fumes can damage the mucous membranes and upper respiratory tract. Short term exposure may cause irritation and inflammation of the upper respiratory tract, coughing, choking, sore throat, shortness of breath, headache, dizziness, and nausea. Long term exposure to acid fumes may cause damage to teeth, bronchial irritation, chronic cough, bronchial pneumonia, and gastrointestinal disturbances.

Skin Contact: Nitric and hydrofluoric acid can cause severe skin burns. Severity of the damage depends on the concentration and duration of exposure. Effects of acid burns may be delayed.

Eye Contact: Nitric and hydrofluoric acid can cause severe eye irritation, corneal burns, permanent eye damage, or blindness. Severity of the damage depends on the concentration and duration of exposure.

Ingestion: If ingested, nitric and hydrofluoric acid can cause severe burns and damage to the gastrointestinal tract.

Numerical Measures of Toxicity

Acute Toxicity: Not classified.

Nitric acid: Rat, Inhalation LC50: 130 mg/m³ (4 h)

Hydrofluoric acid: Rat, Inhalation LC50: 1276 ppm (1 h); 1100 mg/m³ (1 h)

Tantalum nitrate: no data available.

Skin Corrosion/Irritation: This SRM contains >1 % of nitric acid and hydrofluoric acid and it is classified as Category 1B.

Serious Eye Damage/Eye Irritation: This SRM contains >1 % nitric acid and hydrofluoric acid and it is classified as Category 1.

Respiratory Sensitization: No data available.

Skin Sensitization: No data available.

Germ Cell Mutagenicity: No data available.

Carcinogenicity: Not classified.

Listed as a Carcinogen/Potential Carcinogen Yes X No

Nitric acid, hydrofluoric acid, and tantalum nitrate are not listed by NTP, IARC or OSHA as carcinogens/potential carcinogens.

Reproductive Toxicity: Not classified.

Nitric acid: Rat, Oral TDLo: 21 150 mg/kg (pregnant 1 d to 21 d)

Rat, Oral TDLo: 2345 mg/kg (pregnant 18 d)

Hydrofluoric acid: Rat, Inhalation TCLo: 470 μg/m³ (4 h, pregnant 1 d to 22 d)

Specific Target Organ Toxicity, Single Exposure: No data available.

Specific Target Organ Toxicity, Repeated Exposure: No data available.

Aspiration Hazard: No data available.

12. ECOLOGICAL INFORMATION

Ecotoxicity Data

Nitric acid: Fish, hooknose or pogge (Agonus cataphractus) LC50: 100 mg/L to 330 mg/L (48 h)

Hydrofluoric acid: Invertebrate, water flea (Daphnia magna) EC50: 270 mg/L (48 h)

Tantalum nitrate: No data available.

Persistence and Degradability: No data available.

Bioaccumulative Potential: No data available.

Mobility in Soil: No data available.

Other Adverse effects: No data available.

13. DISPOSAL CONSIDERATIONS

Waste Disposal: Dispose of waste in accordance with all applicable federal, state, and local regulations. Subject to disposal regulations: U.S. EPA 40 CFR 262, Hazardous Waste Numbers: nitric acid (D001, D002) and hydrofluoric acid (U134).

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14. TRANSPORTATION INFORMATION

U.S. DOT and IATA: UN1760, Corrosive liquid, n.o.s. (contains nitric acid and hydrofluoric acid), Hazard Class 8, Packing Group II.

15. REGULATORY INFORMATION

U.S. Regulations:

C ERCLA Sections 102a/103 (40 CFR 302.4): Nitric acid, 1000 lbs. (454 kg) final RQ

Hydrofluoric acid, 100 lbs. (45.4 kg) final RQ

SARA Title III Section 302 (40 CFR 355.30): Nitric acid, 1000 lbs. (454 kg) TPQ

Hydrofluoric acid, 100 lbs. (45.4 kg) TPQ

SARA Title III Section 304 (40 CFR 355.40): Nitric acid, 1000 lbs. (454 kg) EPCRA RQ

Hydrofluoric acid, 100 lbs. (45.4 kg) EPCRA RQ

SARA Title III Section 313 (40 CFR 372.65): 1 % de minimis concentration for nitric acid and hydrofluoric

acid.

OSHA Process Safety (29 CFR 1910.119): Regulated for nitric acid at higher concentrations

500 lbs. TQ (≥94.5 % by weight).

Hydrofluoric acid, 1000 lbs. (454 kg) TQ (anhydrous)

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE HEALTH: Yes
CHRONIC HEALTH: No
FIRE: No
REACTIVE: No
PRESSURE: No

State Regulations: Not listed under California Proposition 65.

U.S. TSCA Inventory: Nitric acid and hydrofluoric acid are listed. Tantalum nitrate is not listed.

TSCA 12(b), Export Notification: Not listed.

Canadian Regulations: WHMIS Information is not provided for this material.

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16. OTHER INFORMATION

Issue Date: 02 April 2020

Sources: ChemAdvisor, Inc., SDS *Nitric Acid*, 09 December 2015.

ChemAdvisor, Inc., SDS Hydrofluoric Acid, 09 December 2015.

U.S. National Library of Medicine, Nitric Acid; available at https://pubchem.ncbi.nlm.nih.gov/ (accessed

Apr 2020).

Key of Acronyms:

ACGIH	American Conference of Governmental Industrial	NRC	Nuclear Regulatory Commission
	Hygienists) ITTD	N. J. T. J. D.
ALI	Annual Limit on Intake	NTP	National Toxicology Program
CAS	Chemical Abstracts Service	OSHA	Occupational Safety and Health Administration
CERCLA	Comprehensive Environmental Response,	PEL	Permissible Exposure Limit
	Compensation, and Liability Act		
CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
DOT	Department of Transportation	REL	Recommended Exposure Limit
EC50	Effective Concentration, 50 %	RM	Reference Material
EINECS	European Inventory of Existing Commercial	RQ	Reportable Quantity
	Chemical Substances		
EPCRA	Emergency Planning and Community Right-to-Know	RTECS	Registry of Toxic Effects of Chemical Substances
	Act		
IARC	International Agency for Research on Cancer	SARA	Superfund Amendments and Reauthorization Act
IATA	International Air Transport Association	SCBA	Self-Contained Breathing Apparatus
IDLH	Immediately Dangerous to Life and Health	SRM	Standard Reference Material
LC50	Lethal Concentration, 50 %	STEL	Short Term Exposure Limit
LD50	Lethal Dose, 50 %	STOT	Specific Target Organ Toxicity
LEL	Lower Explosive Limit	TLm	Threshold Limit, median
MSDS	Material Safety Data Sheet	TLV	Threshold Limit Value
NFPA	National Fire Protection Association	TPO	Threshold Planning Quantity
NIOSH	National Institute for Occupational Safety and Health	TSCA	Toxic Substances Control Act
NIST	National Institute of Standards and Technology	TWA	Time Weighted Average
n.o.s.	Not Otherwise Specified	UEL	Upper Explosive Limit
	1100 Galler 11100 Speemed	WHMIS	Workplace Hazardous Materials Information System
		*** 11171115	" orapiace riazardous materials information system

Disclaimer: Physical and chemical data contained in this SDS are provided only for use in assessing the hazardous nature of the material. The SDS was prepared carefully, using current references; however, NIST does not certify the data in the SDS. The certified values for this material are given in the NIST Certificate of Analysis.

Users of this SRM should ensure that the SDS in their possession is current. This can be accomplished by contacting the SRM Program: telephone (301) 975-2200; fax (301) 948-3730; e-mail srmmsds@nist.gov; or via the Internet at https://www.nist.gov/srm.

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