(North America)

(International)



# SAFETY DATA SHEET

## 1. SUBSTANCE AND SOURCE IDENTIFICATION

**Product Identifier** 

**SRM Number:** 2630

SRM Name: Nitric Oxide in Nitrogen (Nominal Amount-of-Substance Fraction 1500 µmol/mol)

Other Means of Identification: Not Applicable.

#### Recommended Use of This Material and Restrictions of Use

This Standard Reference Material (SRM) is a primary gas mixture of nitric oxide in nitrogen supplied in a DOT 3AL-specification aluminum (6061 alloy) cylinder with a water volume of 6 L. Mixtures are shipped with a nominal pressure exceeding 12.4 MPa (1800 psig), which provides the user with 0.73 m<sup>3</sup> (25.8 ft<sup>3</sup>) of useable mixture. The cylinder is the property of the purchaser and is equipped with a CGA-660 stainless steel valve, which is the recommended outlet for this nitric oxide mixture.

#### **Company Information**

National Institute of Standards and Technology Standard Reference Materials Program 100 Bureau Drive, Stop 2300 Gaithersburg, Maryland 20899-2300

Telephone: 301-975-2200 Emergency Telephone ChemTrec: FAX: 301-948-3730 1-800-424-9300 E-mail: SRMMSDS@nist.gov +1-703-527-3887

Website: http://www.nist.gov/srm

## 2. HAZARDS IDENTIFICATION

Classification

Physical Hazard: Compressed Gas **Health Hazard:** Simple Asphyxiant

## **Label Elements** Symbol



#### Signal Word WARNING

#### **Hazard Statement(s)**

H280 Contains gas under pressure; may explode if heated. May displace oxygen and cause rapid suffocation.

#### **Precautionary Statement(s)**

P410+P403 Protect from sunlight. Store in a well-ventilated place.

Hazards Not Otherwise Classified: Not applicable.

Ingredients(s) with Unknown Acute Toxicity: Not applicable.

## 3. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Substance: Nitric oxide in nitrogen, compressed gas

# Other Designations:

Nitric oxide: Nitrogen oxide; nitrogen monoxide; nitric oxide trimer; NO.

Nitrogen: Dinitrogen; nitrogen compressed.

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<b>Hazardous Components</b>	CAS Number	EC Number (EINECS)	Nominal Mass Concentration (%)
Nitrogen	7727-37-9	231-783-9	>99
Nitric oxide	10102-43-9	233-271-0	0.15

#### 4. FIRST AID MEASURES

#### **Description of First Aid Measures:**

**Inhalation:** If adverse effects occur, remove to uncontaminated area. If not breathing, give artificial respiration or oxygen by qualified personnel. Seek immediate medical attention.

Skin Contact: Not applicable.

**Eye Contact:** Immediately flush eyes, including under the eyelids with copious amounts of water for at least 15 minutes. Seek immediate medical attention.

**Ingestion:** Ingestion of a gas is unlikely.

**Most Important Symptoms/Effects, Acute and Delayed:** Potentially fatal if inhaled, blood damage, respiratory tract irritation, skin irritation, eye irritation.

**Indication of any immediate medical attention and special treatment needed, if necessary:** If any of the above symptoms are present, seek immediate medical attention.

## 5. FIRE FIGHTING MEASURES

**Fire and Explosion Hazards:** Negligible fire hazard applicable to the identified NIST cylinder. Cylinders may rupture or explode if exposed to heat. See Section 9, "Physical and Chemical Properties" for flammability properties.

#### **Extinguishing Media:**

Suitable: Use extinguishing media appropriate to the surrounding fire.

Unsuitable: None listed.

**Specific Hazards Arising from the Chemical:** Oxides of nitrogen.

**Special Protective Equipment and Precautions for Fire-Fighters:** Move cylinder from fire area if it can be done without personal risk. Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH approved self-contained breathing apparatus (SCBA).

**NFPA Ratings** (0 = Minimal; 1 = Slight; 2 = Moderate; 3 = Serious; 4 = Severe)

Health = 3 Fire = 0 Reactivity = 0

## 6. ACCIDENTAL RELEASE MEASURES

**Personal Precautions, Protective Equipment and Emergency Procedures:** Immediately contact emergency personnel. Keep unnecessary personnel away. Use suitable protective equipment; see Section 8, "Exposure Controls and Personal Protection". Shut off gas supply if this can be done safely. Isolate area until gas has dispersed.

**Methods and Materials for Containment and Clean up:** Stop leak if possible without personal risk. Isolate hazard area and deny entry. Ventilate closed spaces before entering.

#### 7. HANDLING AND STORAGE

**Safe Handling Precautions:** Secure cylinder to prevent physical damage. Close valve after each use and when empty. Keep value protection cap on cylinder when not in use.

**Storage:** Store and handling in accordance with all current regulations and standards. Keep separated from incompatible substances. Store in well-ventilated area. Subject to storage regulations, OSHA 29 CFR 1910.101.

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#### 8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

#### **Exposure Limits**

#### Nitric oxide

 NIOSH (REL):
 25 ppm (30 mg/m³) TWA

 NIOSH (REL):
 100 ppm (123 mg/m³) IDLH

 ACGIH (TLV):
 25 ppm (30 mg/m³) TWA

 OSHA (PEL):
 25 ppm (30 mg/m³) TWA

#### Nitrogen

ACGIH (TLV): Simple asphyxiant.

**Engineering Controls:** Provide local exhaust or process enclosure ventilation system. Ensure compliance with applicable exposure limits.

**Personal Protection Measures:** In accordance with OSHA 29 CFR 1910.132, subpart I, wear appropriate Personal Protective Equipment (PPE) to minimize exposure to this material.

**Respiratory Protection:** If workplace conditions warrant a respirator, a respiratory protection program that meets OSHA 29CFR 1910.134 must be followed. Refer to NIOSH 42 CFR 84 for applicable certified respirators.

**Eye/Face Protection:** Wear splash resistant safety goggles with a face shield. An eye wash station should be readily available near areas of use.

**Skin and Body Protection:** Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product. Chemical-resistant gloves should be worn at all times when handling chemicals.

#### 9. PHYSICAL AND CHEMICAL PROPERTIES

**NOTE:** The physical and chemical data provided are for the pure components. No physical or chemical data are available for this compressed gas mixture of nitric oxide and nitrogen. The actual behavior of the gas mixture may differ from the individual components.

<b>Descriptive Properties:</b>	Nitrogen (99 %)	Nitric Oxide (0.15 %)	
Appearance (physical state, color, etc.)	colorless compressed gas	colorless compressed gas	
Molecular Formula	$N_2$	NO	
Molar Mass (g/mol)	28	30	
Odor	odorless	pungent odor	
Odor threshold	not available	0.3 ppm to 1 ppm	
pН	not applicable	not available	
Evaporation rate	not applicable	not available	
Melting point/freezing point	−210 °C (−346 °F)	-164 °C to $-161$ °C ( $-263$ °F to $-258$ °F)	
Relative Density (g/L)	1.2506	not available	
Vapor Pressure	760 mmHg at −196 °C	760 mmHg at -151.7 °C	
Vapor Density (air = 1)	0.967	Highest known value: 1.04; Weighted average: 0.97	
Viscosity (cP)	0.01787 at 27 °C	0.0188 at 25 ℃	
Solubility(ies)	water (1.6 % at 20 °C) liquid ammonia	water (4.6 %) sulfuric acid, alcohol, ferrous sulfate solutions, carbon disulfide	
Partition coefficient (n-octanol/water)	not available	not available	
Thermal Stability Properties Autoignition Temperature not applicable		not available	
-	**	-153 °C to -152 °C (-243 °F to -242 °F)	
Thermal Decomposition	not applicable -196 °C (-321 °F)	not available	
Initial boiling point and boiling range	` /	not available	
Explosive Limits, LEL (Volume %)	not applicable		
Explosive Limits, UEL (Volume %)	not applicable	not available	

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**Descriptive Properties:** Nitrogen Nitric Oxide (99 %) (0.15%)**Flash Point** not applicable not available Flammability (solid, gas) not applicable not available 10. STABILITY AND REACTIVITY **Reactivity:** Stable at normal temperature and pressure. Stable **Stability:** X Unstable Possible Hazardous Reactions: None listed. Conditions to Avoid: Avoid heat, flames, sparks, and other sources of ignition. Minimize contact with material. Containers may rupture or explode if exposed to heat. Incompatible Materials: Oxidizing materials, halogens, metal oxides, metals, bases, combustible materials, metal salts, halocarbons, reducing agents, metal carbide. Fire/Explosion Information: See Section 5, "Fire Fighting Measures". **Hazardous Decomposition:** Miscellaneous decomposition products. **Hazardous Polymerization:** Will Occur X Will Not Occur 11. TOXICOLOGICAL INFORMATION **Route of Exposure:** X Inhalation Skin Ingestion Symptoms Related to the Physical, Chemical and Toxicological Characteristics: Nausea, headache, weakness, drowsiness. Potential Health Effects (Acute, Chronic, and Delayed): Inhalation: Nitric oxide: Acute exposure to nitric oxide at low concentrations in a one-time exposure may cause mild coughing, respiratory tract irritation, fatigue, headache, nausea, and vomiting choking abdominal pain, and dyspnea. Exposure to 25 ppm for 60 minutes may cause irritation and minor chest pain; 50 ppm may lead to pulmonary edema with possible recovery; 100 ppm may cause pulmonary edema and death. Nitrogen: Nitrogen compressed gas is a simple asphyxiant. Release in an enclosed space may result in asphyxiation. The symptoms of asphyxia depend on the rapidity with which the oxygen deficiency develops and how long it continues. In sudden acute asphyxia, unconsciousness may be immediate. With slow development, there may be rapid respiration and pulse, air hunger, dizziness, reduced awareness, tightness in the head, tingling sensations, incoordination, faulty judgment, emotional instability, and rapid fatigue. As the asphyxia progresses, nausea, vomiting, collapse, unconsciousness, convulsions, deep coma, and death are possible. **Skin Contact:** Nitric oxide at low concentrations may cause skin irritation. Eye Contact: Nitric oxide at low concentrations in air may cause a stinging sensation of the eyes. Not applicable for exposure to nitrogen. **Ingestion:** Ingestion of a gas is unlikely under normal conditions of use. **Numerical Measures of Toxicity:** 

Acute Toxicity: Not classified.

Nitric oxide: Rat, Inhalation LC50: 1068 mg/m<sup>3</sup> (4 h); 115 ppm (1 h)

Nitrogen: Simple asphyxiant.

**Skin Corrosion/Irritation:** Not applicable.

Serious Eye Damage/ Eye Irritation: Not applicable.

**Respiratory Sensitization:** No data available.

**Skin Sensitization:** No data available.

Germ Cell Mutagenicity: No data available.

SRM 2630 Page 4 of 6 Carcinogenicity: Not classified.

Listed as a Carcinogen/Potential Carcinogen

Yes X No

Nitrogen and nitric oxide are not listed by NTP, IARC or OSHA as carcinogens/potential carcinogens.

**Reproductive Toxicity:** No data available.

Specific Target Organ Toxicity, Single Exposure: Not classified.

Specific Target Organ Toxicity, Repeated Exposure: No data available.

**Aspiration Hazard:** Not classified.

# 12. ECOLOGICAL INFORMATION

Ecotoxicity Data: No data available.

**Persistence and Degradability:** No data available. **Bioaccumulative Potential:** No data available.

**Mobility in Soil:** No data available.

Other Adverse effects: No data available.

#### 13. DISPOSAL CONSIDERATIONS

Waste Disposal: Dispose of waste in accordance with all applicable federal, state, and local regulations.

## 14. TRANSPORTATION INFORMATION

U.S. DOT and IATA: UN1956, compressed gas, n.o.s. (nitric oxide in nitrogen), Hazard Class 2.2.

#### 15. REGULATORY INFORMATION

#### **U.S. Regulations:**

CERCLA Sections 102a/103 (40 CFR 302.4): Identified cylinder not regulated.

SARA Title III Section 302 (40 CFR 355.30): Identified cylinder not regulated.

SARA Title III Section 304 (40 CFR 355.40): Identified cylinder not regulated.

SARA Title III Section 313 (40 CFR 372.65): Identified cylinder not regulated.

OSHA Process Safety (29 CFR 1910.119): Identified cylinder not regulated.

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE HEALTH: No
CHRONIC HEALTH: No
FIRE: No
REACTIVE: No
PRESSURE: Yes

## **State Regulations:**

California Proposition 65: Not listed.

U.S. TSCA Inventory: Nitric oxide and nitrogen are listed.

TSCA 12(b), Export Notification: No components are listed.

**Canadian Regulations:** WHMIS Information is not provided for this material.

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## 16. OTHER INFORMATION

Issue Date: 17 May 2017

**Sources:** ChemADVISOR, Inc., MSDS, *Nitrogen, Compressed Gas*, 09 December 2015.

ChemADVISOR, Inc., MSDS, Nitric Oxide, 09 December 2015.

#### **Key of Acronyms:**

-	Leg of field	11, 1115.		
	ACGIH	American Conference of Governmental Industrial	NRC	Nuclear Regulatory Commission
		Hygienists	. Imp	
	ALI	Annual Limit on Intake	NTP	National Toxicology Program
	CAS	Chemical Abstracts Service	OSHA	Occupational Safety and Health Administration
	CERCLA	Comprehensive Environmental Response,	PEL	Permissible Exposure Limit
		Compensation, and Liability Act		•
	CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
	DOT	Department of Transportation	REL	Recommended Exposure Limit
	EINECS	European Inventory of Existing Commercial	RQ	Reportable Quantity
		Chemical Substances		
	EPCRA	Emergency Planning and Community Right-to-Know	RTECS	Registry of Toxic Effects of Chemical Substances
		Act		•
	IARC	International Agency for Research on Cancer	SARA	Superfund Amendments and Reauthorization Act
	IATA	International Air Transport Association	SCBA	Self-Contained Breathing Apparatus
	IDLH	Immediately Dangerous to Life and Health	SRM	Standard Reference Material
	LC50	Lethal Concentration	STEL	Short Term Exposure Limit
	LD50	Median Lethal Dose or Lethal Dose, 50 %	STOT	Specific Target Organ Toxicity
	LEL	Lower Explosive Limit	TLV	Threshold Limit Value
	MSDS	Material Safety Data Sheet	TPQ	Threshold Planning Quantity
	NFPA	National Fire Protection Association	TSCA	Toxic Substances Control Act
	NIOSH	National Institute for Occupational Safety and Health	TWA	Time Weighted Average
	NIST	National Institute of Standards and Technology	UEL	Upper Explosive Limit
	n.o.s.	Not otherwise specified	WHMIS	Workplace Hazardous Materials Information System
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**Disclaimer:** Physical and chemical data contained in this SDS are provided only for use in assessing the hazardous nature of the material. The SDS was prepared carefully, using current references; however, NIST does not certify the data in the SDS. The certified values for this material are given in the NIST Certificate of Analysis.

Users of this SRM should ensure that the SDS in their possession is current. This can be accomplished by contacting the SRM Program: telephone (301) 975-2200; fax (301) 948-3730; e-mail srmmsds@nist.gov; or via the Internet at http://www.nist.gov/srm.

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