

## Corrigendum

### Corrigendum to “Update of NIST half-life results corrected for ionization chamber source-holder instability”

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Unterweger and Fitzgerald (2014) published corrections to earlier NIST half-life values (Unterweger, 2002) to account for ionization chamber source holder instability.

In preparing the publication, a mistake in the propagation of the correction factor,  $f$ , and its associated uncertainty led to incorrect results in Table 4 of that publication.

Revised values are given below in Table 1. The data in Table 1 are a summary of the results reported on earlier by Fitzgerald (2012) in his Table 6 and should replace those given in Unterweger and Fitzgerald (2014). The authors regret the error.

## References

Fitzgerald, R., 2012. NIST Ionization Chamber "A" Sample-Height Corrections. J Res. Natl. Inst. Stand. Technol. 117, 80-95. <http://dx.doi.org/10.6028/jres.117.003>

Unterweger, M. P., 2002. Half-life measurements at the National Institute of Standards and Technology. Appl. Radiat. Isot. 56, 125-130.  
[https://doi.org/10.1016/S0969-8043\(01\)00177-4](https://doi.org/10.1016/S0969-8043(01)00177-4)

Unterweger, M.P., Fitzgerald, R., 2014. Update of NIST half-life results corrected for ionization chamber source-holder instability, Appl. Radiat. Isot. 87, 92-94.  
<https://doi.org/10.1016/j.apradiso.2013.11.017>

**Table 1** Half-lives with combined standard uncertainties summarized from Table 6 in (Fitzgerald, 2012). For  $^{99m}\text{Tc}$ , \* represents saline and \*\* acid solutions.

Nuclide	Half-life	Nuclide	Half-life
$^{18}\text{F}$	( $1.8295 \pm 0.0003$ ) h	$^{131m}\text{Xe}$	( $11.934 \pm 0.021$ ) d
$^{22}\text{Na}$	( $950.4 \pm 0.4$ ) d	$^{133}\text{Ba}$	( $3832.3 \pm 10.8$ ) d
$^{24}\text{Na}$	( $14.951 \pm 0.003$ ) h	$^{133}\text{Xe}$	( $5.2474 \pm 0.0005$ ) d
$^{32}\text{P}$	( $14.262 \pm 0.003$ ) d	$^{134}\text{Cs}$	( $753.43 \pm 0.28$ ) d
$^{46}\text{Sc}$	( $83.828 \pm 0.066$ ) d	$^{137}\text{Cs}$	( $10915 \pm 55$ ) d
$^{51}\text{Cr}$	( $27.6999 \pm 0.0013$ ) d	$^{139}\text{Ce}$	( $137.7 \pm 0.09$ ) d
$^{54}\text{Mn}$	( $311.97 \pm 0.05$ ) d	$^{140}\text{Ba}$	( $12.7525 \pm 0.0023$ ) d
$^{57}\text{Co}$	( $271.95 \pm 0.27$ ) d	$^{140}\text{La}$	( $40.293 \pm 0.012$ ) h
$^{58}\text{Co}$	( $70.77 \pm 0.11$ ) d	$^{141}\text{Ce}$	( $32.508 \pm 0.024$ ) d
$^{59}\text{Fe}$	( $44.507 \pm 0.007$ ) d	$^{144}\text{Ce}$	( $284.35 \pm 0.08$ ) d
$^{60}\text{Co}$	( $1924 \pm 0.9$ ) d	$^{152}\text{Eu}$	( $4929 \pm 10$ ) d
$^{62}\text{Cu}$	( $9.672 \pm 0.008$ ) m	$^{153}\text{Gd}$	( $239.29 \pm 0.1$ ) d
$^{65}\text{Zn}$	( $244.14 \pm 0.1$ ) d	$^{153}\text{Sm}$	( $46.285 \pm 0.003$ ) h
$^{67}\text{Ga}$	( $3.2615 \pm 0.0005$ ) d	$^{154}\text{Eu}$	( $3138 \pm 4$ ) d
$^{75}\text{Se}$	( $119.78 \pm 0.07$ ) d	$^{155}\text{Eu}$	( $1731 \pm 3$ ) d
$^{85}\text{Kr}$	( $3905 \pm 19$ ) d	$^{166}\text{Ho}$	( $26.794 \pm 0.023$ ) h
$^{85}\text{Sr}$	( $64.848 \pm 0.008$ ) d	$^{169}\text{Yb}$	( $32.011 \pm 0.009$ ) d
$^{88}\text{Y}$	( $106.62 \pm 0.04$ ) d	$^{177}\text{Lu}$	( $6.64 \pm 0.01$ ) d
$^{99}\text{Mo}$	( $65.924 \pm 0.006$ ) h	$^{181}\text{W}$	( $121.03 \pm 0.07$ ) d
$^{99m}\text{Tc}^*$	( $6.0072 \pm 0.0009$ ) h	$^{186}\text{Re}$	( $89.25 \pm 0.07$ ) h
$^{99m}\text{Tc}^{**}$	( $6.012 \pm 0.003$ ) h	$^{188}\text{Re}$	( $17.001 \pm 0.022$ ) h
$^{103}\text{Ru}$	( $39.31 \pm 0.04$ ) d	$^{188}\text{W}$	( $69.77 \pm 0.05$ ) d
$^{109}\text{Cd}$	( $462.6 \pm 0.7$ ) d	$^{192}\text{Ir}$	( $73.802 \pm 0.019$ ) d
$^{110m}\text{Ag}$	( $249.91 \pm 0.03$ ) d	$^{195}\text{Au}$	( $186.01 \pm 0.06$ ) d
$^{111}\text{In}$	( $2.8048 \pm 0.0005$ ) d	$^{198}\text{Au}$	( $2.69516 \pm 0.00021$ ) d
$^{113}\text{Sn}$	( $115.06 \pm 0.08$ ) d	$^{201}\text{Tl}$	( $3.0456 \pm 0.0015$ ) d
$^{117m}\text{Sn}$	( $14 \pm 0.05$ ) d	$^{202}\text{Tl}$	( $12.47 \pm 0.08$ ) d
$^{123}\text{I}$	( $13.2235 \pm 0.0019$ ) h	$^{203}\text{Hg}$	( $46.615 \pm 0.027$ ) d
$^{125}\text{I}$	( $59.47 \pm 0.13$ ) d	$^{203}\text{Pb}$	( $51.92 \pm 0.04$ ) h
$^{125}\text{Sb}$	( $1006.5 \pm 0.6$ ) d	$^{207}\text{Bi}$	( $11403 \pm 61$ ) d
$^{127}\text{Xe}$	( $36.342 \pm 0.003$ ) d	$^{228}\text{Th}$	( $698.4 \pm 0.4$ ) d
$^{131}\text{I}$	( $8.0196 \pm 0.0022$ ) d		

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