

**2018 CODATA RECOMMENDED VALUES OF THE FUNDAMENTAL  
CONSTANTS OF PHYSICS AND CHEMISTRY NIST SP 959** (June 2019)

An extensive constants list is available at [physics.nist.gov/constants](http://physics.nist.gov/constants).

Quantity	Symbol	Numerical value	Unit
${}^{133}\text{Cs}$ hyperfine transition frequency	$\Delta\nu_{\text{Cs}}$	9 192 631 770	Hz
*speed of light in vacuum	$c$	299 792 458	$\text{m s}^{-1}$
*Planck constant	$h$	6.626 070 15 $\times 10^{-34}$	$\text{J Hz}^{-1}$
	$\hbar$	1.054 571 817 ... $\times 10^{-34}$	$\text{J s}$
*elementary charge	$e$	1.602 176 634 $\times 10^{-19}$	C
*Avogadro constant	$N_A$	6.022 140 76 $\times 10^{23}$	$\text{mol}^{-1}$
*Boltzmann constant	$k$	1.380 649 $\times 10^{-23}$	$\text{J K}^{-1}$
*luminous efficacy	$K_{\text{cd}}$	683	$\text{lm W}^{-1}$
electron volt ( $e/\text{C}$ ) J	$\text{eV}$	1.602 176 634 $\times 10^{-19}$	J
Josephson constant $2e/h$	$K_J$	483 597.848 4 ... $\times 10^9$	$\text{Hz V}^{-1}$
von Klitzing constant $2\pi\hbar/e^2$	$R_K$	25 812.807 45...	$\Omega$
molar gas constant $N_A k$	$R$	8.314 462 618...	$\text{J mol}^{-1} \text{K}^{-1}$
Stefan-Boltzmann const. $\pi^2 k^4/(60\hbar^3 c^2)$	$\sigma$	5.670 374 419 ... $\times 10^{-8}$	$\text{W m}^{-2} \text{K}^{-4}$

\*Defining constants of the International System of Units (SI).

Quantity	Symbol	Numerical value	Unit
(unified) atomic mass unit $\frac{1}{12}m(^{12}\text{C})$	$u$	$1.660\,539\,066\,60(50) \times 10^{-27}$	kg
Newtonian constant of gravitation	$G$	$6.674\,30(15) \times 10^{-11}$	$\text{m}^3 \text{ kg}^{-1} \text{ s}^{-2}$
fine-structure constant $e^2/(4\pi\epsilon_0\hbar c)$	$\alpha$	$7.297\,352\,5693(11) \times 10^{-3}$	
inverse fine-structure constant	$\alpha^{-1}$	137.035 999 084(21)	
Rydberg frequency $\alpha^2 m_e c^2 / (2\hbar)$	$cR_\infty$	$3.289\,841\,960\,2508(64) \times 10^{15}$	Hz
vac. magnetic permeability $4\pi\alpha\hbar/(e^2c)$	$\mu_0$	$1.256\,637\,062\,12(19) \times 10^{-6}$	$\text{N A}^{-2}$
vac. electric permittivity $1/(\mu_0 c^2)$	$\epsilon_0$	$8.854\,187\,8128(13) \times 10^{-12}$	$\text{F m}^{-1}$
electron mass	$m_e$	$9.109\,383\,7015(28) \times 10^{-31}$	kg
proton mass	$m_p$	$1.672\,621\,923\,69(51) \times 10^{-27}$	kg
proton-electron mass ratio	$m_p/m_e$	1836.152 673 43(11)	
reduced Compton wavelength $\hbar/(m_e c)$	$\lambda_C$	$3.861\,592\,6796(12) \times 10^{-13}$	m
Bohr radius $\hbar/(\alpha m_e c)$	$a_0$	$5.291\,772\,109\,03(80) \times 10^{-11}$	m
Bohr magneton $e\hbar/(2m_e)$	$\mu_B$	$9.274\,010\,0783(28) \times 10^{-24}$	$\text{J T}^{-1}$

The number in parentheses is the one-sigma ( $1\sigma$ ) uncertainty in the last two digits of the given value.

