Simultaneous Neutron and X-Ray Imaging of 3D Structure of Organic Matter and Fracture in Shales¹

Wei-Shan Chiang^{2,3,4}, Jacob M. LaManna⁵, Daniel S. Hussey⁵, David L. Jacobson⁵, Yun Liu^{3,4,6}, Jilin Zhang², Daniel T. Georgi², Jordan R. Kone², and Jin-Hong Chen²

ABSTRACT

Hydrocarbon production from shales using horizontal drilling and hydraulic fracturing has been the key development in the US energy industry in the past decade and has now become more important globally. Nevertheless, many fundamental problems related to the storage and flow of light hydrocarbons in shales are still unknown. It has been reported that the hydrocarbons in shale rocks are predominantly stored within the kerogen pores with characteristic length scale between 1 to 100 nm. In addition, it is possible that the 3D connectivity of these kerogen pores with fractures from the micrometer to centimeter scale, form the flow path for light hydrocarbons. Therefore, to better model the gas-in-place and permeability in shales, it is necessary to quantify the structural distribution of organic and inorganic components and fractures over a large breadth of length scales.

Simultaneous neutron and X-ray tomography offers a core-scale nondestructive method that can distinguish the organic matter, inorganic minerals, and open and healed fractures in 2.5 cm diameter shales with a resolution of

INTRODUCTION

Natural gas production worldwide from shale resources is projected to grow from 1.2×10^9 m³/d in 2015 to 4.8×10^9 m³/d by 2040 (Aloulou and Zaretskaya, 2016). Shale gas is predicted to account for 30% of world natural gas production by 2040 (Aloulou and Zaretskaya, 2016). Despite the promising future of shale gas as an energy source, fundamental problems related to the storage and transport of hydrocarbons in shales still lack clear understanding. The organic matter in organic-rich source rocks may include kerogen, bitumen, and/or heavier immobile hydrocarbons. about 30 µm and a field of view of about 3 cm. In the reconstructed neutron volume, the hydrogen-rich areas, i.e., organic matter, are brighter because hydrogen has a larger attenuation coefficient and attenuates neutron intensity more significantly. For the X-ray volume, the attenuation coefficient of an element is related to its atomic number Z and the brighter areas indicate the region containing more high-Z elements, such as some heavy minerals. Open fractures do not attenuate either neutrons or X-rays and therefore look dark in both reconstructed neutron and X-ray volumes.

In this study, two shale samples from different locations were investigated using simultaneous neutron and X-ray tomography for the first time. We were able to construct 3D images of the shales and isolate 3D maps of organic matter and minerals including high-Z elements. The distribution of kerogen and fractures can be used in the modeling of hydrocarbon flow in core scale, a 10⁹ upscaling from current methods that model the flow based on SEM images.

Kerogen is imbedded within the inorganic matrix of shale rocks and does not dissolve in any solvent (Loucks et al., 2009). Previous studies have suggested that the majority of light hydrocarbons in the shale rocks are stored within the kerogen, which has pores with the characteristic length scale, between 1 to 100 nm (Ambrose et al., 2013). The total amount and the distribution of kerogens in the shales, therefore, can influence the total gas-in-place (GIP) of the shales. Moreover, the kerogens and minerals in shales have very different properties, such as pore structure (pore size, pore shape, and pore connectivity), surface chemistry, wettability, and mechanical response. These properties are

Manuscript received by the Editor March 27, 2017; revised manuscript received September 19, 2017; manuscript accepted October 5, 2017. ¹Originally presented at the SPWLA 58th Annual Logging Symposium, Oklahoma City, Oklahome, USA, June 17–21, 2017, Paper Q.

²Reservoir Engineering Technology, Aramco Research Center–Houston

³Center for Neutron Research, National Institute of Standards and Technology, Gaithersburg, Maryland

⁴Department of Chemical and Biomolecular Engineering, University of Delaware, Newark, Delaware, USA

⁵Physical Measurement Laboratory, National Institute of Standards and Technology, Gaithersburg, Maryland

⁶Department of Physics and Astronomy, University of Delaware, Newark, Delaware, USA

directly related to the interaction of hydrocarbons with pore surfaces. Thus, the structure and three-dimensional (3D) space arrangement of the components within the rocks can significantly influence the flow path of the hydrocarbons in the shales. To better understand the storage and transport properties of hydrocarbons within the shale rocks, it is necessary to identify the distribution and the structure of organic components, especially the kerogens, and inorganic components, i.e., the minerals, and fractures in the rocks.

The development of focused ion-beam scanning electron microscopy (FIB-SEM) enables the visualization of the localized pores, kerogens, and inorganic matrix in small shale samples (Loucks et al., 2009). Moreover, the FIB-SEM images have been used to construct the 3D pore-scale representations as the input to the simulation and modeling of hydrocarbon transport in the shales (Shabro et al., 2013; Jiang et al., 2017). However, the structure of the rocks and flow properties obtained from FIB-SEM are at the pore scale. There is also the need to obtain the distribution of organic matter, especially kerogen and fractures at the core scale.

In the petroleum industry, X-ray computed tomography (CT) is a common tool to characterize the structure of source rocks. X-rays interact with electrons of the materials and, therefore, the attenuation coefficient of X-rays is proportional to the atomic number, Z, of the elements in the materials. High-Z elements, such as metals, can significantly attenuate X-rays and produce high contrast in the X-ray CT image. This makes X-ray CT a powerful tool to identify the mineral distribution inside the samples. However, organic matter, which is ubiquitous in natural materials, is mainly composed of low-Z elements, such as hydrogen and carbon. These materials only have small attenuation with X-rays and are fairly transparent to X-rays. What is more, when open fractures and pores are present in the samples, it is hard for X-rays to distinguish between these "empty" spaces and the organic matter.

Unlike X-rays, neutrons interact with the atomic nuclei of elements inside the materials. The attenuation coefficient for neutrons has little correlation with Z (Banhart et al., 2010). In particular, hydrogen, which is the most transparent element for X-rays and is very abundant in organic matter, has a very large neutron attenuation coefficient. The high sensitivity of neutrons to organic matter makes neutron imaging a complementary tool to the widely used X-ray CT for thoroughly and correctly characterizing the structure and components of the heterogeneous shale samples at the core scale. Dual neutron and X-ray tomography has been successfully applied to many other materials, such as fuel cells (Manke et al., 2007; Banhart et al., 2010), batteries (Manke et al., 2007), and cultural heritage objects (Mannes et al., 2014; 2015).

In this study, for the first time, "simultaneous" dual X-ray and neutron imaging is used to study shales at

the core scale at the National Institute of Standards and Technology (NIST) Center for Neutron Research (NCNR). The distributions of minerals, organic matter and fractures are successfully identified in a Barnett Shale sample and a Middle East shale sample. The distribution of organic matter is found to have very different configuration and distribution in these two shales.

METHOD

Attenuation of X-Rays and Neutrons By Materials

X-ray and neutron tomography are both noninvasive methods. When an incident X-ray or neutron beam with intensity I_0 interacts with a material, the X-ray or neutron intensity is attenuated by different interaction processes. For a homogeneous medium with thickness *d*, the transmitted intensity, $I_{,0}$ of X-rays or neutrons can be described by the Lambert-Beer law:

$$I_t = I_0 e^{-\mu \cdot d},\tag{1}$$

where μ is the attenuation coefficient, which is a material property and depends on the incident radiation, i.e., X-ray or neutron, and the incident radiation energy. μ can be related to atomic composition of the material by

$$\mu = \sum_{i} \sigma_{ti} N_{i}, \tag{2}$$

where $\sigma_{t,i}$ is the total attenuation cross section for X-rays or neutrons and N_i is the number density of type *i* atom. μ is very different for X-rays and for neutrons because the X-rays interact with electrons while neutrons interact with nuclei in the materials.

For X-rays, the intensity attenuation is due to Thomson (coherent) scattering, photoelectric absorption, Compton scattering, and pair production. Thus, the total cross section for X-rays for atom *i*, σ_{it}^{x-ray} , can be written as

$$\sigma_{i,t}^{x-ray=}\sigma_i^{TS} + \sigma_i^{PA} + \sigma_i^{CS} + \sigma_i^{PP},$$
(3)

where σ_i^{TS} , σ_i^{PA} , σ_i^{CS} , and σ_i^{PP} are X-ray atomic cross sections for Thomson scattering, photoelectric absorption, Compton scattering, and pair production, respectively. Pair production is ignored because the peak energy of 90 keV of the X-ray tube used in this work is well below the 1.022-MeV threshold for its occurrence.

Neutrons have many ways to interact with matter characteristic of different cross sections, such as elastic scattering (σ_i^{ES}), inelastic scattering (σ_i^{IES}), radiative capture (σ_i^{RC}), and fission (σ_i^{F}). Thus, the total cross section for neutrons for atom *i*, $\sigma_{i,t}^{Neutron}$, can be written as

$$\sigma_{it}^{Neutron} = \sigma_i^{ES} + \sigma_i^{IES} + \sigma_i^{RC} + \sigma_i^{F}.$$
(4)

For heterogeneous materials, the integral form should be used:

$$I_t = I_0 e^{-f_l \mu(x)dx}$$
⁽⁵⁾

where *l* denotes the transmission path and *x* is a 3D position vector. The transmission of the beam through the sample depends upon the map of the attenuation coefficient $\mu(x)$ and transmission path.

Table 1 lists the attenuation coefficient, μ , for several common components found in shales for X-rays and neutrons at relevant conditions. Attenuation coefficients for neutrons and X-rays are calculated by using the Neutron Activation and Scattering Calculator provided by NCNR (Kienzle, 2018) and X-Ray Form Factor, Attenuation, and Scattering Tables provided by NIST Physical Measurement Laboratory (Seltzer, 2018), respectively. Clearly, hydrogen-containing materials have large attenuation coefficients for neutrons compared with hydrogen-free elements.

 Table 1—The Attenuation Coefficients of X-Rays and Neutrons for

 Common Components in Shales

Component	μ for X-rays 90 keV ^a (1/cm)	μ for neutrons 1.8 Å ^b (1/cm)
Kerogen ^c (C ₁₀₀ H ₁₆₁ N _{1.85} S _{0.7} O _{9.2})	0.22	6.97
Quartz (SiO ₂)	0.48	0.29
Pyrite (FeS ₂)	1.66	0.44
Siderite (FeCO ₃)	1.20	0.67
Calcite (CaCO ₃)	0.58	0.35
Dolomite (CaMg(CO ₃) ₂)	0.55	0.41
Fluorapatite Ca5(PO4)3F	0.71	0.31
Anhydrite (CaSO ₄)	0.64	0.29
Kaolinite (Al ₂ Si ₂ O ₅ (OH) ₄)	0.47	2.32
Chlorite (Mg ₅ Al ₂ Si ₃ O ₁₀ (OH) ₈)	0.44	2.07
Daphnite (Fe₅Al₂Si₃O₁₀(OH)ଃ)	0.95	2.30
H ₂ O	0.18	5.65
C ₆ H ₁₄	0.12	5.39

^a90 keV is the peak X-ray energy for the spectrum generated by the X-ray tube in this measurement. The attenuation coefficient for X-rays is obtained from the X-Ray Form Factor, Attenuation, and Scattering Tables (Seltzer, 2018).

^bThe neutrons used in this study are thermal neutrons with a Maxwell-Boltzmann distribution centered around 1.8 Å wavelength. The attenuation coefficient for neutrons is obtained from the Neutron Activation and Scattering Calculator (Kienzle, 2018).

Formula obtained from literature (Siskin et al., 1995).

MATERIALS

Two samples were used in this study. The first sample, provided by the Bureau of Economic Geology (BEG), is Barnett Shale from the T.P. Sims #2 well in the Fort Worth Basin. The characterization of this sample can be found in previous study (Louck et al., 2009; Zhang et al., 2014). It has a vitrinite reflectance of $R_o = 1.61\%$ and total organic carbon (TOC) = 3.64%. The second sample is an organic-rich carbonate source rock from the Middle East, here named Middle East shale. The samples are 2.5 cm in diameter and approximately 2 cm in length. The Barnett Shale sample was plugged parallel to the bedding and the Middle East sample was plugged perpendicular to the bedding. Both samples were imaged without any further sample processing.

EXPERIMENTAL SETUP

Experiments were conducted at NCNR on the BT2 neutron imaging facility. This instrument offers a high thermal-neutron flux for radiography and tomography. A 90-keV microfocus X-ray generator is located perpendicular to the neutron beam at the sample location to allow simultaneous neutron and X-ray tomography. Full details of the instrument can be found in the previous work of some of the authors (LaManna et al., 2017). Detector resolution was set to 30 μ m for both neutron and X-ray detectors in order to keep the sample in the field-of-view at all projection angles. Each scan took approximately 18 hours to complete. Volumes were reconstructed from the projection images using the commercial CT software package Octopus and visualized with the open-source Drishti Volume Exploration Tool (Limaye, 2012).

Certain trade names and company products are mentioned in the text or identified in an illustration in order to adequately specify the experimental procedure and equipment used. In no case does such identification imply recommendation or endorsement by the National Institute of Standards and Technology, nor does it imply that the products are necessarily the best available for the purpose.

RESULTS

The two shale samples were studied using the simultaneous X-ray and neutron tomography to extract the 3D structure and spatial arrangement of their components, such as minerals, organic matter, and fractures in the rocks.

Barnett Shale Sample

Figure 1 shows the simultaneous X-ray and neutron CT reconstruction slices for the Barnett Shale sample. The brighter parts of the image indicate regions with larger attenuation, while the darker parts correspond to regions

with less attenuation. For neutrons, the hydrogen-rich areas are brighter because hydrogen has larger cross section and attenuates neutron intensity more significantly. For X-rays, the cross section of an element is proportional to its atomic number Z and the brighter areas indicate the regions containing more high-Z elements such as heavy minerals.



Fig. 1—Upper slices of (a) X-ray, and (b) neutron reconstructed volumes of the Barnett Shale sample. The short-dashed yellow ovals and solid green ovals outline the mineral components of pyrite and apatite, respectively. Solid red rectangles outline the organic-rich regions likely to be kerogens, and the long-dashed magenta rectangles outline the open fractures.

Layered patterns are apparent in the upper part of the sample in both X-ray and neutron tomography slices (Fig. 1). The bright layers in the X-ray slice indicate that these layers have high atomic numbers and, therefore, are composed of minerals (see the layers outlined by the solid green ovals). These bright layers in the X-ray slice correspond to the dark layers in the neutron slice. The stripes are fluorapatite (see Table 1 for attenuation coefficients and Fig. A1 in the Appendix for XRD result).

There is also a bright pocket connected with a bright sharp line on the upper right portion of the slices for both the neutron and X-ray images (outlined by short-dashed yellow ovals). This indicates that this pocket region contains both elements with high atomic number and elements that can strongly attenuate neutron intensity. Separate SEM measurements were conducted on the location of the bright pocket shown in both X-ray and neutron images. SEM results indicate that this pocket is composed of both pyrite, which attenuates X-rays significantly, and organic matter, whose hydrogen content greatly attenuates neutrons. An SEM image of a surface area at the pyrite region is shown in appendix Fig. A2. The mixture of pyrite and organic matter can also be confirmed by Fig. 1, which shows that the pocket is not homogeneously bright and has a dark shadow in both X-ray and neutron images.

The neutron image in Fig. 1b shows sporadic bright stripes that correspond to the dark stripes in the X-ray image in Fig. 1a, as outlined by the solid red rectangles. The stripes are regions with concentrated hydrogens, which attenuate neutrons but are transparent to X-rays. The hydrogen-rich areas are quite likely to be organic kerogen.

There are also regions that look dark in both neutron and X-ray images in Fig. 1 (outlined by the long-dashed magenta rectangles). Because this feature is transparent to, and therefore has no interaction with both neutrons and X-rays, it is reasonable to identify the lines as fractures. If X-rays are the only imaging source (Fig. 1a), it is hard to distinguish between the organic matter (solid red rectangles) and the fractures (long-dashed magenta rectangles). With the aid of neutron imaging, the two different components can be easily recognized.

The lower part of the neutron slice is homogenously brighter than the upper part, indicating that the hydrogen content is higher in the lower part rather than in the upper part of the sample. This is consistent with the fact that the upper part is composed of the layered and pocket inorganicmatter-containing minerals.

Figures 2a and 2b are the 3D X-ray and neutron volume renderings of the T.P. Sims #2 well sample, respectively. The colored regions show the highly attenuated areas. The rock sample can be divided into two parts: one with the mineralrich region containing layers of fluorapatite and a pocket connecting to a healed fracture filled with pyrite and organic matter; the other with a relatively homogeneous hydrogenrich region similar to that seen in the slices shown in Fig. 1. By setting a high grayscale-display threshold, the highattenuation regions, i.e., the yellow and red areas, are selected from the X-ray volume and neutron volume effectively identifying the densest inorganic or mineralized (Fig. 2c) and main organic-rich regions (Fig. 2d), respectively. Figure 2c shows the pocket and the healed fracture that cuts through the layered structure at an oblique angle, both filled with pyrite (and organic matter, as indicated in Fig. 2d). Figure 2d highlights several areas with highly concentrated hydrogen content that corresponds to voids in the X-ray image in

addition to the overall dispersed organic content in the one half of the sample. Figures 2b and 2d suggest that the organic region can be further divided into denser-organic region, and less-dense-organic region. The 3D structure of organic matter allows us to map the connection of kerogen 'globs' and fractures and, thus, suggests a possible flow path for light hydrocarbons at the core scale in this shale. Therefore, neutron tomography provides unique and significant information for modeling flow in shales at the core scale.



Fig. 2—3D volume renderings of the Barnett Shale sample for (a) X-ray, and (b) neutron tomography. (c) Distribution of high-Z components, i.e., the minerals, showing a pyrite pocket and a healed fracture. (d) Distribution of hydrogen-rich components, likely to be kerogens, are identified by isolating the appropriate peak in the grayscale histograms.

It should be noted that the concentric rings in the neutron imaging, shown in Figs. 1b and 2b are CT artifacts called ring noise, which was generated from dead pixels in the detector.

Middle East Shale Sample

Figure 3 shows the simultaneous X-ray and neutron tomography slices for the Middle East shale. As described above, the regions outlined by solid green ovals are bright in the X-ray image but dark in the neutron image and, therefore, are occupied by minerals. There is a region outlined by short-dashed yellow oval that is bright in both X-ray and neutron images and is also mineral-rich. Based on the same reasons described for Barnett Shale sample, these minerals may be anhydrite and pyrite for areas outlined by solid green ovals and short-dashed yellow oval, respectively. The regions outlined by solid red rectangles are dark on the X-ray image but bright in neutron image and likely are organic matter. There is a stripe that is dark in both neutron and X-ray images and is identified as fracture (outlined by long-dashed magenta rectangles).



Fig. 3—Central slices of (a) X-ray, and (b) neutron reconstructed volumes of the Middle East shale sample. The short-dashed yellow ovals and solid green ovals outline the mineral components of pyrite and probable anhydrite, respectively. Solid red rectangles outline the organic-rich regions likely to be kerogens; the long-dashed magenta rectangles outline the fractures.

Figures 4a and 4b are the 3D reconstructed images of the Middle East shale sample for X-ray and neutron modes, respectively. The colored regions are highly attenuated areas. By only displaying the high-attenuation regions, the distributions of inorganic and organic regions are plotted in Figs. 4c and 4d, respectively. Surprisingly, the organic matter has a clearly layered structure with two layers at the top and in the center of the sample (Fig. 4d). This is very different from the structure of the organic matter found in the Barnett Shale sample. Moreover, the inorganic material is also clearly layered and forms a continuous inorganic-rich layered shale (i.e., mudrock) (Fig. 4c).



Fig. 4—3D volume renderings of the Middle East shale sample for (a) X-ray, and (b) neutron tomography. (c) Distribution of high-Z components, i.e., the minerals, and (d) distribution of hydrogenrich components, likely to be kerogens, are identified from the highly attenuated (colorized) regions in 3D X-ray and neutron tomography reconstructions, respectively.

The layered structures of organic matter have two important implications: first, it may give rise to a fast path for light-hydrocarbon movement. Second, it produces strong contrasts in mechanical properties, which in turn may be weak surfaces along which the sample may break. The layer at the top of the sample, for example, is a broken face the occurred during plugging.

DISCUSSION

The distributions of different components within shales are crucial for storage and transport of hydrocarbons in the shales. The detailed structural description on the core scale of two shale samples demonstrates that simultaneous X-ray and neutron tomography is a powerful tool to quantify the distributions of organic and inorganic matter. With more information provided by the combined X-ray and neutron tomography, different minerals, such as pyrite, calcite, and anhydrite, can be distinguished. Moreover, since the hydrogen-rich organic matter and fractures are both very transparent to X-rays, it is challenging for the commonly used X-ray CT alone to unambiguously determine the organicmatter distributions at the core scale. In contrast, neutrons are very sensitive to hydrogens and are useful to characterize the organic-rich shales. By combining X-ray and neutron tomography, more comprehensive information of material composition and distributions in shales is obtained.

It is worth mentioning that the component distributions found in this study are at the core scale, rather than the nanometer to micrometer pore scale. Current simulations and modeling of hydrocarbons are mostly based on the rock structure at the pore scale constructed from SEM images (Shabro et al., 2013; Chen et al., 2015; Saraji and Piri, 2015; Jiang et al., 2017). There is about a 10⁹ order of magnitude upscaling in length scale from the pore to core scale. It will be a significant impact to study the influence of the distributions of different components within the rocks by simulating and modeling the hydrocarbon flow at the core scale.

Since the wettability and surface properties are very different for the minerals and organic matter (Akbarabadi et al., 2017), which can be determined by scattering technique (Chiang et al., 2018), the structure and arrangement in space of these components can directly affect the storage and transport of the hydrocarbons within the shale rocks (Chiang et al., 2016a; 2016b). The two shale samples from two different formations exhibit very different organic matter structures. The laminar organic matter (most likely to be kerogens) in the Middle East shale sample can dramatically change the current views of hydrocarbon flow and the mechanical properties in the shales.

CONCLUSIONS

Simultaneous X-ray and neutron tomography is necessary to more accurately characterize the distributions of different components in shale, such as minerals, organic matter, and fractures in the heterogeneous materials. In this study, the 3D characterizations of minerals and organic matter at the core scale are successfully demonstrated for the first time in shale samples from different locations. These results can be used in future simulation and modeling of hydrocarbon flow at the core scale. In particular, the layered organic structure found in one of the shale samples can significantly change the current explanations of hydrocarbon flow in these shales.

ACKNOWLEDGEMENTS

The authors would like to thank Aramco for allowing the publication of this work. The technical work has benefited from discussions with colleagues at Aramco Research Center-Houston: Younane Abousleiman, Hui-Hai Liu. Y.L. acknowledges the partial support of cooperative agreements 70NANB12H239 and 70NANB10H256 from NIST, U.S. Department of Commerce.

NOMENCLATURE

Acronyms

- BEG = Bureau of Economic Geology
- CT = computed tomography
- FIB-SEM = focused ion beam-scanning electron microscopy GIP = gas-in-place

158

- NCNR = NIST Center for Neutron Research
 - NIST = National Institute of Standards and Technology
 - XRD = X-ray diffraction

Symbols

- I_0 = intensity of incident X-ray or neutron beam
- I_{i} = transmitted intensity of X-rays or neutrons
- N = number density of type *i* atom
- \dot{d} = sample thickness
- l = transmission path of X-rays or neutrons
- x = 3D position vector
- Z =atomic number
- μ = attenuation coefficient
- $\sigma_{t,i}$ = total attenuation cross section of type *i* atom for X-rays or neutrons

REFERENCES

- Akbarabadi, M., Saraji, S., Piri, M., Georgi, D., and Delshad, M., 2017, Nano-scale Experimental Investigation of In-situ Wettability and Spontaneous Imbibition in Ultra-Tight Reservoir Rocks, *Advances in Water Resources*, **107**, 160–179. DOI: 10.1016/j.advwatres.2017.06.004.
- Aloulou, F., and Zaretskaya, V., 2016, Shale Gas Production Drives World Natural Gas Production Growth - Today in Energy, US Energy Information Administration (EIA). https://www.eia.gov/todayinenergy/detail.php?id=27512. Accessed March 1, 2018.
- Ambrose, R.J., Hartman, R.C., Diaz Campos, M., Akkutlu, I.Y., and Sondergeld, C., 2013, New Pore-scale Considerations for Shale Gas in Place Calculations, Paper SPE-131772, presented at the SPE Unconventional Gas Conference, Pittsburgh, Pennsylvania, USA, 23–25 February. DOI: 10.2118/131772-MS.
- Banhart, J., Borbély, A., Dzieciol, K., Garcia-Moreno, F., Manke, I., Kardjilov, N., Kaysser-Pyzalla, A.R., Strobl, M., and Treimer, W., 2010, X-Ray and Neutron Imaging— Complementary Techniques for Materials Science and Engineering, *International Journal of Materials Research*, **101**(9), 1069–1079. DOI: 10.3139/146.110382.
- Chen, L., Zhang, L., Kang, Q., Viswanathan, H.S., Yao, J., and Tao, W., 2015, Nanoscale Simulation of Shale Transport Properties Using the Lattice Boltzmann Method: Permeability and Diffusivity, *Scientific Reports*, 5, Article 8089. DOI: 10.1038/srep08089.
- Chiang, W.-S., Fratini, E., Baglioni, P., Chen, J.-H., and Liu, Y., 2016a, Pore Size Effect on Methane Adsorption in Mesoporous Silica Materials Studied by Small-Angle Neutron Scattering, *Langmuir*, **32**(35), 8849–8857. DOI: 10.1021/acs.langmuir.6b02291.
- Chiang, W.-S., Fratini, E., Baglioni, P., Georgi, D., Chen, J., and Liu, Y., 2016b, Methane Adsorption in Model Mesoporous Material, SBA-15, Studied by Small-Angle Neutron Scattering. *The Journal of Physical Chemistry C*, 120(8), 4354–4363. DOI: 10.1021/acs.jpcc.5b10688.

Chiang, W.-S., Georgi, D., Yildirim, T., Chen, J., and Liu, Y.,

2018, A Non-Invasive Method to Directly Quantify Surface Heterogeneity of Porous Materials, *Nature Communications*, **9**, Article No. 784. DOI: 10.1038/s41467-018-03151-w.

- Jiang, W., Lin, M., Yi, Z., Li, H., and Wu, S., 2017, Parameter Determination Using 3D FIB-SEM Images for Development of Effective Model of Shale Gas Flow in Nanoscale Pore Clusters, *Transport in Porous Media*, **117**(1), 5–25. DOI: 10.1007/s11242-016-0817-5.
- Kienzle, P., 2018, Neutron Activation and Scattering Calculator, NIST Center for Neutron Research. https://www.ncnr.nist. gov/resources/activation/. Accessed March 1, 2018.
- LaManna, J.M., Hussey, D.S., Baltic, E., and Jacobson, D.L., 2017, Neutron and X-Ray Tomography (NeXT) System for Simultaneous, Dual Modality Tomography, *Review of Scientific Instruments*, 88, 113702. DOI: 10.1063/1.4989642.
- Limaye, A., 2012, Drishti: A Volume Exploration and Presentation Tool, *Developments in X-Ray Tomography VIII*, International Society for Optics and Photonics, SPIE Proceedings 8506. DOI: 10.1117/12.935640.
- Loucks, R.G., Reed, R.M., Ruppel, S.C., and Jarvie, D.M., 2009, Morphology, Genesis, and Distribution of Nanometer-Scale Pores in Siliceous Mudstones of the Mississippian Barnett Shale, *Journal of Sedimentary Research*, **79**(12), 848–861. DOI: 10.2110/jsr.2009.092.
- Manke, I., Banhart, J., Haibel, A., Rack, A., Zabler, S., Kardjilov, N., Hilger, A., Meizer, A., and Riesemeier, H., 2007, In Situ Investigation of the Discharge of Alkaline Zn–MnO₂ Batteries With Synchrotron X-Ray and Neutron Tomographies, *Applied Physics Letters*, **90**(21), Paper 214102. DOI: 10.1063/1.2742283.
- Manke, I., Hartnig, C., Grünerbel, M., Lehnert, W., Kardjilov, N., Haibel, A., Hilger, A., Banhart, and Riesemeier, H., 2007, Investigation of water evolution and transport in fuel cells with high resolution synchrotron x-ray radiography, *Applied Physics Letters*, **90**(17), 174105.
- Mannes, D., Benoît, C., Heinzelmann, D., and Lehmann, E., 2014, Beyond the Visible: Combined Neutron and X-ray Imaging of an Altar Stone from the Former Augustinian Church in Fribourg, Switzerland, *Archaeometry*, 56(5), 717–727. DOI: 10.1111/arcm.12061.
- Mannes, D., Schmid, F., Frey, J., Schmidt-Ott, K., and Lehmann, E., 2015, Combined Neutron and X-Ray Imaging for Non-Invasive Investigations of Cultural Heritage Objects, *Physics Procedia*, **69**, 653–660. DOI: 10.1016/jphpro. 2015.07.092.
- Saraji, S., and Piri, M., 2015, The Representative Sample Size in Shale Oil Rocks and Nano-Scale Characterization of Transport Properties, *International Journal of Coal Geology*, **146**, 42–54. DOI: 10.1016/j.coal.2015.4.005.
- Seltzer, S.M., 2018. NIST X-Ray Form Factor, Attenuation and Scattering Tables Form Page, NIST, http://physics.nist.gov/ PhysRefData/FFast/html/ form.html. Accessed March 1, 2018.
- Shabro, V., Kelly, S., Torres-Verdín, C., and Sepehrnoori, K., 2013, Pore-Scale Modeling of Electrical Resistivity and Permeability in FIB-SEM Images of Hydrocarbon-Bearing Shale, Paper AA, *Transactions SPWLA 54th Annual Logging Symposium*, New

Orleans, Louisiana, USA, 22–26 June.

- Siskin, M., Scouten, C.G., Rose, K.D., Aczel, T., Colgrove, S.G., and Pabst, R.E., 1995, Detailed Structural Characterization of the Organic Material in Rundle Ramsay Crossing and Green River Oil Shales, *in* Snape, C.E., Editor, *Composition, Geochemistry and Conversion of Oil Shales,* Springer, 143–158. ISBN: 978-0-7923-3343-2. DOI: 10.1007/978-94-011-0317-6.
- Zhang, T., Yang, R., Milliken, K. L., Ruppel, S.C., Pottorf, R. J., and Sun, X., 2014, Chemical and Isotopic Composition of Gases Released by Crush Methods From Organic Rich Mudrocks, *Organic Geochemistry*, 73, 16–28. DOI: 10.1016/j.orggeochem. 2014.05.003.

APPENDIX



Fig. A1—XRD of the Barnett shale sample. The blue spectrum is the diffractogram from a Bruker D8 Advance Eco (at 40kV 25 mA with a 0.6-mm divergent slit; two 2.5° soller slits; and a LynxEye Detector with all 192 channels used). The red trace is the simulated pattern using Bruker's Topas with the reported minerals; the black trace shows the simulated pattern for the mineral fluorapatite. The amounts of the minerals are validated using another Rietveld software Autoquan/BGMN.



Fig. A2—SEM image of a surface area in the pyrite region in Fig. 2 (colored yellow in Fig. 2a) showing significant amount of organic matter. The organic matter has a relatively large attenuation coefficient to neutrons and thus shows as bright in the neutron images. Image size:

 $130 \times 97 \ \mu m.$

ABOUT THE AUTHORS



Wei-Shan Chiang joined the University of Delaware and National Institute of Standards and Technology as Postdoctoral Researcher in 2014. She is collaborating with researchers in Aramco Research Center-Houston to study the storage and transport of hydrocarbons in shales, and the structure of shale rocks. Before that, she was a PhD student

at the Massachusetts Institute of Technology to improve the understanding of structure-property relationship of concretes and cements. She was also a masters student in National Tsing Hua University working on the phase behaviors of polymer/surfactant complexes.



Jacob LaManna is a staff scientist at the National Institute of Standards and Technology where he joined as a postdoctoral fellow in 2014. He is primarily responsible for the operation and development of the Neutron and X-ray Tomography (NeXT) system. His research

interests are in imaging of multiphase flow in porous materials and their applications in geological systems, energy storage and conversion, and structural materials. He received his PhD in mechanical engineering from the University of Tennessee Knoxville.



Daniel S. Hussey is a research scientist at the National Institute of Standards and Technology. His primary research focus is employing neutron optics to create quantitative multimodal and multiscale imaging methods with applications in two phase flow in geological systems. He holds a MSc and a PhD in Physics

from Indiana University.

PETROPHYSICS



Yun Liu is currently an instrument scientist at the National Institute of Standards and Technology Center for Neutron Research (NCNR), and an affiliated associate professor at the Chemical & Biomolecular Engineering Department and Physics & Astronomy Department at the University of Delaware. He received his PD in Nuclear Science

Engineering at MIT in 2005. He joined NCNR first as a postdoc and later became a member of staff. His research interests cover a wide range of scientific problems, such as porous materials, gas adsorption/desorption, energy materials, biophysics, and colloidal science.



Jin-Hong Chen joined Aramco Research Center-Houston in 2013 and is working on developing technologies for improved evaluation and production in unconventional source rock reservoirs. Previously, he was a petrophysicist developing NMR formation evaluation technology at Baker Hughes. He was a research fellow at Harvard University

and a visiting scientist at MIT. He has a PhD in Physics from Chinese Academy of Science.