

Environmental Science Nano

Increasing evidence indicates low bioaccumulation of carbon nanotubes

Journal:	Environmental Science: Nano
Manuscript ID	EN-PER-09-2016-000389.R1
Article Type:	Perspective
Date Submitted by the Author:	20-Dec-2016
Complete List of Authors:	Bjorkland, Rhema; US Environmental Protection Agency Office of Pollution Prevention and Toxics, Risk Assessment Division; American Association for the Advancement of Science, Science and Technology Fellowship Program Tobias, David; US Environmental Protection Agency Office of Pollution Prevention and Toxics Petersen, Elijah; NIST

SCHOLARONE[™] Manuscripts

The safe and responsible development of nanoenabled materials requires an assessment of the environmental, health and safety implication of engineered nanomaterials and other emerging technologies. Understanding the environmental fate and bioaccumulation potential of carbon nanotubes is key to advancing the risk evaluation and management process. Our work provides a comprehensive review of this topic and summarizes the current knowledge base to provide an evidence-driven assessment as to bioaccumulation potential and trophic transfer risk across a wide variety of taxa.

Increasing evidence indicates low bioaccumulation of

carbon nanotubes

Rhema Bjorkland^{1‡*}, David Tobias² and Elijah J. Petersen³

 ¹AAAS Science & Technology Policy Fellow, Risk Assessment Division, US EPA Office of Pollution Prevention and Toxics
 [†] current address: National Nanotechnology Coordination Office, Arlington, VA, United States
 ² Risk Assessment Division, US EPA Office of Pollution Prevention and Toxics
 ³ National Institute of Standards and Technology, Biosystems and Biomaterials Division, Material Measurement Laboratory, Gaithersburg, MD, United States

*Corresponding author: rhemaker@hotmail.com

Abstract

As the production of carbon nanotubes (CNTs) expands, so might the potential for release into the environment. The possibility of bioaccumulation and toxicological effects has prompted research on their fate and potential ecological effects. For many organic chemicals, bioaccumulation properties are associated with lipid-water partitioning properties. However, predictions based on phase partitioning provide a poor fit for nanomaterials. In the absence of data on the bioaccumulation and other properties of CNTs, the Office of Pollution Prevention and Toxics (OPPT) within the US Environmental Protection Agency (EPA) subjects new pre-manufacture submissions for all nanomaterials to a higherlevel review. We review the literature on CNT bioaccumulation by plants, invertebrates and nonmammalian vertebrates, summarizing 40 studies to improve the assessment of the potential for bioaccumulation. Because the properties and environmental fate of CNTs may be affected by type (single versus multiwall), functionalization, and dosing technique, the bioaccumulation studies were reviewed with respect to these factors. Absorption into tissues and elimination behaviors across species were also investigated. All of the invertebrate and non-mammalian vertebrate studies showed little to no absorption of the material from the gut tract to other tissues. These findings combined with the lack of biomagnification in the CNT trophic transfer studies conducted to date suggest that the overall risk of trophic transfer is low. Based on the available data, in particular the low levels of absorption of CNTs across epithelial surfaces, CNTs generally appear to form a class that should be designated as a low concern for bioaccumulation.

1 Introduction

Carbon nanotubes (CNTs) and other carbon-based nanomaterials are major building blocks of nanotechnology³. CNTs have been incorporated into diverse products, ranging from lightweight data cables, rechargeable batteries, automotive parts, and sporting goods to boat hulls and water filters⁴. They currently have the highest production volumes among carbonaceous engineered nanomaterials (ENMs) worldwide ⁷. As production and use of CNTs grow, so does the potential for their release to the environment and for the exposure of ecological receptors^{8,9}. The prospect of nanomaterial release into the environment and possible bioaccumulation and toxicological effects has prompted research on the fate, transport and effects of these materials on biota. However, the novel or enhanced properties associated with materials that have nanoscale dimensions between 1 nm and 100 nm in at least one dimension, ¹², also creates unique challenges in assessing their likely impact on human health and the environment. A key component for risk assessment of traditional chemicals includes an evaluation of their persistence, potential for bioaccumulation and potential to cause toxic effects. As governments began articulating concerns about these three properties of chemicals, regulators began placing persistence, bioaccumulation and toxicity (PBT) characteristics into a common regulatory scheme in the identification of chemical hazards e.g., Japan's Chemical Substances Control Law, ¹³. Chemicals designated as PBT are priority substances for regulators and environmental managers and may be subject to controls (e.g., limitations on release and toxicity testing). Bioaccumulation, the second pillar in the PBT framework, occurs when the chemical

concentration in an organism exceeds that in its environmental matrix ^{1, 14}. The propensity for a chemical
 to accumulate in tissues could increase the probability of transfer up the food chain from prey to
 predators, thus creating increasingly larger exposures for upper-level predators, including human beings
 ¹⁵. The potential for bioaccumulation, the B in PBT, represents an assessment of the accumulation of a

Environmental Science: Nano

2		
3 4	25	chemical from the environment to an organism's tissues ¹ . If a chemical has a low persistence in the
5 6 7	26	environment, this would usually end concern regarding its PBT properties. However, given the
7 8 9	27	persistence of CNTs in the environment as will be discussed later, this raises the importance of
10 11 12	28	determining their potential for bioaccumulation within the PBT framework.
13 14	29	This review will focus on non-mammalian organisms and ignores inhalation exposures for
15 16	30	terrestrial organisms. Inhalation exposures and the buildup of a chemical in the lungs are important for
17 18	31	determining potential toxic effects but accumulation of a chemical in the lungs alone is not an indication
20 21	32	of high bioaccumulation potential. Furthermore, biomagnification is an important indicator of
22 23	33	bioaccumulation potential and inhalation exposures are not typically connected to the ability of a
24 25	34	chemical to biomagnify in a food web. The potential for bioaccumulation (or bioconcentration, see Box 1
26 27 28	35	for definitions) for many organic chemicals is correlated with phase-distribution properties ¹⁶ . Chemicals
29 30	36	will redistribute (equilibrate) into the most energetically favorable phase; for hydrophobic organic
31 32	37	chemicals this is typically partitioning into another organic phase such as lipid, proteins, or
33 34 25	38	polysaccharides ¹⁷ . In contrast, compounds that are hydrophilic tend to have a low potential to
36 37	39	bioaccumulate or bioconcentrate and do not readily partition into an organism's tissues ¹⁸ .
38 39 40	40	In the 1970s, the concept of bioconcentration as a phenomenon of equilibrium partitioning 19
41 42	41	led to modeling efforts that linked bioconcentration measurements of a chemical to measurements of
43 44	42	its partitioning behaviors (the ratio of contaminant concentrations in two phases at equilibrium). In
45 46 47	43	particular, the octanol-water partitioning coefficient (K_{ow}) has been used to categorize and predict the
48 49	44	bioconcentration factor of organic chemicals as it frequently reflects a chemical's affinity to partition to
50 51	45	lipids within an organism ^{20, 21} . In general, the hierarchy of evidence for the potential for
52 53	46	bioaccumulation or bioconcentration begins with a field measured trophic magnification factor (TMF),
54 55 56	47	followed by field, then laboratory-based biomagnification factors (BMFs), bioaccumulation factors
57 58	48	(BAFs), and then laboratory-measured bioconcentration factors (BCF). The lowest tier is a measured or
59 60		4

Box 1. Definitions

Bioaccumulation - Bioaccumulation is the process by which a chemical substance is absorbed by an organism from all routes of exposures as occurs in the natural environment (i.e., dietary and ambient environment sources) and achieves a level that exceeds those in the exposed sources. Bioaccumulation is distinct from bioconcentration because chemical exposure is in the diet and therefore potential biomagnification is included ^{1, 2}.

Bioaccumulation Factor - Ratio of the steady state chemical concentrations in an aquatic water-respiring organism (C_B , g chemical/kg ww) and the water (C_w , g chemical/L) determined from field data in which sampled organisms are exposed to a chemical in the water and in their diet. Thus BAF = C_B / C_W^{-1}

Bioconcentration- The process by which a chemical substance is absorbed by an organism from the ambient environment only through its respiratory and dermal surfaces, i.e., exposure in the diet is not included (Arnot and Gobas, 2006).

Bioconcentration Factor (BCF) – The ratio of the steady state chemical concentrations in an aquatic waterrespiring organism (CB, g chemical/kg ww) and the water (C_w , g chemical/L) determined in a controlled laboratory experiment in which the test organisms are exposed to a chemical in the water (but not in the diet). Thus BCF = C_B / C_W^{-1} .

Biomagnification - Bioaccumulation of a chemical through an ecological food chain by transfer of residues from the diet into body tissues. The tissue concentration increases at each trophic level in the food web when there is efficient uptake and slow elimination ^{5, 6}.

Biomagnification factor (BMF) – The ratio of the steady state chemical concentrations in a water- or airrespiring organism (CB, g chemical/kg ww) and in the diet of the organism (CD, g chemical/kg ww). BMF is determined either in a controlled laboratory experiment in which the test organisms are exposed to chemical in the diet (but not the water or air) or from field data in which sampled organisms are exposed to chemical in air, water, and diet ¹.

Biodistribution - Distribution of a chemical within an organism ^{9, 10}.

Uptake -that part of the bioaccumulation/bioconcentration process(es) involving the movement of a chemical from the external environment into an organism, either through direct exposure to a contaminated medium and/or by consumption of food containing the chemical ^{9, 11}.

49 estimated octanol-water partition coefficient (K_{ow}). Bioconcentration measurements for dissolved

50 chemicals may have included the total organism mass including the contents of the gut, and for highly

51 bioaccumulated chemicals, distribution into systemic circulation and accumulation in specific tissues

52 were assumed. Nanomaterials that do not penetrate the epithelial surfaces, such as the gut tract,

require removal of the gut or inclusion of a depuration period to distinguish between nanomaterials that

Environmental Science: Nano

have been ingested or uptaken and those that have been absorbed across epithelial tissues and entered into systemic circulation in the organisms; in the nanomaterial bioaccumulation literature, it is common to use the term uptake to refer to nanomaterials that have entered the organism and remain in the gut tract while this term is more typically used for other chemicals to reflect those that have passed through epithelial surfaces and into systemic distribution in the organism. Even if a bioaccumulation study only exposes the organism to nanomaterials suspended in water (i.e. BCF type study), filter feeders like Daphnia may show accumulation of the material in their intestines because they ingest them. Typical lipid normalization approaches may also not be appropriate as a given nanomaterial will not necessarily associate with lipids.

The earliest use of the PBT concept was by Japan and other jurisdictions then adopted this usage ^{22, 23}. In the US, the association of persistence, bioaccumulation and toxicity was set out in the Resources Conservation and Recovery Act (RCRA) of 1976 (42 U.S.C. §6921). By the late 1980s, the OPPT had established working categories for chemicals, including one for PBT compounds ²⁴. In the 1990s, under the RCRA's Waste Minimization Action Plan, EPA developed a scoring system for human and ecological risk potential based on PBT characteristics²⁵. That system placed substances in different categories corresponding to a low, medium and high value for each assessment factor (P, B, and T) giving the substances a ranking from 1 to 3 (where 3 indicates high concern). As mentioned in the previous paragraph, a good correlation between K_{ow} and BCF has been found for many nonionic organic molecules ²⁶. Models based on this relationship have been built into EPA's Estimation Program Interface (EPI) Suite[™] software, a widely used tool for predicting physico-chemical properties and environmental fate of chemicals in the absence of measured data. Such models provide the basis for many of the initial assessments by the New Chemical Review Program (NCRP) within OPPT of the potential for bioaccumulation or bioconcentration for chemicals where test data are not available.

OPPT later refined its approach to include a formal consideration of PBT under the Toxic Release Inventory (TRI) program established under the Emergency Planning and Community Right-to-Know Act (EPCRA, 42 U.S.C. §11001 et seq. (1986). This approach was adopted by OPPT as part of its management of new chemicals under the Toxic Substances Control Act ²⁴. The 1999 Federal notice also outlined a tiered test strategy OPPT believed necessary for a PBT chemical evaluation. This information is provided here to provide context for the process that would typically be used for chemical submissions. The PBT policy takes into account factors such as magnitude of releases, results of physicochemical and potential ecotoxicological testing, and structure-activity relationship (SAR) prediction ²⁴. Evaluation factors for the potential for bioaccumulation or bioconcentration include experimental determination of Log K_{ow} (tier 1) and experimental determination of a fish BCF for tier 2. The European Union (EU) regulates chemicals under REACH (Registration, Evaluation, Authorisation and Restriction of Chemicals) and employs similar evaluation methods for standard industrial chemicals.

While the use of partitioning models to estimate the potential for bioaccumulation or bioconcentration has been available for traditional organic chemicals for decades, this framework is not considered valid for determining how CNTs or other nanomaterials would behave in food webs given the substantial differences in the partitioning behaviors between nanomaterials and organic chemicals. Unlike dissolved organic chemicals, nanomaterial dispersions are colloidal suspensions, requiring energy input to become suspended throughout another phase ²⁷. Therefore, ratios of nanoparticle concentrations in two phases violate the fundamental description of an equilibrium partitioning coefficient. Despite their use in the regulatory framework for organic chemicals, current test guidelines for estimating bioaccumulation (e.g., BCF) using partitioning coefficients are not appropriate to measure the bioconcentration of chemicals that do not reach equilibrium among phases such as nanomaterials (e.g., the organism tissue and water, see Handy et al. ^{28, 29}).

Page 9 of 42

1

Environmental Science: Nano

2
3
4
5
5
6
7
8
õ
9
10
11
12
13
13
14
15
16
17
10
10
19
20
21
22
<u> </u>
23
24
25
26
20
27
28
29
30
30
31
32
33
31
04
35
36
37
38
20
39
40
41
42
10
40
44
45
46
40
41
48
49
50
51
51
52
53
54
55
55
56
57
58
50
0.0
nu

100 Nanomaterials also behave differently than traditional low solubility organic chemicals that are 101 challenging to test in traditional assays for measuring the potential for bioaccumulation. Exposing a test 102 organism to a steady dose of a low solubility organic chemicals can be difficult due to challenges in 103 solubilizing the chemical in media and measuring the concentration of the test substance during the 104 study. However, for the most part, these low solubility organic molecules are still expected to partition 105 to lipids. In contrast, it is not clear that nanomaterials, such as CNTs, will partition to lipids or that 106 equilibrium behavior will be responsible for determining their fate in an organism. This arises from the 107 instability of CNTs in water and the slow and not well understood mechanism for CNTs passing through 108 epithelial surfaces which often leads to concentrations in the organism tissues outside of the gut tract being below the instrument detection limits. 109

110 There is evidence that CNTs will persist in the environment. Hydrolysis is not expected to be a 111 significant environmental degradation pathway for CNTs ³⁰. Photodegradation of CNTs has been shown 112 to transform CNTs by changing their surface chemistry ³¹ or causing a loss of fluorescence when 113 hydrogen peroxide was also present ^{32, 33}, but complete degradation has not been confirmed. Neither 114 pure fungal cultures of white rot fungi, *Trametes versicolor*, nor environmental microbial communities 115 degraded radioactively-labeled SWCNTs after a six-month incubation period ³⁴.

116 Studies testing the enzymatic and microbial degradation of radioactively-labeled multiwall 117 carbon nanotubes (MWCNTs) also showed minimal degradation except when a specific microbial grouping was used ^{35, 36}. The results with radioactively-labeled CNTs contrast with enzymatic studies on 118 non-radioactive CNTs which showed quicker degradation ³⁷. However, the experimental conditions in 119 120 the enzymatic studies did not reflect environmentally relevant concentrations of these enzymes and the 121 CNTs in the study were pre-treated with acid to introduce additional defect sites that should increase the ability of the CNTs to degrade. Thus, these results may not be directly applicable to determining 122 123 environmental persistence values. Overall, the reported data suggests a half-life of CNTs in

2		
3 4	124	environmental systems (soil, sediment, water) greater than 6 months ^{9, 38} . The current persistence scale
5 6 7	125	(P1, P2 and P3) in OPPT is generally based on these guideposts: environmental half-lives lower than 2
7 8 9	126	months (P1), between 2 and 6 months (P2) and greater than 6 months (P3). Therefore, CNTs would be
10 11	127	considered P3 (i.e., high potential for environmental persistence).
12 13 14	128	Field measurements of CNTs, which have only recently entered commerce, are not yet available.
15 16	129	A modeled average CNT surface water concentration in Europe was estimated to be 0.0035 ng/L 39 , a
17 18	130	concentration below the detection limit of all currently available analytical methods ⁴⁰ . However, greater
19 20 21	131	concentrations could be present in the environment at release locations. OPPT does not generally
22 23	132	permit environmental releases of CNTs. As a result of these factors, there are no direct measurements
24 25 26	133	of the bioaccumulation behavior of CNTs in the environment to evaluate.
27 28	134	To assess the potential of CNTs to bioaccumulate or bioconcentrate, we summarize the
29 30	135	literature on CNT bioaccumulation and bioconcentration by invertebrates and non-mammalian
31 32 33	136	vertebrates, and discuss how these measurements were made as well as their implications for assessing
34 35	137	the placement of CNTs in the bioaccumulation component of a PBT framework. Because the
36 37	138	physicochemical behavior of CNTs may be affected by type (single versus multiwall), surface
38 39 40	139	modifications (functionalizations), and exposure conditions, the bioaccumulation studies were reviewed
40 41 42	140	with respect to these factors. In addition, we investigated the extent of CNT absorption across epithelial
43 44	141	tissues and retention of CNTs among species. Other key topics such as general findings on the potential
45 46	142	toxicity of CNTs to ecological receptors and humans ⁴¹⁻⁴³ or the potential for carbon nanotubes to modify
47 48 49	143	the bioaccumulation of co-contaminants ⁴⁴⁻⁴⁷ were not systematically reviewed in this study.
50 51 52	144	Method
53 54	145	We identified recent publications (2005 to 2016) that reviewed single, double-walled and
55 56 57 58	146	multiwall carbon nanotube bioaccumulation and ecotoxicity as the starting point for our summary ^{8, 9, 48,}
59 60		9

Environmental Science: Nano

2		
3		
Λ		
5		
6		
7		
'		
8		
9		
4	~	
I	υ	
1	1	
1	2	
ż	2	
1	3	
1	4	
1	5	
2	2	
1	6	
1	7	
1	8	
2	2	
1	9	
2	0	
ົ	1	
2	1	
2	2	
2	ર	
~	1	
2	4	
2	5	
2	ค	
~	2	
2	1	
2	8	
$\overline{2}$	õ	
_	9	
3	0	
3	1	
Š	÷	
J	2	
3	3	
ર	4	
2	-	
3	5	
3	6	
ົ	7	
3	(
3	8	
3	9	
7	~	
4	υ	
4	1	
4	2	
,	2	
4	3	
4	4	
4	5	
т А	2	
4	6	
4	7	
۵	R	
-	0	
4	9	
5	0	
ភ	1	
ວ -	1	
5	2	
5	3	
- -	1	
c	4	
5	5	
5	6	
-	2	
5	1	
5	8	
F	o	
J	.7	

60

⁴⁹. We reviewed the bioaccumulation behaviors reported in these studies and extracted information on factors that might affect bioaccumulation (detection method, functionalization, exposure concentration, test taxa). We searched the Web of Science to update the article list from the reviews using a range of search terms including, for example, "nanotube" AND "bioaccumulation," to identify manuscripts published up to August 2016.

152 General findings

153 CNTs have been detected and quantified in environmental matrices and organisms using a 154 broad range of analytical techniques, including fluorescence spectroscopy, Raman spectroscopy, 155 electron microscopy, elemental analysis of the metallic impurities in the CNTs, thermal methods, and radiolabeling^{9, 40, 50, 51}. Qualitative measurements (e.g., electron microscopy) do not determine the mass 156 157 or concentration of CNTs but instead only determine their presence or absence, and thus the preferred 158 methods for measuring biodistribution of CNTs are quantitative ones that determines the mass of CNT in 159 organs. While there have been few quantitative CNT biodistribution measurements in organism tissues, numerous qualitative measurements have revealed no absorption of CNTs across the gut tract wall after 160 uptake from the environment in either lower vertebrates or invertebrates ^{8, 48, 52-55}. In addition, 161 162 quantitative measurements of total CNT body burden in organisms have consistently revealed limited bioaccumulation or bioconcentration^{8, 56}. In general, uptake (from the environment) is rapidly followed 163 164 by elimination of CNTs since the presence of food dramatically increases the egestion of significant proportions of ingested CNTs, particularly for aquatic invertebrates ⁵³. In a study investigating 165 166 bioaccumulation in benthic marine organisms, near-infrared spectroscopy was used quantify body 167 burdens of marine taxa (amphipods and mysids) after exposure to SWCNT in sediment and/or food matrices but found no bioaccumulation (measured BAF < 1) 52 . Another study observed no appreciable 168 169 bioaccumulation in any biotic compartments in a wetland mesocosm spiked with SWCNTs in the water column⁵⁷. A summary of studies of CNT bioaccumulation or bioconcentration is provided in Table 1. 170

Environmental Science: Nano

Page 12 of 42

י ר	
2	
3	
4	
5	
6	
7	
8	
a	
10	
10	
11	
12	
13	
14	
15	
16	
17	
18	
10	
13	
20	
21	
22	
23	
24	
25	
26	
27	
28	
20	
29	
30	
31	
32	
33	
34	
35	
36	
37	
20	
20	
39	
40	
41	
42	
43	
44	
45	
46	
47	
71 12	
40 40	
49	
50	
51	
52	
53	
54	
55	
56	
57	
50	
50	
59	
$\mathbf{b}(\mathbf{l})$	

1

171 In addition, studies that have evaluated trophic transfer of SWCNTs using carbon-14 labeled 172 CNTs or near infrared fluorescence either in a mesocosm or a marine benthic food web have shown 173 SWCNTs may be bioavailable for uptake but were rapidly eliminated to below the detection limit during depuration experiments ^{52, 57, 58}. The limits of quantification for NIRF for plants, biofilms, and fish were 174 175 reported to be 1140 ng/g, 250 ng/g, and 780 ng/g (based on wet mass), respectively; these values were determined by concentrations giving analytical signals 3 x blank measurements ⁵⁷. Overall, 176 177 measurements of CNT bioaccumulation using orthogonal techniques have given similar results thus 178 indicating that the results were unlikely to be a result of a bias specific to one of the techniques. 179 As previously mentioned, there are substantial limitations with using equilibrium-based 180 bioaccumulation methodologies (i.e., the correlation of log K_{ow} values to BCF and BAF values) with CNTs 181 since they will not follow the lipid partitioning behavior that is the basis for modeling BCF or BAF using a 182 K_{ow} . There is a lack of data on CNTs to apply criteria recommended by the Pellston Workshop experts for 183 identifying bioaccumulation: e.g., information from field studies, laboratory experimentation, food web 184 modeling, structure-property relationships and molecular computation¹. In addition to reporting the 185 determination of each study of the potential for bioaccumulation or bioconcentration, we also we 186 compared research findings using these metrics (BCFs, BAFs) to enable the most straightforward 187 comparison to regulatory thresholds for the bioaccumulation and bioconcentration determinations for 188 traditional chemicals (Table 1). Across the taxa studied, almost all of the estimates of CNT 189 bioaccumulation are below common regulatory thresholds for designating a chemical as a concern for 190 bioaccumulation. For example, OPPT's New Chemical and TRI programs have adopted two thresholds: BCF> 1000 (B2) and BCF \ge 5000 (B3)²⁴ for characterizing a chemical. Taxa for which BCF \ge 5000 have 191 been estimated include *Daphnia*⁵³ and the chlorophyte *Desmodesmus*⁵⁹. From the perspective of the 192 193 total body burden, the data suggests that there are substantially different bioaccumulation behaviors 194 for CNTs for these species compared to others (e.g., fish, earthworms). However, CNT movement across

Environmental Science: Nano

2	
3	
4	
5	
6	
7	
1	
8	
9	
10	
11	
12	
13	
14	
15	
16	
10	
10	
1Ö	
19	
20	
21	
22	
23	
24	
25	
26	
20	
21	
28	
29	
30	
31	
32	
33	
34	
35	
26	
30	
37	
38	
39	
40	
41	
42	
43	
44	
45	
40	
4/	
48	
49	
50	
51	
52	
53	
54	
55	
55	
30	
5/	
58	
59	

60

195 the gut lining and into the internal tissues has rarely been documented in any organism; evidence of 196 absorption across epithelial tissues in environmentally relevant species exists only for Drosophila¹⁰, but the estimated quantities were small (10⁻⁸ of total dose). Importantly, studies that investigated 197 bioaccumulation using dietary exposure (e.g., ^{10, 34}) or aqueous exposure (e.g., ^{11, 60}) both indicated 198 199 similar low levels of CNT absorption across the gut lining. During the studies investigating aqueous exposures, the extent of CNT settling was often quantified^{60, 61}. Agglomeration of the CNTs and settling 200 out of the water phase was not observed in many studies (e.g., ⁶¹) and is not believed to be the cause of 201 202 the finding that absorption of low bioaccumulation.

203 Studies on bioaccumulation in marine bivalve tissues have found no evidence for bioaccumulation/bioconcentration⁶², or possible absorption across epithelial surfaces only at high 204 exposure concentrations ⁶³. CNTs were detected in the mantle of mussels but occurred potentially as a 205 result of direct surface association of the mantle to the suspended CNTs in the test media as opposed to 206 absorption across epithelial surfaces ⁶³. The small number of reported BCF values for CNTs represent 207 208 uptake into the gut lumen but revealed little to no absorption across the gut tract and into other tissues. 209 When depuration with feeding occurs, there is often a rapid decrease in the gut tract concentration with 210 the CNT concentration often below the detection limit; for example, *Daphnia* exposed to a 211 concentration of 25 µg/L of oxidized or polyethyleneimine functionalized MWCNTs for 24 h nearly fully eliminated (89 % to 99 % of the initial body burden) the MWCNTs after being fed algae for 48 h e.g., ⁶¹. 212 213 Thus, it may be more appropriate to use data from depurated organisms when assessing the 214 bioaccumulation/bioconcentration of nanomaterials, because high BCF values may be predominately or 215 solely a function of high concentrations in the gut tract when filter feeders are used in studies where 216 estimates of whole body are made without purging gut contents prior to analysis. 217 Discussion

Quantifying bioaccumulation

3 4

6

Overall, quantification of CNTs in environmental matrices such as organism tissues, soils, and sediments is challenging because of the difficulties of distinguishing CNTs from the largely carbonaceous background of soils and sediments⁹. Analytical approaches used for hydrophobic organic chemicals are generally not applicable because CNTs samples are often heterogeneous, with varying lengths and diameters and therefore cannot be quantified by chromatographic techniques⁹. In addition, many quantification techniques require extraction of CNTs from these matrices prior to quantitative measurements and these extraction methods are still largely being developed ⁶⁴. The most commonly used approach for quantifying CNT concentrations in complex environmental media (e.g., soils and sediments) and the tissues of ecological receptors to date is through the use of radiolabeled CNTs ^{34, 44, 55, 65-67}. This approach provides unequivocal quantification of the CNTs and avoids potential artifacts encountered when using other measurements of CNT bioaccumulation such as microscopic techniques including SEM and TEM ⁶⁸. While there may be limitations with some quantitative methods with regard to potential artifacts or insufficient detection limits, overall, the quantification methods are considered sufficiently robust that the bioaccumulation findings from these studies are reliable⁹. In other words, the findings described in the previous section are unlikely to result from method-specific artifacts or insufficient limits of detection to determine if, for example, BCF values were greater or less than 1000, the criterion needed to determine if CNTs should be in the B2 or B3 category as discussed above. In addition, recent pioneering advances in near infrared fluorescence microscopy methods ¹⁰ and recently utilized to assess biodistribution in fish ⁶⁹ allow for detection of individual unagglomerated SWCNTs yet still did not show absorption through the gut tract and into other tissues. In addition, similar results have been observed when bioaccumulation studies were conducted in the same or different laboratories using orthogonal techniques (e.g., infrared fluorescence methods and radioactive labeling ⁵² or the microwave method ⁷⁰ and radioactive labeling ⁵⁴,

Environmental Science: Nano

~	
3	
4	
5	
ิล	
2	
1	
8	
9	
1	0
!	0
1	1
1	2
1	ર
	4
1	4
1	5
1	6
1	7
1	<i>i</i>
1	8
1	9
2	0
~	4
2	1
2	2
2	3
ົ	1
~	4
2	5
2	6
2	7
<u>_</u>	0
2	0
2	9
3	0
Q	1
2	י ר
3	2
3	3
3	4
2	5
0 0	0
3	6
3	7
Q	Q
2	0
3	9
4	0
4	1
, /	2
4	~
4	კ
4	4
4	5
л Л	ĥ
4	0
4	1
4	8
۵	ġ
т г	0
С	U
5	1
5	2
5	2
2	3
5	4
5	5
5	6
5	7
<u>с</u>	1
5	8
_	٥

60

⁵⁶). Lastly, similar findings have been observed in studies investigating the bioaccumulation of
 radioactively labeled few layer graphene for several of the same ecological receptors ^{71, 72}.

244 Limitation of current bioaccumulation concept to CNTs

The behavior of CNTs and some other ENMs does not fit classical concept of bioaccumulation, 245 which assumes membrane passage 73 and accumulation into lipid phases 74 . Unlike organic chemicals, 246 247 nanomaterials do not reach thermodynamic equilibrium among the phases during octanol-water distribution measurements ^{27, 75}, although accumulation at the interface of the phases has been 248 observed ⁷⁶. Thus, predictions based on phase partitioning behaviors like log K_{ow} provide a poor fit for 249 250 CNTs in their as-produced form, which are generally not stable long-term in aqueous dispersions 251 without additional dispersants or surfactants (e.g., sodium deoxycholate). CNT stability in solvents is 252 often limited, compared to many organic chemicals. However, it is possible that nanomaterials may 253 associate with more cellular compartments of an organism than just the lipid layers if absorption 254 through the gut tract occurs.

CNTs do not readily pass through the membranes lining the gut lumen and where detected, the 255 quantity of absorbed material is extremely low ¹⁰. Because CNT absorption across the gut has rarely 256 been observed, predators consuming exposed animals will be exposed to CNTs predominately or only in 257 the gut tract of the prey organisms. Given that depuration of CNTs has been observed in feeding studies, 258 259 the concentration of CNTs in the exposed prey organisms would depend on the feeding conditions and 260 whether the organisms are consistently exposed to CNTs or during limited intervals. Moreover, the 261 predators are also unlikely to have absorption of the CNTs through their GI lining, thus indicating a low 262 probability of biomagnification.

We identified no studies that have documented the absorption of CNTs through the gut tract in daphnids even when electron microscopy was used ^{11, 77}. However, absorption of nanomaterials by

2		
3		
4		
5		
6		
7		
'n		
ი ი		
9	~	
1	Ū	
1	1	
1	2	
1	3	
1	4	
1	5	
1	6	
1	7	
1	'n	
1	ი ი	
1	ฮ ก	
2	Ū.	
2	1	
2	2	
2	3	
2	4	
2	5	
2	6	
2	7	
2 2	, 0	
2	0	
2	9	
3	0	
3	1	
3	2	
3	3	
3	4	
3	5	
ર	6	
2	7	
ა ი	י ה	
3	ð	
3	9	
4	0	
4	1	
4	2	
4	3	
4	4	
4	5	
т Л	6	
4 1	0 7	
4	ו ר	
4	8 8	
4	9	
5	0	
5	1	
5	2	
5	3	
5	4	
5	5	
J F	ม ค	
О г	0 7	
5	/ c	
5	8	
5	9	

1

daphnids into tissues other than the gut tract has been observed for quantum dots ⁷⁸, carboxylated 265 polystyrene beads ⁷⁹, and silver nanowires ⁸⁰. It is important to point out that a larger number of 266 267 studies have not identified absorption of nanomaterials through the GI tract and into other tissues by daphnids: quantum dots ⁸¹, fullerenes ⁸², and gold nanoparticles ^{83, 84}. It is currently unclear why 268 269 absorption into systemic circulation is observed in some studies but not others for tests with similar 270 nanomaterials (e.g., quantum dots). This finding could be a result of differences in the nanoparticles 271 themselves (size, charge, surface coating), test organism (e.g., Daphnia age and health), the method 272 used to assess bioaccumulation and associated potential biases, and the method used to conduct the 273 bioaccumulation experiments. Additional research is needed to investigate this topic.

274 With the broad range of potential surface functionalizations being explored for CNT 275 applications, it is important to also consider whether CNT surface characteristics would influence their 276 potential for bioaccumulation or bioconcentration. The bioaccumulation and bioconcentration studies 277 conducted with CNTs with varying surface chemistries have not yet shown distinctly different bioaccumulation results (Table 2), but only a limited number of studies on this topic have been 278 conducted ⁸⁵. In a biodistribution study investigating *D. magna* biodistribution of four types of 279 280 functionalized SWCNTs (i.e., hydroxylated, silicon dioxide, poly aminobenzenesulfonic acid, and 281 polyethylene glycol coated) after mixing with natural organic matter (NOM), none of the functionalized 282 CNTs showed detectable absorption through the gut tract using transmission electron microscopy (TEM) ¹¹. Once CNTs are released into the environment, it is likely that they will be covered with NOM 283 284 regardless of the initial surface functionalization based on the strong adsorption capacity of CNTs for NOM ⁸⁶. Thus, bioaccumulation behaviors of CNTs with varying functionalizations will likely be similar to 285 286 each other in the natural environment in that they will likely be coated with NOM. Continued advances 287 in approaches to quantify CNTs (e.g., near infrared fluorescence spectroscopy, microwave methods) will 288 facilitate the collection of additional bioaccumulation data to explore any potential effects from

Environmental Science: Nano

2	
3	
Δ	
- -	
5	
6	
7	
8	
0	
9	
10	
11	
12	
12	
13	
14	
15	
16	
17	
17	
18	
19	
20	
<u>۲</u> ۱	
22	
23	
24	
27	
25	
26	
27	
20	
20	
29	
30	
31	
22	
32	
33	
34	
35	
20	
36	
37	
38	
30	
39	
40	
41	
42	
12	
43	
44	
45	
46	
47	
47	
48	
49	
50	
50	
51	
52	
53	
50	
J4	
55	
56	
57	
50	
วช	
59	
60	

311

289 different CNT functionalizations on bioaccumulation. Prospective studies should consider investigating if 290 differences in bioaccumulation from these functionalizations persist in the presence of NOM, which would represent more realistic environmental conditions. In the studies conducted in a mesocosm⁵⁷ or 291 with CNTs wrapped with NOM (e.g., ^{11, 77, 60}), similar bioaccumulation findings were observed as 292 293 compared to studies without NOM, namely a lack of absorption across the gut tract, and thus NOM is 294 not expected to change CNT's bioaccumulation behaviors.

Implications for current risk assessment paradigm 295

296 Designating a chemical as bioaccumulative has important regulatory implications. For example, the EU's REACH guidelines give consideration to waiving certain tests for compounds with low potential 297 to bioaccumulate and/or low potential to cross biological membranes to reduce animal testing ⁸⁷. OPPT 298 299 currently considers CNTs a category that may present a potential concern for bioaccumulation due to a lack of data on which to assess their environmental risk ⁸⁸. As a replacement for the traditional 300 301 framework that views the buildup of a chemical in the lipids of fish as the indication that a chemical is 302 bioaccumulative, an alternative framework for nanomaterials would assess first if any material is being absorbed from the gut to other tissues. A growing body of work finds a low potential for 303 304 bioaccumulation for CNTs due to the absence of material being absorbed across the gut tract. The 305 findings of bioaccumulation studies are robust across multiple organisms and multiple quantification 306 methods, and the lines of evidence show a lack of CNT transport across epithelial layers at detectable concentrations. 307

308 While research on CNT trophic transfer potential has mainly been limited to SWCNTs, biomagnification was not identified in aquatic systems ^{52, 57}. In one study on the trophic transfer of 309 310 MWCNTs from bacteria to protozoa, the BMF was below 1 (ranging from 0.01 to 0.04) for all conditions tested^{89,90}. There are also some studies which have demonstrated the capacity for metal-based ENMs

Environmental Science: Nano

1
2
3
4
5
6
7
0
0
9
10
11
12
13
14
15
16
17
18
19
20
21
22
22
20
24
20
20
27
28
29
30
31
32
33
34
35
36
37
38
20
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
5/
55
50
00
ວ/ 50
58
59
60

312	such as gold (Au), cerium dioxide, lanthanum oxide and titanium oxide nanoparticles to be transferred
313	along a food chain ⁹¹⁻⁹⁷ . While one study observed BMF values up to 11.6 when hornworms (<i>Manduca</i>
314	<i>sexta</i>) ingested leaves of tomato plants that had accumulated AuNPs ⁹⁶ , in most studies,
315	biomagnification was also not observed (i.e., BMF <1) ^{92, 94, 95, 97} .
316	Using the traditional measures of bioaccumulation and bioconcentration for CNTs without
317	significant caveats may be misleading ^{29,48} . The lack of absorption into organism tissues is a significant
318	difference between the bioaccumulation behavior of CNTs and dissolved chemicals. Given that exposure
319	concentrations investigated generally exceed modelled environmental concentrations of CNTs by
320	several orders of magnitude and the lack of biomagnification in the studies conducted to date, these
321	findings suggest that the overall potential for trophic transfer should also be considered low.
322	Overall, we recommend that classes of ENMs be investigated on a case by case basis with regard
323	to their potential for bioaccumulation and bioconcentration. This is consistent with scientific
324	recommendations ^{98, 99} and the current United States national policy position on avoiding sweeping
325	generalizations on nano-enabled products ¹⁰⁰ . Based on the literature review and analysis conducted in
326	this paper, CNTs appear to be a group of substances that should be designated low or no concern for
327	bioaccumulation.
328	Acknowledgements
329	R.B. acknowledges the support of EPA and the AAAS Science & Technology Policy Fellowship Program.
330	Disclaimer
331 332 333 334 335	The views expressed in this manuscript are solely those of the authors and do not represent the policies of the U.S. Environmental Protection Agency. Mention of trade names of commercial products should not be interpreted as an endorsement by the U.S. Environmental Protection Agency. Certain commercial product or equipment is described in this paper in order to specify adequately the experimental procedure. In no case does such identification imply recommendation or endorsement by the National

Institute of Standards and Technology, nor does it imply that it is necessarily the best available for the
purpose.

Literat	ure cited
1.	F. A. Gobas, W. de Wolf, L. P. Burkhard, E. Verbruggen and K. Plotzke, Integrated environmental
	assessment and management, 2009, 5 , 624-637.
2.	J. A. Arnot and F. A. P. C. Gobas, Environmental Reviews, 2006, 14, 257-297.
3.	M. F. L. De Volder, S. H. Tawfick, R. H. Baughman and A. J. Hart, <i>Science</i> , 2013, 339 , 535-539.

- NASA and National Nanotechnology Initiative, Technical Interchange Proceedings, Washington,
 D.C., 2014.
- 5. G. M. Rand and S. R. Petrocelli, *Fundamentals of aquatic toxicology: methods and applications*, FMC Corp., Princeton, NJ, 1985.
- R. Stephenson Gerald, G. Ferris Ian, T. Holland Patrick and M. Nordberg, *pac*, 2006, **78**, 2075-2154.
- 7. F. Gottschalk, T. Sun and B. Nowack, *Environmental Pollution*, 2013, **181**, 287-300.
- 8. P. Jackson, N. Jacobsen, A. Baun, R. Birkedal, D. Kuhnel, K. Jensen, U. Vogel and H. Wallin, *Chemistry Central Journal*, 2013, **7**, 154.
- E. J. Petersen, L. Zhang, N. T. Mattison, D. M. O'Carroll, A. J. Whelton, N. Uddin, T. Nguyen, Q. Huang, T. B. Henry and R. D. Holbrook, *Environmental science & technology*, 2011, 45, 9837-9856.
- 10. T. K. Leeuw, R. M. Reith, R. A. Simonette, M. E. Harden, P. Cherukuri, D. A. Tsyboulski, K. M. Beckingham and R. B. Weisman, *Nano Letters*, 2007, **7**, 2650-2654.
- A. J. Edgington, E. J. Petersen, A. A. Herzing, R. Podila, A. Rao and S. J. Klaine, *Nanotoxicology*, 2014, 8, 2-10.
- National Nanotechnology Initiative, *Environmental, health, and safety research needs for engineered nanoscale materials*, Nanoscale Science, Engineering, and Technology Subcommittee, Committee on Technology, National Science and Technology Council, 2006.
- 13. Government of Japan, Journal, 1973.
- US EPA, Methodology for Deriving Ambient Water Quality Criteria for the Protection of Human Health (2000). Technical Support Document, Volume 2: Development of National Bioaccumulation Factors. EPA-822-R-03-030, Office of Science and Technology, Office of Water, U.S. Environmental Protection Agency, 2003.
- 15. F. A. P. C. Gobas, J. B. Wilcockson, R. W. Russell and G. D. Haffner, *Environmental Science & Technology*, 1999, **33**, 133-141.

16. J. P. Connolly and C. J. Pedersen, *Environmental Science & Technology*, 1988, **22**, 99-103.

- 17. A. Leo, C. Hansch and D. Elkins, *Chemical Reviews*, 1971, **71**, 525-616.
- 18. R. Sibley, D. Peakall and S. P. Hopkin, *Principles of Ecotoxicology*, Taylor & Francis, 2000.
- 19. J. L. Hamelink, R. C. Waybrant and R. C. Ball, *Transactions of the American Fisheries Society*, 1971, **100**, 207-214.
- 20. W. B. Neely, D. R. Branson and G. E. Blau, *Environmental Science & Technology*, 1974, **8**, 1113-1115.
- 21. H. O. Esser and P. Moser, *Ecotoxicology and Environmental Safety*, 1982, **6**, 131-148.
- 22. A. D. Abelkop, J. D. Graham and T. V. Royer, *Persistent, Bioaccumulative, and Toxic (PBT) Chemicals: Technical Aspects, Policies, and Practices*, CRC Press, Boca Raton, FL, 2015.
- 23. Indiana University School of Public and Environmental Affairs (SPEA), *Scientific and Policy Analysis of Persistent, Bioaccumulative, and Toxic Chemicals:A Comparison of Practices in Asia, Europe, and North America. The Report of a Consensus Panel*, 2013.
- 24. Federal Register, *Journal*, 1999.
- 25. M. D. Ralston, D. L. Fort, J. H. Jon and J. K. Kwiat, in *Persistent, Bioaccumulative, and Toxic Chemicals II*, American Chemical Society, 2000, vol. 773, ch. 13, pp. 151-181.
- 26. G. D. Veith, D. J. Call and L. T. Brooke, *Canadian Journal of Fisheries and Aquatic Sciences*, 1983,
 40, 743-748.
- 27. A. Praetorius, N. Tufenkji, K.-U. Goss, M. Scheringer, F. von der Kammer and M. Elimelech, *Environmental Science: Nano*, 2014, **1**, 317-323.
- R. D. Handy, G. Cornelis, T. Fernandes, O. Tsyusko, A. Decho, T. Sabo-Attwood, C. Metcalfe, J. A. Steevens, S. J. Klaine, A. A. Koelmans and N. Horne, *Environmental Toxicology and Chemistry*, 2012, **31**, 15-31.
- 29. E. J. Petersen, S. A. Diamond, A. J. Kennedy, G. G. Goss, K. Ho, J. Lead, S. K. Hanna, N. B. Hartmann, K. Hund-Rinke, B. Mader, N. Manier, P. Pandard, E. R. Salinas and P. Sayre, *Environmental Science & Technology*, 2015, **49**, 9532-9547.
- 30. S. W. Bennett, A. Adeleye, Z. Ji and A. A. Keller, *Water Research*, 2013, **47**, 4074-4085.
- 31. J. L. Bitter, J. Yang, S. Beigzadeh Milani, C. T. Jafvert and D. H. Fairbrother, *Environmental Science: Nano*, 2014, **1**, 324-337.
- 32. W.-C. Hou, S. BeigzadehMilani, C. T. Jafvert and R. G. Zepp, *Environmental Science & Technology*, 2014, **48**, 3875-3882.

Environmental Science: Nano

2		
4	33.	WC. Hou, CJ. He, YS. Wang, D. K. Wang and R. G. Zepp, Environmental Science & Technology,
5 6		2016, 50 , 3494-3502.
7	34.	A. N. Parks, G. T. Chandler, K. T. Ho, R. M. Burgess and P. L. Ferguson, Environmental Toxicology
8 9		and Chemistry, 2015, 34 , 247-251.
10 11	35.	L. Zhang, E. J. Petersen, M. Y. Habteselassie, L. Mao and Q. Huang, Environmental Pollution,
12		2013, 181 , 335-339.
13 14	36.	D. X. Flores-Cervantes, H. M. Maes, A. Schäffer, J. Hollender and HP. E. Kohler, Environmental
15 16		Science & Technology, 2014, 48 , 4826-4834.
17	37.	B. L. Allen, P. D. Kichambare, P. Gou, I. I. Vlasova, A. A. Kapralov, N. Konduru, V. E. Kagan and A.
18 19		Star Nano Letters 2008 8 3899-3903
20	20	V Zhao, R. L. Allon and A. Star. The Journal of Physical Chemistry A 2011 11E 0526 0544
21 22	50.	1. Zhao, B. L. Allen and A. Star, <i>The Journal of Physical Chemistry A</i> , 2011, 115 , 9556-9544.
23	39.	F. Gottschalk, T. Sonderer, R. W. Scholz and B. Nowack, Environmental Science & Technology,
24 25		2009, 43 , 9216-9222.
26	40.	E. J. Petersen, D. X. Flores-Cervantes, T. D. Bucheli, L. C. C. Elliott, J. A. Fagan, A. Gogos, S. Hanna,
27 28		R. Kägi, E. Mansfield, A. R. M. Bustos, D. L. Plata, V. Reipa, P. Westerhoff and M. R. Winchester,
29		Environmental Science & Technology, 2016, 50 , 4587-4605.
30 31	41.	A. Shinde and C. S. J. Tsai, Toxicology Research, 2016, 5, 248-258.
32 33	42.	R. Wang, A. N. Meredith, M. Lee, D. Deutsch, L. Miadzvedskaya, E. Braun, P. Pantano, S. Harper
34 35		and R. Draper, Nanotoxicology, 2016, 10, 689-698.
36	43.	B. Zhu, X. Xia, N. Xia, S. Zhang and X. Guo, Environmental Science & Technology, 2014, 48, 4086-
37 38		4095.
39 40	44.	P. L. Ferguson, G. T. Chandler, R. C. Templeton, A. DeMarco, W. A. Scrivens and B. A. Englehart,
40		Environmental Science & Technology 2008 42 3879-3885
42 43	46	A N Darks C T Chandler I M Dartis I C Sullivan M M Darron M C Cantwell D M
44	45.	A. N. Parks, G. T. Chandler, L. M. Portis, J. C. Sunivari, M. M. Perfori, M. G. Cantwell, K. M.
45 46		Burgess, K. T. Ho and P. L. Ferguson, <i>Nanotoxicology</i> , 2014, 8 , 111-117.
47	46.	E. J. Petersen, R. A. Pinto, P. F. Landrum and J. W. J. Weber, Environmental Science &
48 49		Technology, 2009, 43 , 4181-4187.
50	47.	M. H. Shen, X. H. Xia, F. Wang, P. Zhang and X. L. Zhao, <i>Environ. Toxicol. Chem.</i> , 2012, 31 , 202.
51 52	48.	WC. Hou, P. Westerhoff and J. D. Posner, Environmental Science: Processes & Impacts, 2013,
53 54		15 , 103-122.
54 55	49.	S. J. Sarma, I. Bhattacharya, S. K. Brar, R. D. Tyagi and R. y. Surampalli, Critical Reviews in
56 57		Environmental Science and Technology, 2015, 45 , 905-938.
58 59		
60		20

50.	K. Doudrick, P. Herckes and P. Westerhoff, Environmental Science & Technology, 2012, 46,
	12246-12253.
51.	P. E. Rasmussen, ML. Avramescu, I. Jayawardene and H. D. Gardner, Environmental Science &
	Technology, 2015, 49 , 12888-12896.
52.	A. N. Parks, L. M. Portis, P. A. Schierz, K. M. Washburn, M. M. Perron, R. M. Burgess, K. T. Ho, G.
	T. Chandler and P. L. Ferguson, Environmental Toxicology and Chemistry, 2013, 32, 1270-1277.
53.	E. J. Petersen, J. Akkanen, J. V. Kukkonen and W. J. Weber Jr, Environmental Science &
	Technology, 2009, 43 , 2969-2975.
54.	E. J. Petersen, Q. Huang and J. W. J. Weber, Environmental Science & Technology, 2008, 42,
	3090-3095.
55.	E. J. Petersen, Q. Huang and W. Weber, Environmental Health Perspectives, 2008, 116, 496.
56.	E. J. Petersen, R. A. Pinto, L. Zhang, Q. Huang, P. F. Landrum and W. J. Weber, Environmental
	Science & Technology, 2011, 45 , 3718-3724.
57.	A. Schierz, B. Espinasse, M. R. Wiesner, J. H. Bisesi, T. Sabo-Attwood and P. L. Ferguson,
	Environmental Science: Nano, 2014, 1, 574-583.
58.	A. Schierz, A. N. Parks, K. M. Washburn, G. T. Chandler and P. L. Ferguson, Environ. Sci. Technol.,
	2012, 46 , 12262.
59.	S. Rhiem, M. J. Riding, W. Baumgartner, F. L. Martin, K. T. Semple, K. C. Jones, A. Schäffer and H.
	M. Maes, Environmental Pollution, 2015, 196, 431-439.
60.	H. M. Maes, F. Stibany, S. Giefers, B. Daniels, B. Deutschmann, W. Baumgartner and A.
	Schaeffer, Environmental Science & Technology, 2014, 48, 12256–12264.
61.	E. J. Petersen, R. A. Pinto, D. J. Mai, P. F. Landrum and W. J. Weber, Environmental Science &
	Technology, 2011, 45 , 1133-1138.
62.	A. N. Parks, R. M. Burgess, K. T. Ho and P. L. Ferguson, Integrated Environmental Assessment and
	Management, 2014, 10 , 471-478.
63.	S. K. Hanna, R. J. Miller and H. S. Lenihan, Journal of Hazardous Materials, 2014, 279, 32-37.
64.	K. Doudrick, N. Corson, G. Oberdörster, A. C. Eder, P. Herckes, R. U. Halden and P. Westerhoff,
	ACS Nano, 2013, 7 , 8849-8856.
65.	L. Zhang, E. J. Petersen, W. Zhang, Y. Chen, M. Cabrera and Q. Huang, Environmental Pollution,
	2012, 166 , 75-81.
66.	L. Zhang, E. J. Petersen and Q. Huang, Environmental Science & Technology, 2011, 45, 1356-
	1362.
	71
	Z 1

3 4

Environmental Science: Nano

3 4	67.	Q. Zhao, E. J. Petersen, G. Cornelis, X. Wang, X. Guo, S. Tao and B. Xing, Carbon, 2016, 99, 229-
5		237.
7	68.	E. J. Petersen, T. B. Henry, J. Zhao, R. I. MacCuspie, T. L. Kirschling, M. A. Dobrovolskaia, V.
8 9		Hackley, B. Xing and J. C. White, Environmental Science & Technology, 2014, 48, 4226-4246.
10	69.	J. H. Bisesi, J. Merten, K. Liu, A. N. Parks, A. R. M. N. Afrooz, J. B. Glenn, S. J. Klaine, A. S. Kane, N.
11 12		B Saleh P I Ferguson and T Sabo-Attwood Environmental Science & Technology 2014 48
13		
14 15		
16	70.	S. LI, F. Irin, F. O. Atore, M. J. Green and J. E. Canas-Carrell, Science of the Total Environment,
17 18		2013, 445–446 , 9-13.
19	71.	X. Guo, S. Dong, E. J. Petersen, S. Gao, Q. Huang and L. Mao, Environmental Science &
20 21		Technology, 2013, 47 , 12524-12531.
22	72.	L. Mao, C. Liu, K. Lu, Y. Su, C. Gu, Q. Huang and E. J. Petersen, <i>Carbon</i> , 2016, 109 , 566-574.
24	73.	Y. Zhang, L. Zhu, Y. Zhou and J. Chen, <i>Journal of Environmental Sciences</i> , 2015, 30 , 223-230.
25 26	74.	D. Mackay and A. Fraser, Environmental Pollution, 2000, 110, 375-391.
27	75.	E. J. Petersen, Q. Huang and W. J. Weber, Environmental Toxicology and Chemistry, 2010, 29,
29		1106-1112.
30 31	76.	K. D. Hristovski, P. K. Westerhoff and J. D. Posner, Journal of Environmental Science and Health,
32 33		Part A, 2011, 46 , 636-647.
34 35	77.	A. J. Edgington, A. P. Roberts, L. M. Taylor, M. M. Alloy, J. Reppert, A. M. Rao, J. Mao and S. J.
36		Klaine, Environmental Toxicology and Chemistry, 2010, 29, 2511-2518.
37 38	78.	A. Feswick, R. J. Griffitt, K. Siebein and D. S. Barber, Aquatic Toxicology, 2013, 130–131, 210-218.
39 40	79.	P. Rosenkranz, Q. Chaudhry, V. Stone and T. F. Fernandes, Environmental Toxicology and
41 42		Chemistry, 2009, 28 , 2142-2149.
43	80.	L. D. Scanlan, R. B. Reed, A. V. Loguinov, P. Antczak, A. Tagmount, S. Aloni, D. T. Nowinski, P.
44 45		Luong, C. Tran, N. Karunaratne, D. Pham, X. X. Lin, F. Falciani, C. P. Higgins, J. F. Ranville, C. D.
46 47		Vulpe and B. Gilbert, ACS Nano, 2013, 7 , 10681-10694.
48	81.	B. Jackson, H. Pace, A. Lanzirotti, R. Smith and J. Ranville, Anal Bioanal Chem, 2009, 394 , 911-
49 50		917.
51 52	82.	K. Tervonen, G. Waissi, E. J. Petersen, J. Akkanen and J. V. K. Kukkonen, Environmental
53		Toxicology and Chemistry, 2010, 29 , 1072-1078.
54 55	83.	S. B. Lovern, H. A. Owen and R. Klaper, Nanotoxicology, 2008, 2 , 43-48.
56 57	84.	A. T. Wray and S. J. Klaine, Environmental Toxicology and Chemistry, 2015, 34, 860-872.
58 59		
60		22

- 85. A. M. Cano, K. Kohl, S. Deleon, P. Payton, F. Irin, M. Saed, S. A. Shah, M. J. Green and J. E. Cañas-Carrell, *Chemosphere*, 2016, **152**, 117-122.
 - 86. K. Yang and B. Xing, *Environmental Pollution*, 2009, **157**, 1095-1100.
 - M. Müller and M. Nendza, *Final Report . Literature Study: Effects of molecular size and lipid solubility on bioaccumulation potential*, Fraunhofer Institute forMolecular Biology and Applied Ecology and Analytisches Laboratorium fürUmweltuntersuchungen und Auftragsforschung, 2007.
 - 88. US EPA, Interim Technical Guidance for assessing screening level environmental fate and transport of, and general population, consumer and environmental exposure to nanomaterials. Unpublished report, US EPA, Office of Pollution Prevention and Toxics, 2010.
 - 89. M. Mortimer, E. Petersen, B. Buchholz and P. Holden, *Nanomaterials*, 2016, **6**, 181.
 - 90. M. Mortimer, E. J. Petersen, B. A. Buchholz, E. Orias and P. A. Holden, *Environmental Science & Technology*, 2016, **50**, 8876-8885.
 - 91. J. M. Unrine, S. E. Hunyadi, O. V. Tsyusko, W. Rao, W. A. Shoults-Wilson and P. M. Bertsch, *Environmental Science & Technology*, 2010, **44**, 8308-8313.
 - 92. J. M. Unrine, W. A. Shoults-Wilson, O. Zhurbich, P. M. Bertsch and O. V. Tsyusko, *Environmental Science & Technology*, 2012, **46**, 9753-9760.
 - 93. X. Zhu, J. Wang, X. Zhang, Y. Chang and Y. Chen, *Chemosphere*, 2010, **79**, 928-933.
 - 94. J. Hawthorne, R. De la Torre Roche, B. Xing, L. A. Newman, X. Ma, S. Majumdar, J. Gardea-Torresdey and J. C. White, *Environmental Science & Technology*, 2014, **48**, 13102-13109.
 - 95. R. De la Torre Roche, A. Servin, J. Hawthorne, B. Xing, L. A. Newman, X. Ma, G. Chen and J. C. White, *Environmental Science & Technology*, 2015, **49**, 11866-11874.
 - 96. J. D. Judy, J. M. Unrine and P. M. Bertsch, *Environmental Science & Technology*, 2011, **45**, 776-781.
 - 97. J. D. Judy, J. M. Unrine, W. Rao and P. M. Bertsch, *Environmental Science & Technology*, 2012,
 46, 12672.
 - 98. A. J. Kennedy, M. S. Hull, J. A. Steevens, K. M. Dontsova, M. A. Chappell, J. C. Gunter and C. A.
 Weiss, *Environmental Toxicology and Chemistry*, 2008, 27, 1932-1941.
 - 99. H. Godwin, C. Nameth, D. Avery, L. L. Bergeson, D. Bernard, E. Beryt, W. Boyes, S. Brown, A. J.
 Clippinger, Y. Cohen, M. Doa, C. O. Hendren, P. Holden, K. Houck, A. B. Kane, F. Klaessig, T.
 Kodas, R. Landsiedel, I. Lynch, T. Malloy, M. B. Miller, J. Muller, G. Oberdorster, E. J. Petersen, R.

Environmental Science: Nano

	C. Pleus, P. Sayre, V. Stone, K. M. Sullivan, J. Tentschert, P. Wallis and A. E. Nel, ACS Nano, 2015,
	9 , 3409-3417.
100.	J. P. Holdren, C. R. Sunstein and I. A. Siddiqui, Policy Principles for the U.S. Decision-making
	Concerning Regulation and Oversight of Applications of Nanotechnology and Nanomaterials,
	Executive Office of the President of the United States, 2011.
101.	F. Mouchet, P. Landois, E. Sarremejean, G. Bernard, P. Puech, E. Pinelli, E. Flahaut and L.
	Gauthier, Aquatic Toxicology, 2008, 87, 127-137.
102.	J. N. Mwangi, N. Wang, C. G. Ingersoll, D. K. Hardesty, E. L. Brunson, H. Li and B. Deng,
	Environmental Toxicology and Chemistry, 2012, 31 , 1823-1830.
103.	E. Smirnova, A. Gusev, O. Zaytseva, O. Sheina, A. Tkachev, E. Kuznetsova, E. Lazareva, G.
	Onishchenko, A. Feofanov and M. Kirpichnikov, Front. Chem. Sci. Eng., 2012, 6, 132-138.
104.	X. Liu, D. Vinson, D. Abt, R. H. Hurt and D. M. Rand, Environmental Science & Technology, 2009,
	43 , 6357-6363.
105.	G. Zhai, S. M. Gutowski, K. S. Walters, B. Yan and J. L. Schnoor, Environmental Science &
	Technology, 2015, 49 , 7380-7390.
106.	M. H. Lahiani, E. Dervishi, J. Chen, Z. Nima, A. Gaume, A. S. Biris and M. V. Khodakovskaya, ACS
	Appl. Mater. Interfaces, 2013, 5 , 7965.
107.	S. Lin, J. Reppert, Q. Hu, J. S. Hudson, M. L. Reid, T. A. Ratnikova, A. M. Rao, H. Luo and P. Ke,
	Small, 2009, 5 , 1128.
108.	C. Larue, M. Pinault, B. Czarny, D. Georgin, D. Jaillard, N. Bendiab, M. Mayne-L'Hermite, F. Taran,
	V. Dive and M. Carrière, Journal of Hazardous Materials, 2012, 227–228, 155-163.
109.	M. Li and C. P. Huang, <i>Carbon</i> , 2011, 49 , 1672-1679.
110.	Y. Zhu, Q. Zhao, Y. Li, X. Cai and W. Li, Journal of Nanoscience and Nanotechnology, 2006, 6,
	1357-1364.
111.	G. Chen, J. Qiu, Y. Liu, R. Jiang, S. Cai, Y. Liu, F. Zhu, F. Zeng, T. Luan and G. Ouyang, Scientific
	Reports, 2015, 5 , 15682.
112.	M. C. Martínez-Ballesta, L. Zapata, N. Chalbi and M. Carvajal, Journal of Nanobiotechnology,
	2016, 14 , 1-14.
113.	M. H. Lahiani, E. Dervishi, I. Ivanov, J. H. Chen and M. Khodakovskaya, Nanotechnology, 2016,
	27 , 13.
114.	A. Gogos, J. Moll, F. Klingenfuss, M. van der Heijden, F. Irin, M. J. Green, R. Zenobi and T. D.
	Bucheli, Journal of Nanobiotechnology, 2016, 14 , 1-11.
	24

- 115. C. J. Smith, B. J. Shaw and R. D. Handy, *Aquatic Toxicology*, 2007, **82**, 94-109.
- 116. J. H. Bisesi, T. Ngo, S. Ponnavolu, K. Liu, C. M. Lavelle, A. Afrooz, N. B. Saleh, P. L. Ferguson, N. D. Denslow and T. Sabo-Attwood, *Nanomaterials*, 2015, **5**, 1066-1086.
- 117. M. Yang, S. Kwon, Y. Kostov, A. Rasooly, G. Rao and U. Ghosh, *Environ Chem Lett*, 2011, **9**, 235-241.
- 118. A. P. Roberts, A. S. Mount, B. Seda, J. Souther, R. Qiao, S. Lin, P. C. Ke, A. M. Rao and S. J. Klaine, *Environmental Science & Technology*, 2007, **41**, 3025-3029.
- 119. T. Galloway, C. Lewis, I. Dolciotti, B. D. Johnston, J. Moger and F. Regoli, *Environmental Pollution*, 2010, **158**, 1748-1755.
- 120. P. Ghafari, C. H. St-Denis, M. E. Power, X. Jin, V. Tsou, H. S. Mandal, N. C. Bols and X. Tang, *Nature Nanotechnology*, 2008, **3**, 347-351.
- 121. F. Mouchet, P. Landois, E. Flahaut, E. Pinelli and L. Gauthier, *Nanotoxicology*, 2007, **1**, 149-156.

10

Table 1. Summary of qualitative and quantitative carbon nanotube bioaccumulation results for single-walled carbon nanotubes (SWCNT), double-walled (DWCNT), and multiwall carbon

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referenc
DWCNT	ND ^b	0.7 nm to 2.2 nm	Acid-purified	Ramen spectroscopy; scanning electron microscopy (SEM)	Aqueous medium (distilled tap water with nutritive salts).	10 mg/L to 500 mg/L	12 days	Amphibian	Xenopus laevis		Masses of CNT accumulated on gills and gut tract of the tadpoles.	101
DWCNT	ND	1.2 nm to 3.2 nm	Acid-purified	Field Emission Gun SEM/ High resolution Transmission Electron Microscopy (TEM)	Aqueous medium (distilled tap water with nutritive salts).	1 mg/L to 1000 mg/L	12 days	Amphibian	Ambystoma mexicanum		Ingested CNTs accumulated in the gut even at lowest exposures tested.	
MWCNT	<1 μm	5 nm to 20 nm	None	Carbon-14 labeling,TEM	Aqueous medium	1 mg/L	24 h to 72 h	Algae	Desmodesmus subspicatus	5000 (BCF)	Most material agglomerated around cells, but single CNTs were detected in the cytoplasm. Large amounts of CNTs detached from the cells after moving them to water without CNTs.	59
MWCNT	5 μm to 15 μm	10-20 nm	None or Acid- purified	TEM	Sand substrate in aqueous medium	10 ³ mg/L	14 days	Amphipod	Hyalella azteca		Qualitative determination of the presence of CNTs in the gut as well as on the outer surface of the test organisms, although no evidence of penetration of CNTs through cell membranes. No significant removal on depuration.	102
MWCNT	2 μm	20 nm to 70 nm	None	TEM	Substrate of gauze layers moistened with MWCNT solution	0 mg/L to (1 x 10 ³) mg/L	10 days	Angiosperm	Onobrychis arenaria		Qualitative demonstration in plant seedlings of the translocation of MWCNTs from the roots via stems to leaves.	- 103
MWCNT	0.5 μm	36.5 nm ± 12.7 nm	Acid-purified, 14C- labeled	SEM; AMS (accelerator mass spectrometry);	In growth medium	0.01 mg/L, 1 mg/L		Bacteria	Pseudomonas aeruginosa		MWCNTs identified in endosperm.	90
						26						

Environmental Science: Nano

	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referen
				LSC								
IWCNT	5 μm to 15 μm	10 nm to 20 nm	None/Acid- purified	TEM		10 ³ mg/L	14 days	Dipterid	Chironomus dilutus		Qualitative determination of the presence of CNTs in the gut as well as on the outer surface of the test organisms, although no evidence of penetration of CNTs through cell membranes. No significant removal on depuration.	102
IWCNT	ND	ND	None	Field Emission Scanning Electron microscopy	Added to larval food gel	100 mg/kg 10 ³ mg/kg	4 days	Dipterid	Drosophila melanogaster		Nanomaterials observed as dark concentrations in tissues of hatched adults. Nanomaterials consumed by the larvae were assimilated into the developing fly and sequestered into the tissue	104
IWCNT	0.2 μm to 1 μm	ND	None	Carbon-14 labeling	Aqueous medium with/without organic matter	1 mg/L	7 days	Fish	Danio rerio	16 (wet) and 73 (dry) (BCF)	MWCNTs mainly accumulated in the gut, but large relative amounts of radioactivity were also detected in gills, skin, and muscle samples of briefly exposed fish (3 h). MWCNTs were largely eliminated via the digestive tract. In the presence of DOC, 10-fold decrease in uptake after 48 h. No distribution to the liver, the gonads, and the brain was observed. Low amounts of radioactivity were detected in the blood of fish exposed for more than 1 week	60

Environmental Science: Nano

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Reference
MWCNT	0.5 μm to 2 μm	10 nm to 20 nm	None	Raman microscopy	Suspended in Pluronic F-108	50-200 mg/L	5 days post fertilizatio n	Fish (embryo)	Danio rerio		Accumulation in embryos were exposure concentration- dependent.	42
MWCNT	0.5 μm to 2 μm	10 nm to 20 nm	Carboxylated	Raman microscopy	Suspended in Pluronic F-108	50-200 mg/L	5 days post fertilizatio n	Fish (embryo)	Danio rerio		Accumulation in embryos were exposure concentration- dependent.	42
MWCNT	0.05 to 2.0 μm	20 nm to 30 nm	Amine and carboxylate functionalization	TEM	Hydroponically	10mg/L to 50 mg/L	18 days	Legume (soy bean) and monocot (corn)	<i>Glycine max</i> and <i>Zee mays</i>		MWCNTs accumulated in the xylem and phloem cells and within intracellular sites.	105
MWCNT	0.05 μm to 2.0 μm	20 nm to 30 nm	Carboxylate- functionalized	TEM	Hydroponic solution	10 mg/L to 50 mg/L	18 days	Legume (Soy bean) and monocot (corn)	Glycine max and Z. mays		MWCNT accumulated in xylem and phloem and intracellular sites. Stems had lower levels of MWCNTs. Functionalization did not affect uptake and translocation.	105
MWCNT		15nm to 40 nm	HCL-purified and carboxylated	TEM; Raman spectroscopy	In growth medium	100 mg/L	11 days	Legume (soy bean),	G. max.		Aggregates of MWCNTs detected inside the endosperm of exposed seeds	106
MWCNT		15nm to 40 nm	HCL-purified and carboxylated	TEM; Raman spectroscopy	Deposited through air spray on seed	25-100 mg/L	24 h	Monocot (corn and barley) Legume (soy bean), barley	Z. mays, Hordeum vulgare and G. max.		Varied-size clusters of MWCNTs detected inside the endosperm of exposed seeds.	106
MWCNT	0.5 μm to 2 μm	40 nm to 70 nm		Raman spectroscopy	In germination medium	2.5mg/L to 800 mg/L	14 days	Monocot (rice)	Oryza sativa		The uptake of MWNTs at concentrations of 20 mg/L to800 mg/L was found to be insignificant, with some aggregates appearing in the vascular system and almost none in the plant tissue.	107

Environmental Science: Nano

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Reference
MWCNT	2.65 μm ±1.55 μm	10 nm to 150 nm	MWCNT stabilized in gum Arabic and humic acids	Radioimaging, TEM, Raman spectroscopy	in hydroponic media	50 mg/L	7 days	Monocot (wheat) and rosid (rapeseed)	Triticum aestivum; Brassica napus	Transfer Factor ⁶ 4.739 × $10^{-6} \pm 1.126$ × $10^{-6} \pm 1.126$ × $10^{-6} \pm 1.126$ × $10^{-6} \pm 1.126$ × 10^{-6} dispersed in gum Arabic (GA). In humic acid (HA), 1.113 × $10^{-6} \pm 0.066 \times 10^{-6}$. For rapeseed- 1.699 × $10^{-6} \pm 0.694 \times 10^{-6}$ in GA and 0.830 × $10^{-6} \pm 0.276$ × 10^{-6}	Radioimaging qualitatively demonstrated uptake of MWCNT by plant roots and translocated to leaves.	108
MWCNT	386 nm to 407 nm	30 nm to 70 nm	HCL purified or acid oxidized MWCNT	Carbon-14 labeling	Sediment	0.037 mg/g	14 days	Oligochaete	Lumbriculus variegatus	BAF for acid- purified CNT in peat- amended sediment = $0.39 (\pm 0.08)$ and non- amended sediment= $0.67 (\pm 0.026)$	Oxidizing the MWCNT had no effect on BAF.	75
MWCNT	386 nm to 407 nm ⁴	30 nm to 70 nm	HCl purified	Carbon-14 labeling	Sediment	(3.7 x10 ²) mg/kg	28 days	Oligochaete	Lumbriculus variegatus	0.40 ± 0.1 (BAF)	Bioaccumulation factors and order of magnitude lower than PAHs. Almost complete depuration after 3 days in CNT- free sediment or water.	55
MWCNT	386 nm to 407 nm ^d	30 nm to 70 nm	HCl purified	Carbon-14 labeling	Soil	30 mg/kg and (3 x10 ²) mg/kg	14 days	Oligochaete	Eisenia foetida.	0.023±0.01, 0.014±0.003, 0.016±0.001 (BAF)	MWCNTs into the tissues of E. foetida is minimal in comparison to that of a representative PAH counterpart, pyrene.	54

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Reference
MWCNT	407 nm	30 nm to 70 nm	Polyethyleneimine (PEI) coating w/ negative, positive, or neutral surface charges	Carbon-14 labeling	Sediment	(5x 10 ²) mg/kg	28 days	Oligochaete	Eisenia fetida	0.03 (BAF)	No substantial absorption of carbon nanotubes having PEI surface modifications. The PEI-grafted MWCNTs had higher BAF values compared to the nonmodified MWCNTs, but standard deviations were consistently large, hindering definitive conclusions about relative uptake rates.	56
MWCNT	10 μm 20 μm	30 nm to 50 nm	None	Microwave method	Soil	(3 x 10 ³) mg/kg	14 days	Oligochaete	Eisenia fetida	0.015±0.004 (BAF)	Low potential to bioaccumulate; minimal uptake and ready elimination on depuration.	70
MWCNT	5 μm to 15 μm	10 nm to20 nm	None/Acid- purified	TEM	Water exposure but sand was provided as a substrate	(1x 10 ³) mg/L	14 days	Oligochaete	Lumbriculus variegatus		Qualitative determination of the presence of CNTs in the gut as well as on the outer surface of the test organisms, although no evidence of penetration of CNTs through cell membranes.	102
MWCNT	10 μm to30 μm	10 nm to 30 nm	Hydroxylated and carboxylated MWCNTs	ТЕМ	48 h waterborne exposure to 32 mg/L to 120.2 mg CNT/L in water with or without algae as food	32 mg/L to 120.2 mg/L	48 h	Planktonic crustacea	Ceriodaphnia dubia		Qualitative demonstration of MWCNT retention in gut at all concentrations.	98
MWCNT	407 nm	30 nm to 70 nm	Acid-oxidized	Carbon-14 labeling	waterborne exposure	0.04 mg/L, 0.1 mg/L and 0.4 mg/L	48 h	Planktonic crustacea	Daphnia magna	360000 ± 40000, 440000 ± 190000, and 350000 ± 80000 (BCF)	Minimal depuration w/o feeding, however the fraction released rises 50 % to 85 % depurated with feeding.	53
MWCNT	407 nm	30 nm to 70 nm	Polyethyleneimine (PEI) coating w/ negative, positive, or neutral surface charges	Carbon-14 labeling	in artificial freshwater	0.025 mg/L, 0.25 mg/L	48 h	Planktonic crustacea	Daphnia magna	6000 to 46000 (BCF)	Surface coating did not substantially affect accumulation or elimination rate	61
						30						

Environmental Science: Nano

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referen
MWCNT	ND	10 nm to 70 nm	Ozone - treated	TEM, XRD	Aqueous medium	10 mg/L	24 h	Planktonic crustacea	Ceriodaphnia dubia		Nanoparticle accumulation in brood chamber and digestive tract. CNTs Largely eliminated during depuration.	109
MWCNT	ND	ND	Bisphosphonic acid	TEM, Raman spectroscopy	Filtered pond water medium	0.1 mg/L to 200 mg/L	5 days	Protozoa	Stylonychia mytilus		MWCNTs exclusively localized to the mitochondria of the cells.	110
MWCNT	0.5 μm	36.5 nm ± 12.7 nm	Acid-purified, 14C- labeled	SEM; AMS LSC	In growth medium	0.3 mg/L, 1 mg/L	22 h	Protozoa	Tetrahymena thermophila	2900 ± 800 L/kg (at 0.3mg/L MWCNT exposure) 1200 ± 800 L/kg (at 1 mg/L, BCF)	BCF estimates were highest after 2 h (35,000 ± 16000). MWCNT accumulated in the protozoan did not biomagnify, based on estimated BMFs (<1).	90
MWCNT	0.5 μm	36.5 nm ± 12.7 nm	Acid-purified, 14C- labeled	SEM; AMS	Bacteria exposed to 0.01 mg/L MWCNTs	0.004 mg/L to protozoans	22 h	Protozoa	Tetrahymena thermophila	790 ± 200 (BCF)	BCF estimates were highest after 16 h (2200 ± 900). MWCNT accumulated in the protozoan did not biomagnify, based on estimated BMFs (<1).	90
MWCNT	0.5 μm	36.5 nm ± 12.7 nm	Acid-purified, 14C- labeled	SEM, LSC	Bacteria exposed to 1 mg/L MWCNTs	0.3 mg/L to protozoans	22 h	Protozoa	Tetrahymena thermophila	790 ± 300 (BCF)	BCF estimates were highest after 16 h (5700 ± 3000). MWCNT accumulated in the protozoan did not biomagnify, based on estimated BMFs (<1).	90
MWCNT	NS	10 nm	Acid-purified, carboxylated	TEM and Raman spectroscopy	In aqueous solutions	1, 10 mg/L	16 days	Rosid (mustard)	Brassica juncea		MWCNTs permeated into roots of intact plants.	111
MWCNT	0.1 μm to 0.5 μm	6 nm to 9 nm		TEM	In nutrient solution	10- 60 mg/L	7 days	Rosid (broccoli)	Brassica Oleracea		MCNTs were taken up in the roots localized in cell vacuole, intercellular space and cytoplasm. No MWCNTs were detected in plant leaves.	112
MWCNT	0.1 μm to 0.5 μm	6 nm to 9 nm		TEM	In nutrient solution, with 12mM NaCl to create salt- stressed	10 mg/L	7 days	Rosid (broccoli)	Brassica Oleracea		Saline-stressed plants showed a higher accumulation of isolated MWCNTs than non-saline treated plants.	112

Environmental Science: Nano

MWCNT 1 μ 10 MWCNT 1 μ 12	μm to 0 μm μm to	100 nm to 200 nm	None; helical morphology	TEM and Raman spectroscopy	In tobacco	50 mg/l					
MWCNT 1 µ 12	µm to				callus growth medium	00 mg/ 2	24 h (seed) 10 days (seedling)	Solanid (tomato)	Lycospersicon esculentum	MWCNTs identified in endosperm.	113
	2 μm	13 nm to 18 nm	Carboxylate- functionalized, long morphology	TEM and Raman spectroscopy	In tobacco callus growth medium	50 mg/L	24 hours (seed) 10 days (seedling)	Solanid (tomato)	Lycospersicon esculentum	Black aggregates of MWCNTs identified in endosperm.	113
MWCNT 0.5 to	.5 μm ο 2 μm	20nm to 30 nm	Carboxylate- functionalized, short morphology	TEM and Raman spectroscopy	In tobacco callus growth medium	50 mg/L	24 hours (seed) 10 days (seedling)	Solanid (tomato)	Lycospersicon esculentum	MWCNTs identified in endosperm.	113
MWCNT 0.0 to μm	.05 μm ο 2.0 .m	20 nm to 30 nm	None	TEM	Hydroponic solution	10 mg/L to 50 mg/L	18 days	Soybeans and corn	Glycine max and Z. mays	MWCNT accumulated in xylem and phloem and intracellular sites. Stems had lower levels of MWCNTs.	105
MWCNT 0.C to μm	.05 μm ο 2.0 .m	20 nm to 30 nm	Amine- functionalized	TEM	Hydroponic solution	10 mg/L to 50 mg/L	18 days	Soybeans and corn	Glycine max and Z. mays	MWCNT accumulated in xylem and phloem and intracellular sites. Stems had lower levels of MWCNTs. Functionalization did not affect uptake and translocation.	105
MWCNT 10 30	0 μm to 0 μm	20 nm to 30 nm	None	Microwave- induced heating (MIH); multi- angle light scattering; Raman spectroscopy; TEM	Soil	3 mg/kg and 2933 mg/kg	14 weeks	Wheat and corn	Triticum spp., Z. mays	Levels of MWCNT taken up could not be fully quantified since they were below the limit of quantification. TEM imaging of root cross sections was not conclusive. Two estimates of translocation of MWCNT into plants were above the limits of detection (0.15 % and 9.8 %).	114

Environmental Science: Nano

Гуре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referen
SWCNT	5 μm to 15 μm	2 nm	None/Acid- purified	TEM	In waer	(1x10 ³) mg/L	14 days	Amphipod	Hyalella azteca		Qualitative determination of the presence of CNTs in the gut as well as on the outer surface of the test organisms, although no evidence of penetration of CNTs through cell membranes. No significant removal on depuration.	102
SWCNT	ND	1.22 nm to 1.96 nm	None	Fluorimetry and stable isotope analysis	Filtered seawater with phytoplankton feed	1 mg/L to 3 mg/L	28 days	Bivalve mollusc (marine)	Mytilus galloprovincialis		CNT accumulated in biodeposit (feces and pseudofeces). Metal residues associated with CNTS were detected in Visceral, mantle and gill tissue ((0.04 ± 0.02) mg/g to (1.04 ± 0.1) mg CNTs/g tissue.	63
SWCNT	ND		None	Near-infrared Fluorescence spectroscopy (NIRF)	SWCNT-spiked algal (<i>isochrysis</i> <i>galbana</i>) food.	100 mg/ kg, (1 x 10 ³) mg/ kg SWCNT- amended algae	14 days	Bivalve mollusc (marine)	Mercenaria mercenaria		No evidence marine bivalves) fed marine algae (<i>lsochrysis galbana</i>) exposed to SWCNT accumulated SWCNT, or that the bivalve served as a vector for SWCNT to polychaetes consuming the bivalves.	62
SWCNT	500 nm to 1.5 μm	4 nm to 5 nm	Carboxylated SWCNT in marine sediment	Carbon-14 labeling	marine sediment	(3.64 x10 ²) mg/kg	14 days	Copepod	Amphiascus tenuiremis		No detectable bioaccumulation after depuration.	44
SWCNT	5 μm to 15 μm	2 nm	None/Acid- purified	TEM	In water	(1 x10 ³) mg/L	14 days	Dipterid	Chironomus dilutus		Qualitative determination of the presence of CNTs in the gut as well as on the outer surface of the test organisms, although no evidence of penetration of CNTs through cell	102
						33						

Page 35 of 42

47 48 40

Environmental Science: Nano

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referen
											membranes. No significant removal on depuration.	
SWCNT	ND		None	NIRF	Paste containing SWCNTs containing SWCNTs in BSA buffer in yeast paste	25 mg/L	4 days to 5 days	Dipterid	D.melanogaster		Only a tiny fraction (10 ⁻⁸) of these SWCNTs become incorporated into tissues. after traversing the gut wall, nanotubes in the hemolymph accumulate in the dorsal vessel as a result of its pumping action.	10
SWCNT	ND	ND	None	Field Emission Scanning Electron microscopy	Added to larval food gel	100 mg/kg 10 ³ mg/kg	4 days	Dipterid	Drosophila melanogaster		Nanomaterials observed as dark concentrations in tissues of hatched adults. Nanomaterials consumed by the larvae were assimilated into the developing fly and sequestered into the tissue	104
SWCNT	ND	0.7 nm to 1.3 nm	acid-purified and carboxylated	NIRF and carbon- 14 labeling	Added SWCNTS to food source	10 mg/kg and 100 ¹⁴ C mg- SWCNT/kg dried algae + / or sediment	28 days	Estuarine amphipod	Leptocheirus plumulosus	0.013 ± 0.002 to 0.068 ± 0.016 (nondepurate d); 0.0040 ± 0.0008- to0.0074 ± 0.0012 (depurated) (BAF)	Nondepurated organisms exposed to SWCNT amended sediment and algae had significantly elevated body burden. Uptake via sediment was more critical for accumulation than uptake via algae for amphipods. After 24 h depuration, only the highest SWCNT-amended sediment and algae treatment showed significantly increased body burden compared to background.	52
						34					significantly increased body burden compared to background.	

Environmental Science: Nano

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referen
SWCNT	1.5 μm	0.78 nm	None	NIRF	Added to water column of experimental wetland mesocosm	2.5 mg/L	10 months	Fish	Gambusia holbrooki		CNT length and diameter are product-specified. No bioaccumulation in aquatic vertebrates (fish) or plants was identified (below NIRF detection limits).	57
SWCNT	5 μm to 30 μm	1.1 nm	None	TEM	In aqueous solution (with SDS solvent)	0.1 mg/L to 0.5 mg/L	10 days	Fish	Oncorhynchus mykiss		Stress-induced drinking caused SWCNT ingestion and accumulation of CNT in gut tract. SWCNT's also precipitated on gill mucosa.	115
SWCNT	1.5 μm	0.8 nm	None	NIRF	Pelleted fish fish food amended with SWCNT in gum arabic solution	50 mg SWCNTs/kg food	96 h	Fish	Pimephales promelas		SWCNTs among the intestinal lumen contents but no apparent association with intestinal epithelia or underlying tissue.	116
SWCNT	1.5 μm	0.8 nm	None	NIRF	force fed SWCNT in gum arabic solution	Dosed at 0.01 ml of 426 mg/L SWCNT for 7 gavages	7 days	Fish	Pimephales promelas		NIRF images showed strong SWCNT-derived fluorescence signals in whole fish and excised intestinal tissues. Fluorescence was not detected in tissues other than intestines, indicating that no appreciable intestinal absorption occurred.	69
SWCNT	0.5 μm to 3 μm	1.4 nm	None	Raman microscopy	Suspended in Pluronic F-108	50 mg/L to 200 mg/L	5 days post fertilizatio n	Fish (embryo)	Danio rerio		Accumulation in embryos were exposure concentration- dependent.	42
SWCNT	0.5 μm to 3 μm	1.4 nm	Carboxylated	Raman microscopy	Suspended in Pluronic F-108	50 mg/L to 200 mg/L	5 days post fertilizatio n	Fish (embryo)	Danio rerio		Accumulation in embryos were exposure concentration- dependent.	42
SWCNT	ND	0.7 nm to 1.3 nm	acid-purified and carboxylated None	NIRF and carbon- 14 labeling	Dietary inclusion Sediment	10 mg SWNT/kg dry sediment, 10 mg SWNT/L	7 days	Marine amphipod	Ampelisca abdita		SWCNT detected in nondepurated amphipods exposed to amended food items (algae). SWCNT not detected in	52

.

Keterenc	Results	Factora	Species	Taxon	Duration	Conc.	Exposure	Detection method	Functionalization	Diameter	Length	Туре
Is in treatment nent alone.	amphipods in trea with sediment alc					(Cyclotella spp.)or 10 mg SWNT/kg brine Shrimp (Artemia spp)						
ulation of ⁶² oserved.	No accumulation SWCNT observed		Nereis virens	Marine polychaete	14 days	Worms exposed in 10 mg SWCNT/ kg dry sediment and 100 and 1000 mg SWCNT/ Kg SWCNT- amended prey	SWCNT suspension in gum arabic added to marine sediment. SWCNT-spiked algae fed to <i>Mercenaria</i> bivalve which was feds to the polychaete	NIRF	None	ND	ND	SWCNT
aken up into ⁸⁵ it itions between ind 24 mg/kg. tion to leaves s between 2 d 10 mg/kg.	SWCNTs taken up roots most concentrations be 0 mg/kg and 24 m Translocation to l and stems betwee mg/kg and 10 mg		Z. mays	Monocot (corn)	40 days	10 mg/kg, 100 mg/kg	In amended soil	TEM and microwave method	None	1 nm to 4 nm	5 μm to 30 μm	SWCNT
aken up into ⁸⁵ it itions between ind 24 mg/kg. tion to leaves s between 2 d 10 mg/kg. as not it on lization.	SWCNTs taken up roots most concentrations be 0 mg/kg and 24 m Translocation to l and stems betwee mg/kg and 10 mg Uptake was not dependent on functionalization.		Z. mays	Monocot (corn)	40 days	10 mg/kg, 100 mg/kg	In amended soil	TEM and microwave method	Hydroxyl- functionalized	1 nm to 4 nm	5 μm to 30 μm	SWCNT
aken up into ⁸⁵ it itions between ind 24 mg/kg. tion to leaves s between 2 d 10 mg/kg.	SWCNTs taken up roots most concentrations be 0 mg/kg and 24 m Translocation to l and stems betwee mg/kg and 10 mg		Z. mays	Monocot (corn)	40 days	10 mg/kg, 100 mg/kg	In amended soil	TEM and microwave method	Surfactant stabilized	1 nm to 4 nm	5 μm to 30 μm	SWCNT
its a a r is pi	SWCNTs roots mo concentra 0 mg/kg a Transloca and stem mg/kg ar		Z. mays	Monocot (corn)	40 days	10 mg/kg, 100 mg/kg 36	In amended soil	TEM and microwave method	Surfactant stabilized	1 nm to 4 nm	5 μm to 30 μm	SWCNT

Environmental Science: Nano

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Reference
SWCNT	ND	0.7 nm to 1.3 nm	None	NIRF and carbon- 14 labeling	Sediment	0.01 mg/kg	7 days	mysid	Americamysis bahia		SWCNT not detected in either depurated or nondepurated mysids	52
SWCNT	ND	1 nm to 2nm	HCl purified	Carbon-14 labeling	Sediment	30 mg/kg dry sediment	28 days	Oligochaete	Lumbriculus variegatus	0.28 ± 0.03 (BSAF)	Bioaccumulation factors an order of magnitude lower than PAH. Almost complete depuration after 3 days in CNT- free environment (sediment and/or water).	55
SWCNT	ND	1 nm to 2 nm	HCl purified	Carbon-14 labeling	Soil	30 mg/kg	14 days 100 mg/kg	Oligochaete	Eisenia foetida.	BAF: 0.0061 ± 0.002 (Chelsea soil) 0.022 ± 0.003 (Ypsilanti soil) 0.0078 ± 0.005 (Chelsea soil)	Low levels of uptake; most nanotubes in soil mass remaining in the worm's gut after depuration.	54
SWCNT	ND	ND	Acid-purified	Fluorescence spectroscopy	Sediment	50 mg/g and 250 mg/g	7 days	Oligochaete	Lumbriculus variegatus	0.0021 ± 0.0011 (BAF)	Detected the presence of labeled carbon nanotubes in worms exposed for 1 week to CNT-laden sediment.	117
SWCNT	5 μm to 15 μm	2 nm	None/Acid- purified	ТЕМ	Water exposure but sand was provided as a substrate	(1 x 10 ³) mg/L	14 days	Oligochaete	Lumbriculus variegatus		Qualitative determination of the presence of CNTs in the gut as well as on the outer surface of the test organisms, although no evidence of penetration of CNTs through cell membranes	102
SWCNT	ND	1.2 nm	Phospholipid (lysophophatidyl- choline or LPC)	Micro-Raman	Waterborne exposure	2.5 mg/L	96 h	Planktonic crustacea	Daphnia magna		Qualitative demonstration that <i>D.</i> <i>magna</i> were able to ingest solubilized LPC- SWCNTs and egest precipitated SWCNTs	118

⊿0

Туре	Length	Diameter	Functionalization	Detection method	Exposure	Conc.	Duration	Taxon	Species	Factora	Results	Referenc
SWCNT	500 nm to 1.5 μm	4 nm to 5 nm	Carboxylated SWCNT	Carbon-14 labeling	Marine sediment	(3.64 x 10 ²) mg/kg	14 days	Polychaete	Streblospio benedicti		No detectable bioaccumulation after depuration.	44
SWCNT	0.5 μm to 2.0 μm	1 nm to 2 nm	SWCNT dispersed with surfactant	coherent anti- Stokes Raman (CARS) scattering microscopy	Sediment exposure	3 mg/kg to 30 mg/kg	10 days	Polychaete	Arenicola marina		Qualitative; bioaccumulation in worm exposed to SWCNT- amended sediment is minimal	119
SWCNT	2 nm to 10 nm	< 500 nm	Acid oxidized SWCNT	Atomic force microscopy and SEM	In Osterhout's medium.	0 mg/L to 0.0172 mg/L	72 h	Protozoa	Tetrahymena thermophila		SWCNT internalization and subsequent egestion were observed.	120
^d This dat	r Factor (T a was pro	F)= CNT co vided in a la	ntent in leaves/Cl	NT content in exp	osure suspensi	on.						
						38						

Parameter	Daphnids	Soil invertebrates	Protozoans	Drosophila	Benthic and	Fish	Amphihians
ululleter	Dupinius	Oligochaetes	/Ciliates	Drosopinia	sediment-dwelling (aquatic and marine) invertebrates		, unpriloidit
≀ange (BCF, BAF, >r BSAF)	SWCNT: Not est. MWCNT: <i>D.</i> <i>magna</i> , 360000 ± 40000, 440000 ± 190000, and 350000 ± 80000 (BCF) ⁵³ , 6000-46,000 ⁶¹	SWCNT: <i>E. foetida</i> , 0.0061 to 0.022 (BAF) ⁵⁴ MWCNT: <i>E. foetida</i> 0.014 to 0.023 (BAF average 0.02± 0.006) ⁵⁴ ; 0.03 ⁵⁶ , 0.015 ± 0.004 ⁷⁰ (BAF)	SWCNT: Not est. MWCNT: Not est.	SWCNT: Not est. MWCNT: Not est.	SWCNT: L. plumulosus, 0.013 \pm 0.002 to 0.068 \pm 0.016 (nondepurated), 0.0040 \pm 0.0008 to 0.0074 \pm 0.0012 (depurated, BAF ⁵² ; L. Variegatus, 0.0021 \pm 0.0011 (BAF) ¹¹⁷ , 0.28 \pm 0.03(BSAF) ⁴⁸ , MWCNT: L. variegatus, 0.39 (\pm 0.08 to 0.67 (\pm 0.026) ^{55, 75} .	SWCNT: Not est. MWCNT: <i>D.</i> <i>rerio</i> , 73 (BCF dry mass), 16 (BCF wet mass) ⁶⁰	SWCNT: No est. MWCNT: Not est.
ffect of unctionalization	SWCNT: No comparative studies MWCNT: Surface coating did not affect accumulation or elimination rates by <i>D. magna</i> ⁶¹ .	SWCNT: No comparative studies MWCNT: No increase in bioaccumulation with increased concentration of oxygen functional groups by <i>E.</i> <i>foetida</i> ; surface coating did not affect accumulation or elimination rates	SWCNT: No comparative studies. MWCNT: No comparative studies.	SWCNT: No comparative studies MWCNT: No comparative studies.	SWCNT: No differences between varied functionalized CNTs on <i>L. variegatus</i> reported ¹⁰² . MWCNT: No increase in bioaccumulation by <i>L. variegatus</i> with increased concentration of oxygen functional groups ⁷⁵ .	SWCNT: No comparative studies MWCNT: No studies.	SWCNT: No comparativ studies. MWCNT: N comparativ studies.
Absorption of	SWCNT: Absorption from	SWCNT: No studies	SWCNT:	SWCNT: Only a	SWCNT: Absorption	SWCNT: Present	SWCNT: No

10

Parameter	Daphnids	Soil invertebrates Oligochaetes	Protozoans /Ciliates	Drosophila	Benthic and sediment-dwelling (aquatic and marine) invertebrates	Fish	Amphibia
epithelial cells	gut tract to other tissue not detected by <i>D. magna</i> ¹¹⁸ . MWCNT: Ingested material by <i>D. magna</i> largely eliminated on depuration. Absorption from gut tract to other tissue not detected ^{61, 77} .	complete elimination during depuration ^{56, 70} .	and subsequent egestion were observed ¹²⁰ . MWCNT: Exclusive localization into the mitochondria of the cells ¹¹⁰ .	fraction of the quantity ingested became incorporated into organs of the larvae ¹⁰ . MWCNT: Not studied.	other tissues not shown; depurated <i>L, variegatus</i> worms had very little SWCNT in their tissue ^{55, 102, 117} . No accumulation found in <i>Mercenaria</i> <i>mercenaria</i> ³⁴ , accumulation in visceral, mantle and gill tissues in <i>Mytilus</i> <i>galloprovincialis</i> ⁶³ ; MWCNT: Almost complete elimination on depuration in <i>L.</i> <i>variegatus</i> after 72 h ^{55, 102} .	Pimephales promelas, no appreciable uptake through the intestinal epithelium ¹¹⁶ . MWCNT: Largely eliminated via the digestive tract with very little detected in the blood and muscle tissue ⁶⁰ .	DWCNT: Present in gut lumer Ambyston mexicanu and Xenoj laevis ^{101, 1}
SWCNT versus MWCNT	No comparative studies.	No differences found in accumulation behaviors between SWCNT and MWCNT for <i>E.</i> <i>foetida</i> ⁵⁴ .	No comparative studies.	Investigated MWCNT and SWCNT on <i>D.</i> <i>melanogaster</i> but no quantitative comparison made on uptake ¹⁰⁴ .	No absorption across the gut for either type of CNT in amphipod <i>H.</i> <i>azteca</i> and dipterid <i>C. dilutus</i> ¹⁰² . No differences found in accumulation behaviors between SWCNT and MWCNT for <i>L. variegatus</i> ⁵⁵ .	No comparative studies.	No comparat studies.

		Taxon					
Parameter	Daphnids	Soil invertebrates Oligochaetes	Protozoans /Ciliates	Drosophila	Benthic and sediment-dwelling (aquatic and marine) invertebrates	Fish	Amphibians
				41			