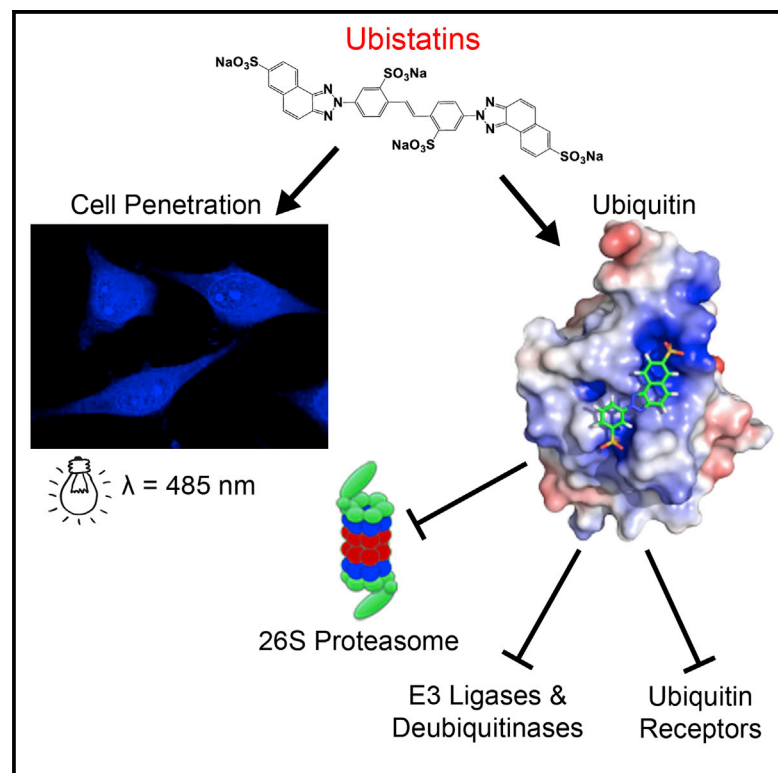


# Structure

## Structural Basis for the Inhibitory Effects of Ubistatins in the Ubiquitin-Proteasome Pathway

### Graphical Abstract



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### In Brief

Nakasone et al. characterize a panel of ubistatin derivatives and show that ubistatins inhibit ubiquitination and shield ubiquitin conjugates from disassembly by a range of deubiquitinases and by the 26S proteasome. Ubistatin B penetrates cancer cells and alters the cellular ubiquitin landscape. The structures of ubiquitin complexes with ubistatin B and hemi-ubistatin revealed that hydrophobic and charge/polar interactions are critical for ubistatin:ubiquitin binding.

### Highlights

- Ubistatin B binds Ub selectively and prefers K48-linked Ub chains over K11 or K63
- Hydrophobic and charge/polar interactions are critical for ubistatin:Ub binding
- Ubistatins block disassembly of Ub conjugates by various DUBs and by 26S proteasome
- Ubistatin B penetrates cancer cells and alters the cellular Ub landscape



# Structural Basis for the Inhibitory Effects of Ubistatins in the Ubiquitin-Proteasome Pathway

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<https://doi.org/10.1016/j.str.2017.10.007>

## SUMMARY

The discovery of ubistatins, small molecules that impair proteasomal degradation of proteins by directly binding to polyubiquitin, makes ubiquitin itself a potential therapeutic target. Although ubistatins have the potential for drug development and clinical applications, the lack of structural details of ubiquitin-ubistatin interactions has impeded their development. Here, we characterized a panel of new ubistatin derivatives using functional and binding assays. The structures of ubiquitin complexes with ubistatin B and hemi-ubistatin revealed direct interactions with ubiquitin's hydrophobic surface patch and the basic/polar residues surrounding it. Ubistatin B binds ubiquitin and diubiquitin tighter than a high-affinity ubiquitin receptor and shows strong preference for K48 linkages over K11 and K63. Furthermore, ubistatin B shields ubiquitin conjugates from disassembly by a range of deubiquitinases and by the 26S proteasome. Finally, ubistatin B penetrates cancer cells and alters the cellular ubiquitin landscape. These findings highlight versatile properties of ubistatins and have implications for their future development and use in targeting ubiquitin-signaling pathways.

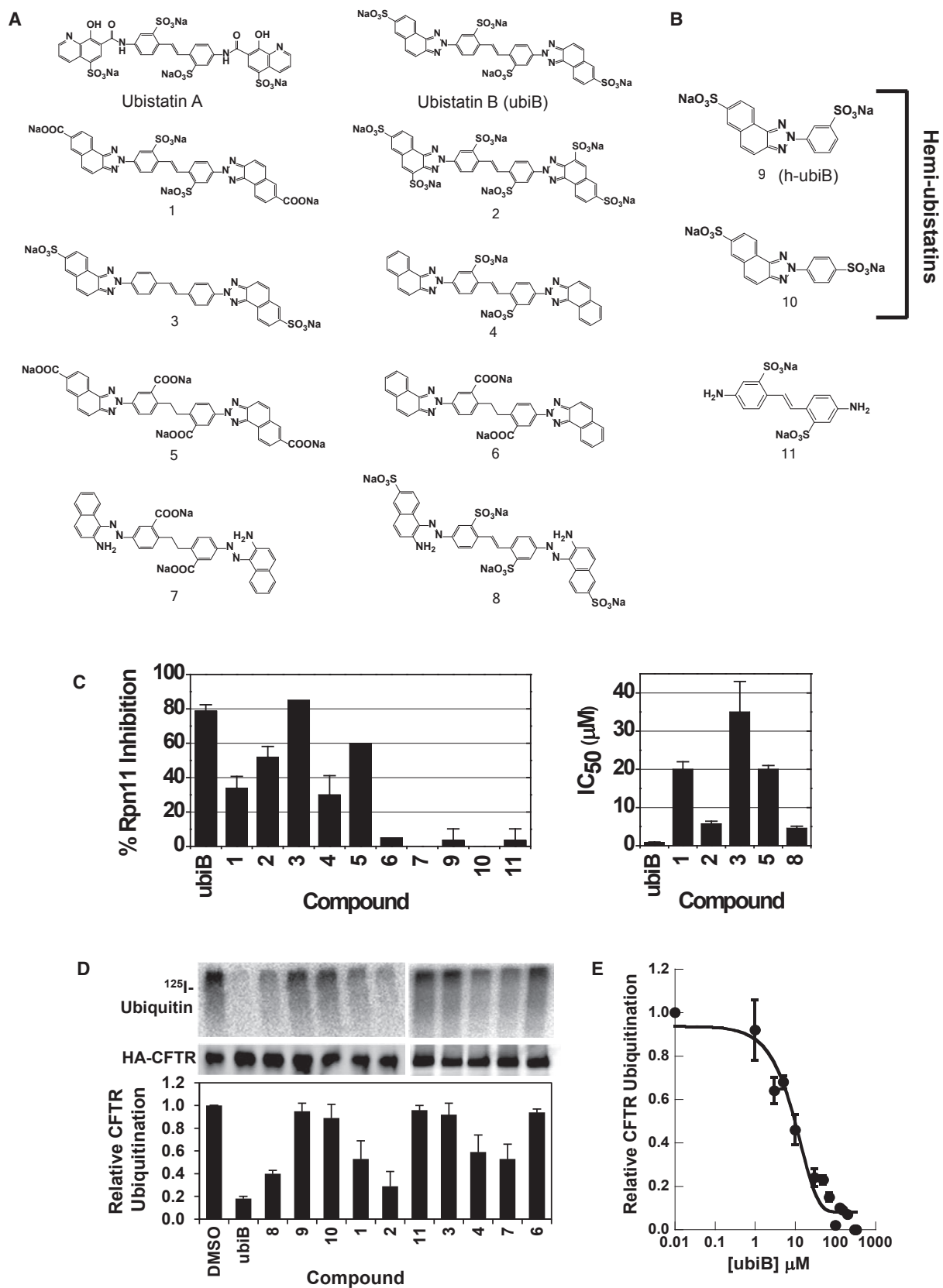
## INTRODUCTION

The ubiquitin-proteasome system (UPS) is the principal negative regulatory mechanism for short-lived proteins in eukaryotes (Hershko and Ciechanover, 1998). Proteins destined for degradation are post-translationally tagged with ubiquitin (Ub) or a chain of Ub molecules (polyUb) attached to a lysine side chain on a protein through the action of the E1, E2, E3 enzyme cascade

(Pickart, 1997). Provided the protein is appropriately modified with Ub, shuttling factors facilitate its transport to the 26S proteasome where dedicated receptors initiate its degradation (Finley et al., 2012; Pickart, 1997).

In eukaryotic organisms deficiencies in components of the UPS—such as E3 Ub ligases, essential subunits of the 26S proteasome, and even mutations in the Ub molecule itself—have severe consequences, including cell death and the onset of disease (Finley et al., 2012; Pickart and VanDemark, 2000; Roscoe et al., 2013). There is currently an intensive effort underway to develop therapeutics for targets upstream of the proteasome, including ubiquitin-selective chaperone p97/VCP, E3 ubiquitin ligases and their regulators, and deubiquitinases (DUBs) (Deshaies, 2014; Skaar et al., 2014). Nevertheless, clinically approved E3 ligase and DUB inhibitors are lacking. Thus far, peptide-based proteasome inhibitors (such as bortezomib and carfilzomib), which bind to the proteolytic  $\beta$  subunits in the 20S core, are the only clinically approved therapeutics targeting the UPS (Cvek, 2012). Because malignancies such as multiple myeloma require enhanced UPS function, these cells are particularly susceptible to proteasome inhibitors (Goldberg, 2012). However, through mechanisms that are not clear, some patients relapse and become refractory to proteasome inhibitors (Ruschak et al., 2011). Unlike the enzymatic components of the UPS (proteasomal subunits, E3 Ub ligases, DUBs), Ub is significantly less likely to tolerate mutations and therefore is potentially a more robust target for therapeutics.

Ubistatins are a family of small molecules that inhibit degradation of polyubiquitinated proteins by the UPS (Verma et al., 2004). By directly binding to polyUb, ubistatins act upstream of the proteasome, preventing recognition by downstream polyUb receptors. Specifically, ubistatins A and B outcompete the proteasomal polyUb receptor Rpn10 and the shuttle protein Rad23 for binding to polyUb chains. Direct binding of ubistatin A to K48-linked di-Ub (K48-Ub<sub>2</sub>) was detected and mapped to hydrophobic patch residues (L8, I44, V70) on the surface of Ub (Verma et al., 2004). A subsequent study examined how



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receptor:Ub interactions were altered by ubistatin in spliceosome assembly assays (Bellare et al., 2008). Although it was established that ubistatin A binds polyUb, the actual mechanism of interactions of ubistatin A and other ubistatin variants with (poly) Ub remained unclear. The lack of high-resolution structural data for the ubistatin:Ub interaction was a significant impediment to our understanding of the mechanism of inhibition and to further development of ubistatins for potential clinical applications.

To this end, a panel of ubistatin B derivatives was synthesized and characterized using functional and nuclear magnetic resonance (NMR)-based binding assays. We discovered that the most active compounds contain strongly acidic groups. These efforts led us to focus on ubistatin B and its corresponding hemi-ubistatin. Unexpectedly, ubistatin B binds two Ub molecules simultaneously, while hemi-ubistatin binds Ub significantly more weakly and in a 1:1 stoichiometry. We next determined the structure of hemi-ubistatin:Ub complex by NMR, revealing important contributions from key charged groups in Ub. Combined with data from small-angle neutron scattering (SANS), we generated a model of the ubistatin B:Ub complex. Further, we demonstrated that ubistatins inhibit a range of deubiquitinases, including those embedded within the 26S proteasome. Finally, we found that ubistatin B penetrates cell membranes and alters the cellular ubiquitin landscape. These data provide a gateway for future development of more potent preclinical candidates to treat cancer cells addicted to an elevated requirement for the UPS.

## RESULTS

### Synthesis and Initial Functional Screening of Ubistatin Derivatives

Ubistatin B (ubiB) derivatives (**1–8**) were synthesized to assess the contribution of substituent groups to Ub binding (Figure 1A). External and internal sulfonic acid groups were either removed or replaced with a less acidic carboxylic acid group. To address the contribution of the two symmetric ring systems to Ub binding, we synthesized “hemi”-ubistatin compounds, **9** (h-ubiB) and **10**, containing only a single ring system, as well as compound **11** containing the central bis-sulfonic acid motif (Figure 1B).

Ubistatins A and B were shown to inhibit Ub-dependent substrate proteolysis and deubiquitination by 26S proteasomes (Verma et al., 2004). To examine if the isopeptidase activity of the proteasome is impaired, we screened the panel of ubistatin compounds for their ability to inhibit disassembly of a model polyubiquitinated substrate by purified mammalian proteasome. The results indicate that ubistatins can efficiently block DUB activity of proteasome-embedded Rpn11 (Figure 1C and Table S1). Of all the analogs, ubiB had the strongest effect, with a half

maximal inhibitory concentration ( $IC_{50}$ ) = 1.1  $\mu$ M, while hemi-ubistatins were ineffective at inhibiting Rpn11 activity.

Given that ubistatins block recognition of (poly)Ub by receptors, we set out to test whether ubistatins have any effect on ubiquitination. Toward this goal, we screened the ubistatin compounds for their ability to inhibit ubiquitination of the cystic fibrosis transmembrane conductance regulator (CFTR) *in vitro*. Previous work established that Ub conjugation is required to select this protein for ER associated degradation (ERAD), and even the wild-type form of CFTR is a robust substrate for the ERAD pathway (Cheng et al., 1990). Other work established that the acquisition of Ub in this *in vitro* assay mirrored the selection of ERAD substrates for degradation in yeast cells (Nakatsukasa et al., 2008). Under the conditions tested, hemi-ubistatins had a minimal effect on CFTR ubiquitination, while the full ubistatins, ubiB and **2**, allowed only  $18\% \pm 2\%$  and  $29\% \pm 13\%$  of the protein to be modified, respectively (Figure 1D and Table S2). Of all the compounds tested, ubiB was clearly the most efficient at inhibiting ubiquitination (Figure 1D), and a titration of ubiB revealed that it inhibited CFTR ubiquitination with  $IC_{50}$  of  $\approx 10$   $\mu$ M (Figure 1E). Based on the ability of ubistatins to interact directly with Ub (Verma et al., 2004), these results reflect the ability of the tested compounds to inhibit the cellular ubiquitination machinery, and suggest that different substituent groups alter the affinities for Ub.

Collectively, these assays expose inhibition of ubiquitination and DUB activity as inherent properties of ubistatins. Furthermore, the findings suggest that substrate degradation via the UPS can be inhibited either upstream by preventing initial ubiquitination, or at the end of the pathway by impairing DUB activity at the proteasome.

### Ubistatins Recognize a Common Surface on Ubiquitin

Next, we wished to gain insight into the mode of ubistatin binding to Ub. Select full ubistatins (ubiB, **1**, **2**, **3**, **4**, **5**) and hemi-ubistatins (h-ubiB and **10**) were titrated and screened for direct binding to  $^{15}$ N-enriched Ub by monitoring changes in  $^1$ H- $^{15}$ N NMR spectra. Ubistatins capable of binding Ub produced residue-specific chemical shift perturbations (CSPs) highlighting the binding surface in Ub and the residues mediating key contacts (Figures 2A–2F). The pattern of residue-specific CSPs for all ubistatins that exhibited binding was consistent with binding to Ub's hydrophobic patch centered on L8, I44, and V70 (Figures 2B, 2C, 2E, and 2F). Notably, although ubiB, **1**, **2**, **3**, **4**, and **5** all contain the same carbon/nitrogen backbone and only differ in their substituent groups, they exhibited varying capacities to bind Ub. Both ubiB and **2** possess internal and external sulfonates and produced CSPs in and around Ub's hydrophobic patch (Figures 2B and 2E), while **1** containing external

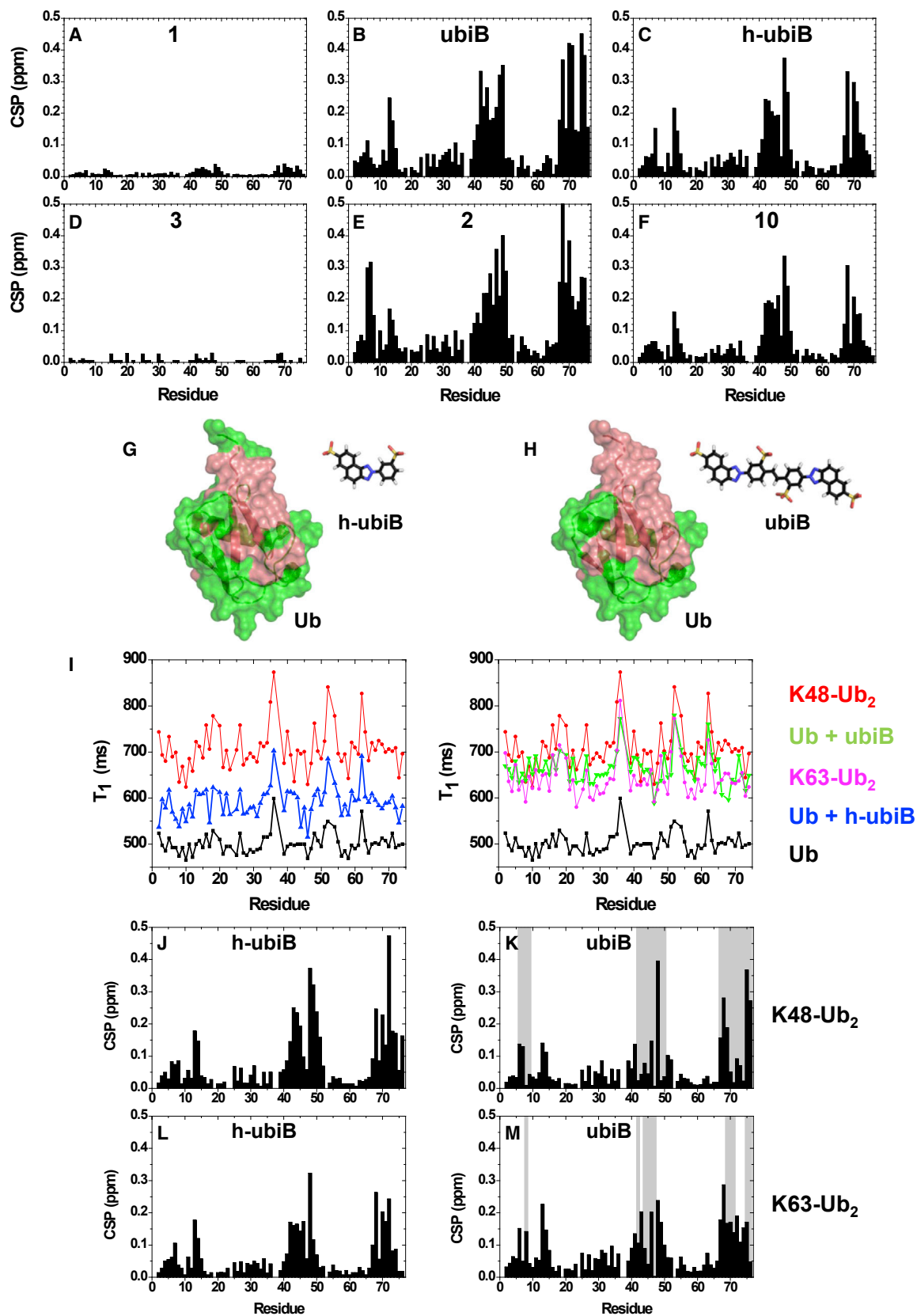
### Figure 1. Chemical Structures of Various Synthesized Ubistatin Derivatives and the Development of a Structure-Activity Relationship

(A and B) Full ubistatin compounds (A) and hemi-ubistatins (B).

(C) The effect of ubistatin variants on the DUB activity of Rpn11 in purified proteasome: percent inhibition measured at 10  $\mu$ M (left) and  $IC_{50}$  (right). Data represent the mean and SD (error bars) determined from multiple measurements. See also Table S1.

(D) *In vitro* CFTR ubiquitination reactions in the presence of the indicated ubistatin compounds. Ubiquitinated CFTR (top) is quantified relative to the amount of HA-tagged CFTR protein present in the immunoprecipitate (middle). The results are tabulated in the graph (bottom) and Table S2.

(E) Dose-dependent effect of ubistatin B (ubiB) on CFTR ubiquitination. *In vitro* CFTR ubiquitination reactions were performed as in (D) with the final concentrations of ubiB as shown. The reactions with 0.01  $\mu$ M ubiB were set to 100%, and the extent of CFTR ubiquitination was equivalent to that of a DMSO control (not shown). Data represent the mean and SD (error bars) of at least three determinations.



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**Table 1.  $^{15}\text{N}$   $T_1$  Values (in ms) for Ub and Ub Chains Free and in the Presence of Ubistatins**

	Ub	Ub		Ub <sup>72</sup>	Ub <sup>R74A</sup>	Ub <sup>R74E</sup>	Ub <sup>R42E,R72E</sup>	K48-	K48-	K48-	K63-	K63-	
Ub	+ h-ubiB	+ ubiB	Ub <sup>72</sup>	+ ubiB	+ ubiB	+ ubiB	+ ubiB	Ub <sub>2</sub>	Ub <sub>2</sub>	Ub <sub>2</sub>	Ub <sub>2</sub>	Ub <sub>2</sub>	Ub <sub>4</sub>
497 (14)	590 (21)	660 (24)	446 (19)	482 (28)	544 (26)	556 (28)	493 (27)	701 (28)	810 (34)	942 (66)	632 (24)	848 (31)	1,248 (86)
												1,248 <sup>a</sup> (63)	

Shown are mean values and SDs (in parentheses) averaged over residues in structured regions of Ub. See also Figures 2, S4, S6, and S8.

<sup>a</sup>Data taken from (Varadan et al., 2005).

carboxylates and internal sulfonates showed essentially no binding to Ub (Figure 2D). Titration with **3**, which contains only external sulfonates also revealed a lack of binding (Figure 2A). The inability to bind Ub was also observed for **4**, which contains only internal sulfonates, as well as for **5**, which contains both internal and external carboxylates. These results are in strong agreement with both the ubiquitination and proteasome DUB assays and suggest that critical substituent groups, namely acidic sulfonic acid groups, are a major factor that governs Ub binding.

Unexpectedly, two hemi-ubistatin variants of ubiB, h-ubiB and **10**, which were inactive in the functional assays, each produced large residue-specific CSPs (Figures 2C and 2F). We mapped above average CSPs caused by ubiB and h-ubiB to a nearly identical surface on Ub (Figures 2G and 2H). The ability of h-ubiB to bind Ub in the same manner as ubiB suggests that h-ubiB, i.e., half of a full ubistatin B, presents an independent Ub-binding element. To exclude the possibility that the observed CSPs reflect a general property of Ub's hydrophobic patch to form non-specific interactions with small hydrophobic molecules, Ub titration with ANS (8-anilino-1-naphthalene-sulfonic acid) resulted in only small CSPs (<0.1 ppm, not shown), supporting the conclusion that hemi-ubistatins represent a new class of specific Ub-binding molecules.

Throughout titration with ubiB, Ub spectra exhibited a global decrease in signal intensity (not shown), reflecting an increase in size (slower tumbling) as a consequence of complex formation. This prompted us to address the stoichiometry of h-ubiB and ubiB complexes with Ub. We detected a noticeable increase, from  $497 \pm 14$  ms to  $660 \pm 25$  ms (Table 1), in the  $^{15}\text{N}$  longitudinal relaxation time ( $T_1$ ) of Ub upon binding to ubiB, indicating a significant size increase of the tumbling moiety (Figure 2I, right panel). The  $^{15}\text{N}$   $T_1$  values for Ub in the presence of ubiB were between those for K48-Ub<sub>2</sub> and K63-Ub<sub>2</sub>, indicating that two Ubs were bound to ubiB simultaneously. By contrast, h-ubiB binding resulted in lower Ub  $^{15}\text{N}$   $T_1$  values ( $590 \pm 21$  ms; Figure 2I, left panel) which fall between those for free Ub and K63-Ub<sub>2</sub>, suggesting the presence of 1:1 and 2:1 Ub:h-ubiB complexes. These observations were further

corroborated by the dose-dependent decrease in the translational diffusion coefficient of Ub upon addition of ubiB (Figure S1).

That a single surface on Ub appeared to bind the ubistatins led us to examine whether the same surface mediated ubistatins' binding to polyUb. Applying the same methodology, we titrated h-ubiB and ubiB separately into solutions of K48-Ub<sub>2</sub> or K63-Ub<sub>2</sub> in which the distal Ub unit was  $^{15}\text{N}$  labeled. The patterns of residue-specific CSPs for all four titrations (Figures 2J–2M and S2A) strongly resembled those for monomeric Ub, indicating that both ubiB and h-ubiB employ a conserved binding mechanism for interaction with Ub and with polyUb. Notably, because the hydrophobic patches of the two Ub units in K48-Ub<sub>2</sub> are oriented entirely differently from those in K63-Ub<sub>2</sub> (Pickart and Fushman, 2004), our results suggest that both ubiB and h-ubiB recognize this binding surface regardless of how it is presented. Furthermore, this result corroborates the conclusion that hemi-ubistatin is the fundamental Ub-binding unit in the full ubistatin molecule. The greater number of residue-specific NMR signal attenuations caused by ubiB in K48-Ub<sub>2</sub> compared with K63-Ub<sub>2</sub> (Figures 2K and 2M) also suggests that ubiB has a greater affinity for K48 linkages, in agreement with previous observations (Verma et al., 2004). In addition, the lack of signal attenuations for both linkage types in the presence of h-ubiB (Figures 2J, 2L, and S2A) is in line with our conclusion from the functional *in vitro* assays (Figures 1C and 1D) that hemi-ubistatins have lower affinity for (poly)Ub than the full ubistatin. We also observed a significant increase in  $^{15}\text{N}$   $T_1$  values for both K48-Ub<sub>2</sub> and K63-Ub<sub>2</sub> upon addition of the ubistatins, and even a greater difference (judging by the  $^{15}\text{N}$   $T_1$  values) between h-ubiB and ubiB in the apparent size of the resulting complexes with Ub<sub>2</sub>s (Figure S2B).

### Structure of the Ubiquitin:Hemi-Ubistatin Complex

Our NMR titration data established that both full and hemi-ubistatins bind the hydrophobic patch in Ub regardless of Ub's polymeric state or linkage composition, provided that the correct substituent groups in the compound are present. We next proceeded to determine the structure for the Ub:h-ubiB

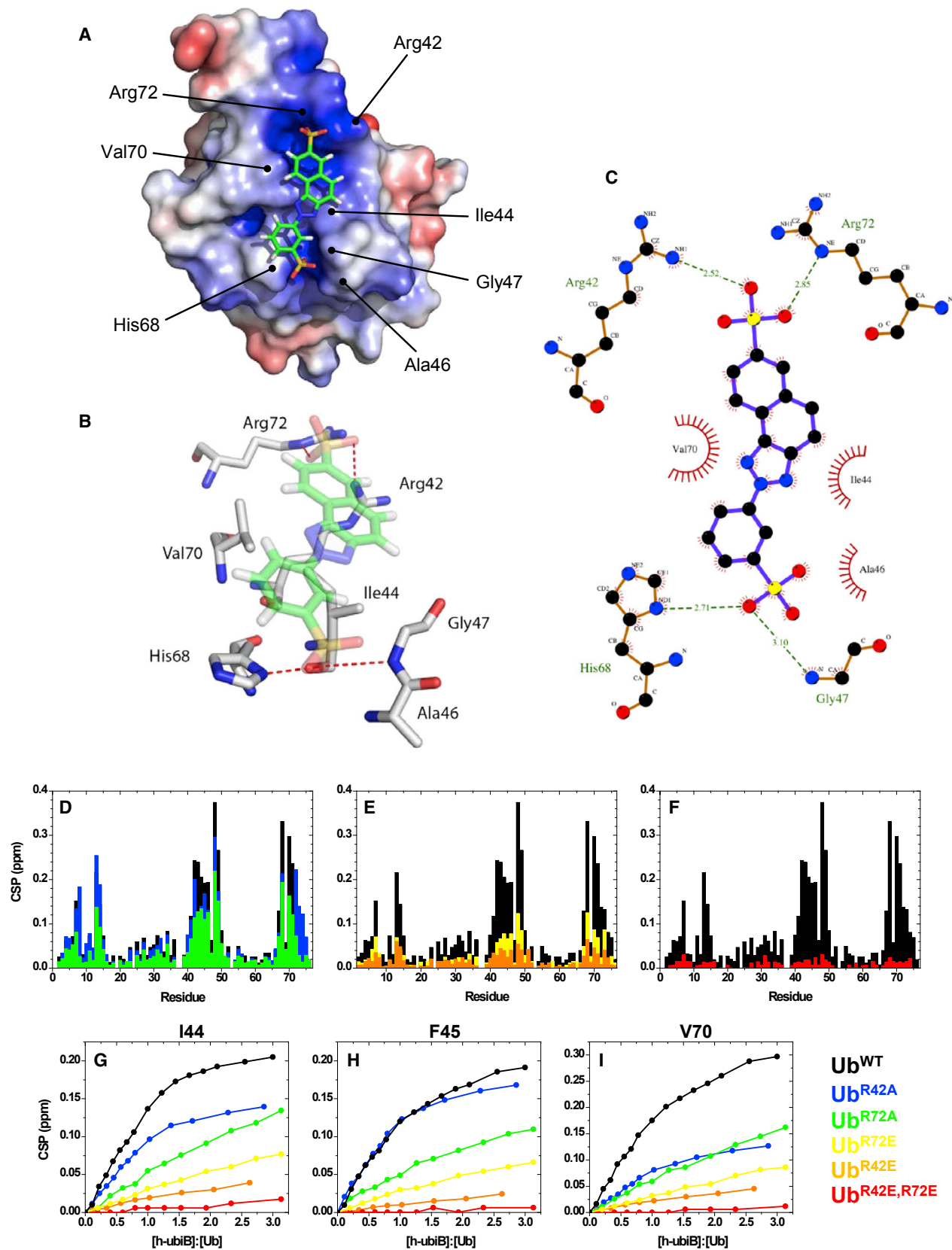
### Figure 2. Screening of Ubistatins for Direct Binding to Ub or Ub<sub>2</sub> Using NMR Titration Assays

(A–F) Residue-specific amide CSPs in Ub at the endpoint of titration with the indicated ubistatins ([ubistatin]:[Ub] = 2 except for **3** where [ubistatin]:[Ub] = 1). The starting Ub concentration was 1 mM. (A) **1**; (B) ubiB; (C) h-ubiB; (D) **3**; (E) **2**; (F) **10**.

(G and H) Residues exhibiting above average CSPs at the endpoint of titration with (G) h-ubiB or (H) ubiB are mapped (painted red) onto the surface of Ub (PDB: 1D3Z); these are residues with CSP >0.082 ppm in (G) and >0.11 ppm in (H).

(I)  $^{15}\text{N}$   $T_1$  values as a function of residue number in: free Ub (black), Ub at the endpoint in titration with h-ubiB (blue, left panel) or with ubiB (green, right), as well as for distal Ub in unbound K48-Ub<sub>2</sub> (red) and K63-Ub<sub>2</sub> (magenta, right).

(J–M) NMR mapping of the interactions of ubiB or h-ubiB with  $^{15}\text{N}$ -labeled distal Ub in K48-Ub<sub>2</sub> or K63-Ub<sub>2</sub>. Residue-specific perturbations in  $^1\text{H}$ - $^{15}\text{N}$  NMR spectra of (J) K48-Ub<sub>2</sub> with h-ubiB, (K) K48-Ub<sub>2</sub> with ubiB, (L) K63-Ub<sub>2</sub> with h-ubiB, and (M) K63-Ub<sub>2</sub> with ubiB. The titration continued until the molar ratio [ubistatin]:[Ub<sub>2</sub>] ≈ 4. Black bars represent the magnitude of amide CSPs at the endpoint of titration, while gray bars indicate residues exhibiting strong signal attenuations reflecting severe line broadening caused by intermediate or slow exchange. See also Figures S1 and S2.



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complex, in order to reveal structural details of these interactions.

NOESY spectra showed strong cross-peaks between h-ubiB protons and methyl protons of Ub (Figure S3A), which unambiguously positioned h-ubiB in direct contact with I44 and V70 of Ub (Figure S3B). However, these short-range (<5 Å) distances were insufficient to distinguish between multiple orientations of h-ubiB on the Ub surface. Hence, we employed site-directed paramagnetic labeling with nitroxide spin label, MTSL, to obtain long-range distance restraints. The optimal sites for MTSL attachment to Ub were chosen by simulating the paramagnetic effect on each proton in h-ubiB caused by placing MTSL on any residue in Ub, based on nuclear Overhauser effect (NOE)-derived models of the Ub:h-ubiB complex (Figure S4A). With this insight, we carried out several experiments attaching MTSL to Ub residues 36, 48, or 63 (cysteine mutants Ub<sup>I36C</sup>, Ub<sup>K48C</sup>, or Ub<sup>K63C</sup>, respectively). MTSL-labeled Ub<sup>I36C</sup> and Ub<sup>K48C</sup> yielded long-range distance restraints subsequently used for structure calculations (Table S3). As a negative control, Ub<sup>K63C</sup>, which positions MTSL far from h-ubiB resulted in a negligible paramagnetic effect (not shown).

The final structure of the Ub:h-ubiB complex (Figures 3A and 3B) was obtained using the HADDOCK program (de Vries et al., 2010); the statistics are in Table S4 (see also Figures S4C–S4F). Consistent with intermolecular NOEs, the naphthotriazole moiety of h-ubiB exhibits several hydrophobic contacts with V70 and I44 in Ub (Figures 3B and 3C). A detailed analysis of the structure suggests that hydrophobic contacts are further stabilized by electrostatic and polar interactions between the sulfonate groups of h-ubiB and several residues in Ub surrounding the hydrophobic patch. Specifically, Ub residues R42 and R72 are in close proximity to one of the two sulfonates of h-ubiB while H68 and G47 are within hydrogen-bonding distance of the other (Figures 3B and 3C). These contacts are in excellent agreement with the large amide CSPs observed for these residues (Figure 2C).

### Electrostatic Interactions Facilitate Ubistatin-Ubiquitin Binding

To validate the contribution of the electrostatic interactions inferred from our Ub:h-ubiB structure, we systematically mutated cationic residues R42 and R72 one-by-one to neutral (Ala) or oppositely charged (Glu) residues. The electrostatic surface potential of each mutant suggests that the ubistatin binding site undergoes a dramatic change in polarity (Figure S5A). We assessed the ability of h-ubiB to bind several single-site mutants: Ub<sup>R42A</sup>, Ub<sup>R42E</sup>, Ub<sup>R72A</sup>, Ub<sup>R72E</sup>, as well as the double mutant Ub<sup>R42E,R72E</sup> using the same NMR assay as for Ub<sup>WT</sup> (Figures 3D–3F). The results clearly show that the charge state at residues

42 and 72 has a dramatic impact on the ability of Ub to bind h-ubiB (Figures 3D–3F). By contrast, changing the bulk charge of Ub with the K63D mutation had little effect on h-ubiB binding (Figures S5B and S5C), verifying that the observed effect of the R42 and R72 mutations was due to changes in specific electrostatic/polar interactions between Ub and h-ubiB. The spectral similarity to Ub<sup>WT</sup> (Figures S5D and S5E) confirmed that the structure of Ub remained intact for the mutants analyzed.

The impact each Ub mutation has on the strength of the h-ubiB interaction is best illustrated by comparing the magnitudes of the residue-specific CSPs at the titration endpoint with those for Ub<sup>WT</sup>, as well as residue-specific titration curves (Figures 3G–3I). Notably, both single-site “neutral” mutants (Ub<sup>R42A</sup> and Ub<sup>R72A</sup>) exhibited only a slight decrease in the CSPs compared with Ub<sup>WT</sup> (Figure 3D), suggesting that the remaining interactions (including the hydrophobic effect and the electrostatic attraction to the remaining arginine) are sufficient for binding. Single-site “negative” mutations, Ub<sup>R42E</sup> and Ub<sup>R72E</sup>, significantly reduced the binding, emphasizing the role of the electrostatic contacts and demonstrating that the hydrophobic interaction between Ub and h-ubiB can be overpowered by the point-charge repulsion (Figure 3E). To underscore this point, the double mutant, Ub<sup>R42E,R72E</sup>, showed virtually no detectable CSPs in the presence of h-ubiB, indicating complete abolition of binding (Figure 3F). Following our observation that h-ubiB and ubiB bind to the same surface on Ub, we also tested whether ubiB binds Ub<sup>R42E,R72E</sup>. In line with the h-ubiB results, ubiB produced only negligible CSPs when titrated into Ub<sup>R42E,R72E</sup> and no detectable increase in the <sup>15</sup>N T<sub>1</sub> (Figures S5G and S5H, Table 1). These results clearly demonstrate that R42 and R72 are required for ubistatin binding. Furthermore, the importance of the external sulfonic acid groups in ubistatins for efficient Ub binding was corroborated by observations that compounds **4** (lacking external sulfonates) and **1** (external sulfonates replaced with carboxylates) failed to bind Ub (Figure 2D).

To explore the selectivity of ubistatins for Ub, we titrated h-ubiB into yeast Rub1 (ortholog of human Nedd8), a Ub-like (UBL) protein highly homologous to Ub. Like Ub, Rub1 contains the L8, I44, V70 hydrophobic surface patch surrounded by basic residues (K6, R42, R74), although R72 is replaced by a threonine (Singh et al., 2012). Interestingly, only minimal CSPs were observed in Rub1, mostly clustered around I44 and V70 (Figure S6). This suggests that residues beyond the hydrophobic patch are major contributors to ubistatin binding, rendering h-ubiB highly specific for Ub.

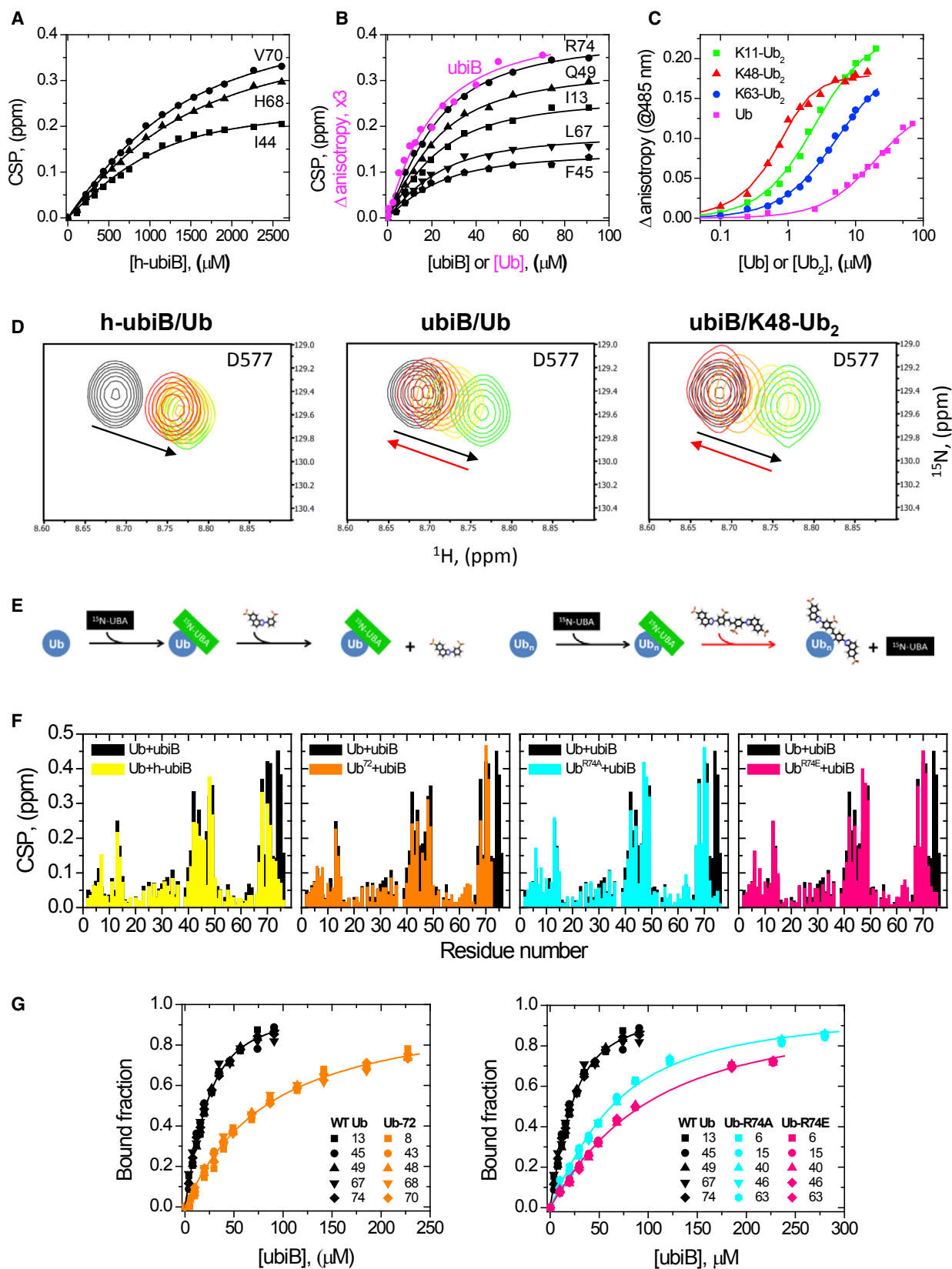
### The Hemi- and Full Ubistatins Differ in Binding Affinity

We next set out to characterize the binding affinities of h-ubiB and ubiB. The NMR titration curves of <sup>15</sup>N-labeled Ub with

**Figure 3. Structure of the Ub:h-ubiB Complex**

(A) Ub is represented by its electrostatic surface (equipotential contours from −3kT/e to +3kT/e, positive potential is painted blue and negative is red), while h-ubiB is shown in stick representation. Key Ub residues involved in the interaction are labeled.  
(B) 3D representation of the contacts between h-ubiB and the interacting groups in Ub. Red dashed lines represent hydrogen bonds.  
(C) LIGPLOT representation of the hydrophobic contacts (red spoked arcs) and hydrogen bonds (green dashed lines) between h-ubiB and Ub.  
(D–I) Validation of the structure of Ub:h-ubiB complex by site-directed mutagenesis. Residue-specific CSPs from titration with h-ubiB of Ub<sup>WT</sup> (black) and (D) single-Ala mutants, Ub<sup>R42A</sup> (green) and Ub<sup>R72A</sup> (blue), (E) single-Glu mutants, Ub<sup>R42E</sup> (yellow) and Ub<sup>R72E</sup> (orange), and (F) double-Glu mutant, Ub<sup>R42E,R72E</sup>, red. Representative titration curves for Ub residues (G) I44, (H) F45, and (I) V70, colored according to the indicated mutants.  
See also Figures S4–S6 and Tables S3 and S4.





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**Table 2. Dissociation Constants (in  $\mu\text{M}$ ) for h-ubiB and ubiB Binding to Ub, Ub<sub>2</sub>, and Ub Mutants**

Ub + h-ubiB	Ub + ubiB	Ub + ubiB*	Ub <sup>72</sup> + ubiB	Ub <sup>R74A</sup> + ubiB	Ub <sup>R74E</sup> + ubiB	Ub <sup>R42E</sup> + ubiB	Ub <sup>R42E,R72E</sup> + ubiB	K48-Ub <sub>2</sub> + ubiB*	K11-Ub <sub>2</sub> + ubiB*	K63-Ub <sub>2</sub> + ubiB*
595 (200)	11.35 (2.2)	14.0* (1.6)	73.1 (7.3)	40.3 (9.1)	67.6 (15.0)	396 (77)	>1,000	0.264* (0.023)	1.83* (0.05)	4.88* (0.22)

The numbers in parentheses represent SDs. Data marked with an asterisk were obtained from ubiB fluorescence anisotropy measurements; the rest of the  $K_d$  values were obtained from NMR titrations of <sup>15</sup>N-labeled Ub or Ub<sub>2</sub> with ubiB or h-ubiB. See also Figures 4 and S7.

h-ubiB exhibited a characteristic hyperbolic shape (Figure 4A), but did not reach saturation even nearing a 3-fold molar excess of h-ubiB ([h-ubiB] = 2.6 mM). From these data we obtained a  $K_d$  of  $595 \pm 200 \mu\text{M}$  (averaged over 10 residues). By contrast, titration curves for ubiB at the same conditions were consistent with “stoichiometric” binding (not shown), indicating that the  $K_d$  was significantly below the concentration of Ub (0.9–1 mM). With this in mind, we performed ubiB titrations starting with a Ub concentration of  $20 \mu\text{M}$  (Figures 4B and S7B), which yielded a  $K_d$  value of  $11.4 \pm 2.2 \mu\text{M}$  (Table 2). To verify this result independently, we took advantage of the native fluorescence of ubiB and measured the change in fluorescence anisotropy upon titration with Ub (Figure 4B). The resulting  $K_d$  =  $14.0 \pm 1.6 \mu\text{M}$  was in agreement with our NMR data.

Finally, to quantify the strength of ubiB binding to di-Ub chains, we measured the change in fluorescence anisotropy of ubiB upon titration with K11-, K48-, or K63-linked Ub<sub>2</sub> (Figure 4C). K48-Ub<sub>2</sub> exhibited the tightest binding ( $K_d$  =  $264 \pm 23 \text{ nM}$ ), supporting previous reports (Verma et al., 2004); however, other Ub<sub>2</sub>s also bound ubiB quite tightly, with  $K_d$  values of  $1.83 \pm 0.05 \mu\text{M}$  (K11-Ub<sub>2</sub>) and  $4.88 \pm 0.22 \mu\text{M}$  (K63-Ub<sub>2</sub>).

To further verify our results, we devised a direct NMR competition assay (Figures 4D and 4E), which tested the ability of ubiB and h-ubiB to compete for Ub binding against the Ub-associated (UBA) domain of ubiquilin-1, one of the strongest Ub binders ( $K_d$  ~  $20 \mu\text{M}$ ) among known Ub-binding domains (Raasi et al., 2005; Zhang et al., 2008). In this assay, unlabeled Ub was added to  $200 \mu\text{M}$  <sup>15</sup>N-labeled UBA at a 1:1 molar ratio, causing the UBA signals to shift to the Ub-bound state. Then, either h-ubiB or ubiB was gradually added. The addition of ubiB caused the UBA signals to return to their positions in the unbound state (Figure 4D, center), indicating that ubiB outcompetes UBA for binding to Ub. Furthermore, ubiB outcompeted UBA for binding to K48-Ub<sub>2</sub> (Figure 4D, right). By contrast, h-ubiB was unable to outcompete UBA, even at 3-fold molar excess (Figure 4D, left).

These results indicate that the relative affinity of ubiB for Ub and K48-linked polyUb is greater than that of the UBA (Figure 4E), while the affinity of h-ubiB is drastically lower, in agreement with our NMR and fluorescence titration data (Table 2).

Several reasons may account for the stronger binding of Ub to ubiB compared with h-ubiB. First, ubiB contains twice the number of negatively charged ( $\text{SO}_3^-$ ) groups, which could strengthen the interaction with the positively charged residues surrounding the hydrophobic patch of Ub. Second, ubiB presents a larger contiguous Ub-binding surface spanning two h-ubiB molecules, which doubles the effective local concentration of the Ub-binding sites (ubiB versus h-ubiB) and provides an increased likelihood for Ub to re-associate with the same ubiB molecule after dissociation. Furthermore, the extended binding surface could facilitate bidentate and possibly avid binding of the two halves of ubiB to Ub or polyUb. Last but not least, Ub:ubiB binding could be strengthened by additional interatomic contacts that are not present or weaker in the Ub:h-ubiB complex. It is also possible that the resulting ubiB:2xUb complex is further strengthened by the interactions between the two Ubs bound to the same ubiB molecule.

To shed light on possible differences in Ub's contacts with h-ubiB and ubiB, we compared shifts in the NMR signals of each residue in Ub upon titration with these two compounds. A detailed analysis revealed that despite strong similarity of the magnitudes and the directions of the signal shifts caused by ubiB and h-ubiB for most of the Ub residues, there is a striking difference in the behavior of the C-terminal residues (Figures 4F, S7A, and S7B). Specifically, R74, G75, and even G76 exhibited substantial CSPs upon addition of ubiB, whereas their signals barely shifted upon h-ubiB binding. To examine possible contributions of these residues to stronger ubiB binding, we deleted the C-terminal tail of Ub (residues 73–76); this variant is referred to as Ub<sup>72</sup>. This resulted in a 6-fold decrease in the binding affinity ( $K_d$  =  $73.1 \mu\text{M}$ , Table 2). Furthermore, there was

#### Figure 4. Titration Analysis of Ubistatin Binding to Ub and Differently Linked Ub<sub>2</sub> Molecules

(A and B) Representative NMR binding curves for the indicated residues in Ub upon titration with (A) h-ubiB and (B) ubiB. The lines represent the results of a global fit to a 1:1 binding model. Shown in (B) in magenta, for comparison, are fluorescence data for a reverse titration of ubiB with Ub (from C).

(C) Change in ubiB fluorescence anisotropy, monitored at 485 nm, upon titration with Ub (magenta), K11-Ub<sub>2</sub> (green), K48-Ub<sub>2</sub> (red), or K63-Ub<sub>2</sub> (blue). The lines represent the results of fit to a 1:1 binding model.

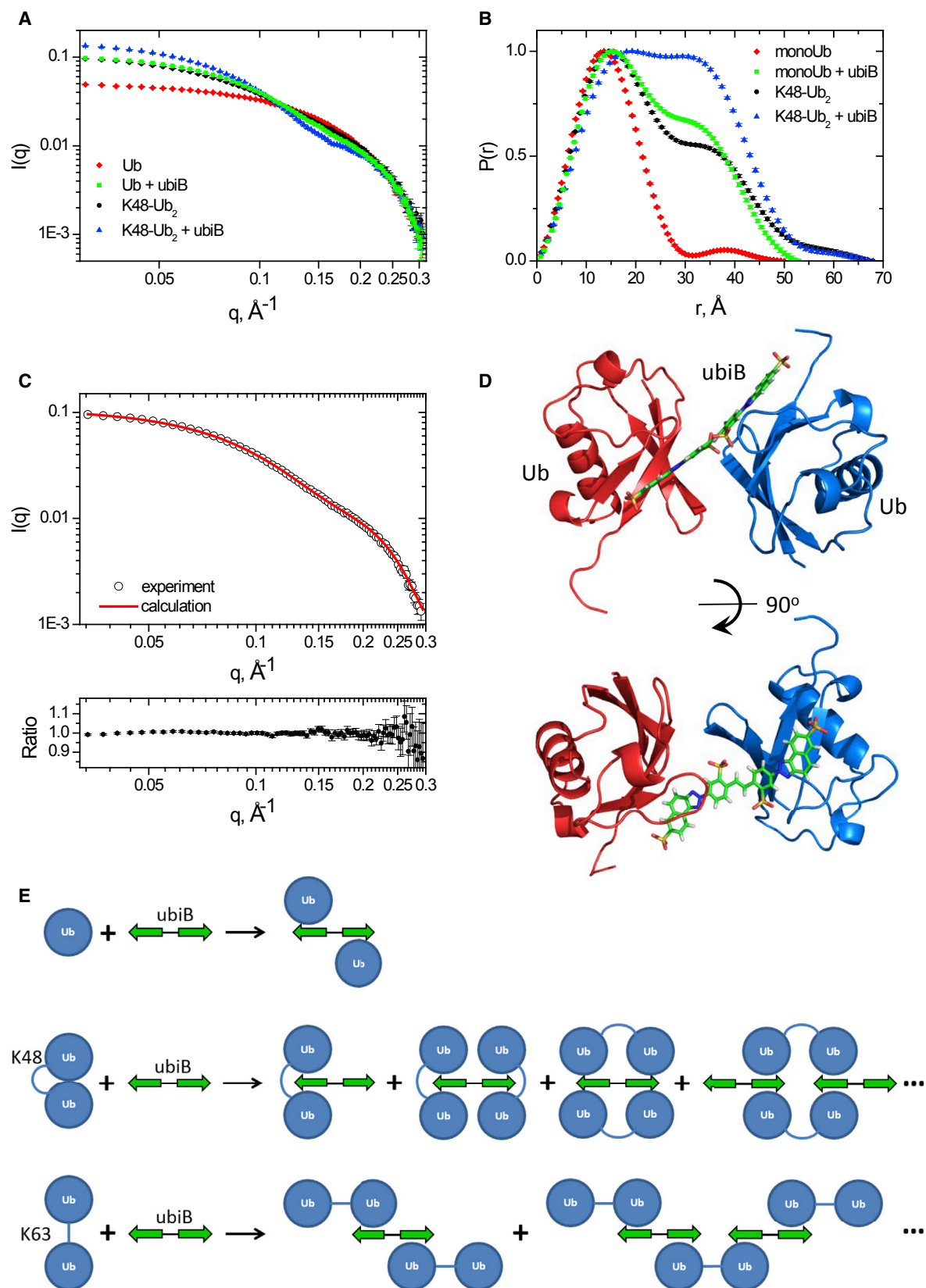
(D) Competition between h-ubiB or ubiB against ubiquilin-1 UBA domain: (left) h-ubiB and Ub, (middle) ubiB and Ub, and (right) ubiB and K48-Ub<sub>2</sub>. <sup>1</sup>H-<sup>15</sup>N SOFAST-HMQC signals of UBA residue D577: unbound (black contours) and Ub- or Ub<sub>2</sub>-bound state (green). Signals upon increasing addition of ubistatin are colored yellow (1:1 molar ratio), orange (2:1), and red (3:1). Note that the UBA signals that shifted as a result of Ub or Ub<sub>2</sub> binding (see black arrow) return back to their unbound positions upon addition of ubiB (red arrow), indicating a displacement of UBA from Ub or Ub<sub>2</sub> by ubiB.

(E) Schematic illustration of the results of the competition assay.

(F) Comparison of the residue-specific CSPs in Ub at the endpoint of titration with ubiB and h-ubiB (left) or in wild-type (WT) Ub and the indicated Ub variants upon titration with ubiB.

(G) Comparison of the normalized NMR titration curves for select residues in WT Ub, Ub<sup>72</sup>, Ub<sup>R74A</sup>, and Ub<sup>R74E</sup>. The lines represent the results of global fit of the data to a 1:1 binding model (Table 2). Data coloring is the same as in (F).

See also Figure S7.



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**Table 3. Radius of Gyration ( $R_g$ , in Å) Derived from the SANS Data for Various Samples Studied**

Sample	Ub	Ub + ubiB	K48-Ub <sub>2</sub>	K48-Ub <sub>2</sub> + ubiB
$R_g$	11.3 (0.1)	17.4 (0.1)	18.4 (0.1)	20.1 (0.1)

The numbers in parentheses represent SDs.

only a minor increase in  $^{15}\text{N}$   $T_1$  of Ub<sup>72</sup> upon ubiB binding, suggesting a 1:1 stoichiometry of the resulting complex (Table 1). These results clearly indicate that Ub's C terminus plays a role in stronger ubiB binding and Ub dimerization. Notably, the directions of the signal shifts and the CSPs for the rest of the Ub residues remained essentially unchanged (Figures 4F and S7A–S7C), suggesting that the nature of the local interactions between Ub<sup>72</sup> and ubiB remained similar to Ub:h-ubiB interactions. Since Ub's C terminus contains a basic residue (R74) that may contribute to ubiB binding through electrostatic interactions, we examined the effect of R74 substitution with Ala (Ub<sup>R74A</sup>) or Glu (Ub<sup>R74E</sup>) (Figures S7D–S7G). These mutations weakened ubiB binding by 3.5- to 6-fold, respectively (Table 2, Figure 4G), and the  $^{15}\text{N}$   $T_1$  at the end of the titration was markedly lower than for Ub:ubiB complex and even slightly less than for Ub:h-ubiB complex (Table 1, Figure S7H). Notably, no CSPs were observed for the C-terminal residues (Figures 4F and S7F), and, as in the case of Ub<sup>72</sup>, the directions of the signal shifts and the CSPs for the rest of Ub residues were essentially the same as for h-ubiB:Ub binding (Figures S7D and S7E). Collectively, these results point to the role of Ub's C-terminal residues, especially R74, in strengthening ubiB:Ub binding and promoting the formation of the ternary ubiB:2xUb complex.

### Structural Modeling of Ubistatin B Interactions with Ub

Thus far, our data demonstrated that ubiB binds to Ub's hydrophobic patch and the resulting complex includes two Ub molecules. Furthermore, the strong similarity of both the magnitudes (CSPs, Figures 2B and 2C) and the directions of the NMR signal shifts (Figures S7A and S7B) upon Ub binding to ubiB and h-ubiB indicates similarity of the interatomic contacts between Ub and these compounds. In order to gain additional structural insights into these interactions, we performed SANS measurements for Ub and K48-Ub<sub>2</sub> in complex with ubiB. A drastic change in the SANS profiles of both Ub and K48-Ub<sub>2</sub> upon addition of ubiB (Figures 5A and 5B) indicates that the overall shape (hence structural arrangement) of these proteins in complex with ubiB is different from their free states. In fact, the analysis showed a significant increase in the radius of gyration ( $R_g$ ) for both Ub and Ub<sub>2</sub> (Table 3). Specifically, ubiB binding to Ub caused a 2-fold increase in  $I(q \approx 0)$  and resulted in an  $R_g$  value close to that of unbound K48-Ub<sub>2</sub>. These results clearly corroborate the stoichiometry of the ubiB:2xUb complex inferred from our NMR data.

Furthermore, the overall similarity between the SANS profiles, as well as the pair distribution functions for the Ub + ubiB and free K48-Ub<sub>2</sub> samples suggests that structural arrangement of the two Ubs in complex with ubiB resembles that of K48-Ub<sub>2</sub>.

Inspired by these observations, we built a model of the ubiB:2xUb complex based on a combination of the NMR and SANS data and using HADDOCK (de Vries et al., 2010) supplemented with molecular dynamics (MD) simulations. The generated structure of the complex that agrees best with the SANS data (Figure 5C) is shown in Figure 5D; the convergence of the top ten structures is illustrated in Figure S8A.

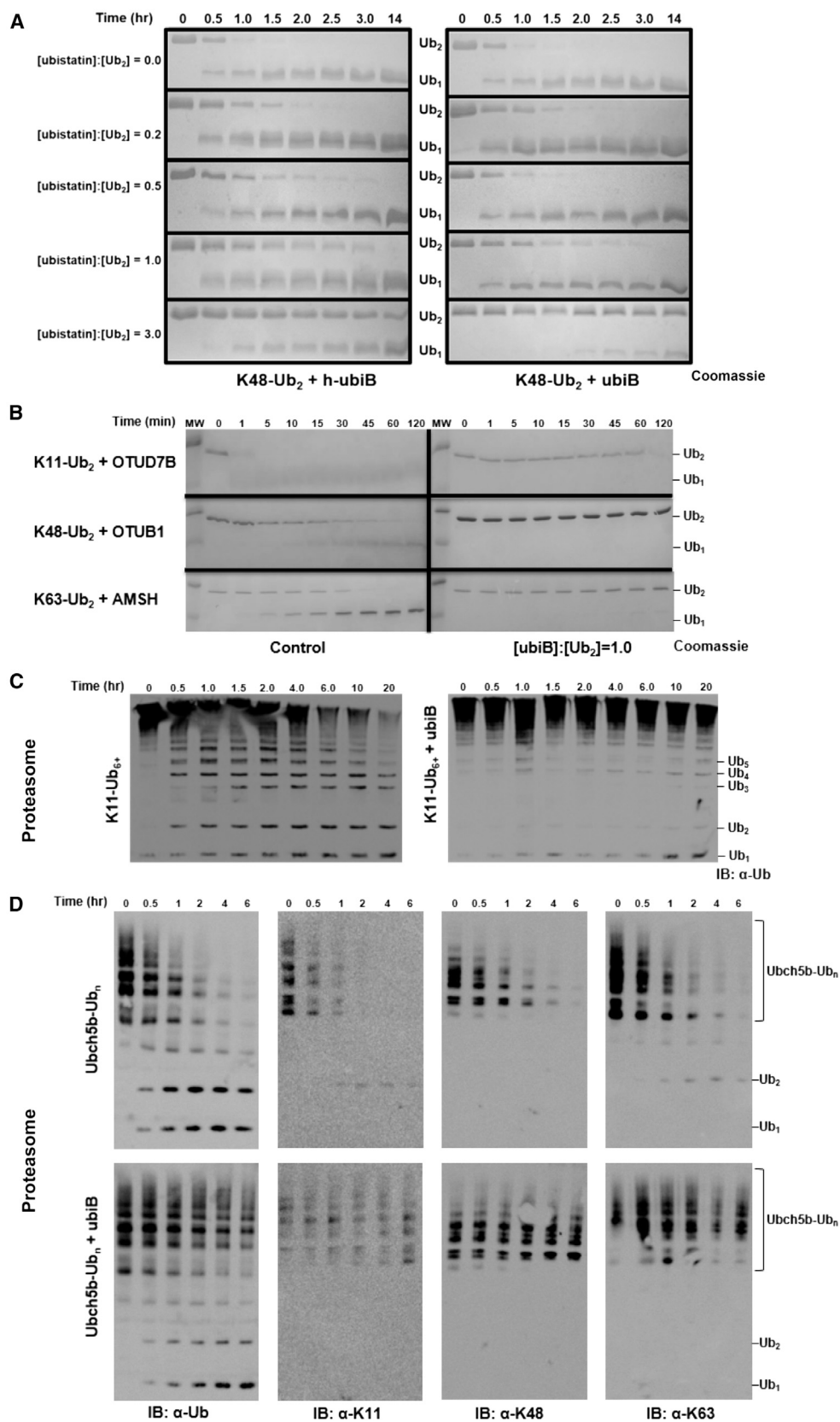
The structure of the ternary complex, Ub:ubiB:Ub, features a sandwich-like arrangement, with two Ubs encompassing the ubiB molecule situated between them (Figure 5D). The Ubs are positioned one on each side of the compound and oriented such that they contact the ligand through their  $\beta$ -sheet surface containing the hydrophobic patch. This arrangement allows the ubiB molecule to interact with two Ubs simultaneously by utilizing both halves (i.e., the naphthotriazole moieties), one for each Ub. The interface between each Ub and the corresponding half of ubiB is similar to that between Ub and h-ubiB, and the flexible C terminus of Ub is positioned such that it can interact with ubiB's sulfonates (see above). It is worth mentioning that the NMR restraint-driven docking also produced a different arrangement, with both Ubs positioned on the same face of ubiB while still contacting different halves of ubiB (Figure S8B). However, that structure was in poor agreement with the SANS data (cf. Figure 5C). It is noteworthy that the SANS data can clearly distinguish between the structural arrangement in which the two Ubs are positioned on the opposite faces (sides) of ubiB and the one where both Ubs are bound to the same face.

The structure of the Ub:ubiB:Ub complex (Figure 5D) was then used as the starting structure for a 2  $\mu\text{s}$  MD simulation. Although both ubiB and Ubs moved and rearranged during the simulation, the ternary complex remained intact. Interestingly, some of the MD snapshots featured the C-terminal G76 of one Ub in close proximity to K48 of the other Ub (Figure S8C) and resembled the structural arrangement in the closed conformation of K48-Ub<sub>2</sub> (Cook et al., 1992; Varadan et al., 2002), thus suggesting that ubiB can intercalate into K48-Ub<sub>2</sub>.

The SANS and  $P(r)$  profiles for K48-Ub<sub>2</sub> in the ubiB-bound state are drastically different from those of both the ternary Ub:ubiB:Ub complex and K48-Ub<sub>2</sub> alone (Figures 5A and 5B). The analysis revealed a larger  $R_g$  compared with K48-Ub<sub>2</sub> (Table 3) and a marked increase in the fraction of atoms with distances  $>20$  Å from each other (Figure 5B), suggesting a substantial increase in the apparent size of Ub<sub>2</sub> upon ubiB binding. Combined with the higher  $I(q \approx 0)$  value at approximately the same Ub<sub>2</sub> concentration (Figure 5A) and significantly longer  $^{15}\text{N}$   $T_1$  than for free Ub<sub>2</sub> (Figure S2B, Table 1), these results point to the presence of more than one Ub<sub>2</sub> molecule in the complex (as

### Figure 5. Characterization of the Complexes of Ub and K48-Ub<sub>2</sub> with ubiB Using SANS

(A and B) SANS intensity (A) and pair distribution function (B) for Ub or K48-Ub<sub>2</sub> alone and in the presence of an equimolar amount of ubiB. (C) Agreement between the experimental SANS intensity  $I(q)^{\text{exp}}$  (open circles) and the  $I(q)^{\text{calc}}$  calculated for the structure shown in (D). Top, overlay of the  $I(q)$  values; bottom, the ratio of  $I(q)^{\text{exp}}$  to  $I(q)^{\text{calc}}$ . The error bars in (A) and (C) represent SEs in the experimental  $I(q)$  values, and in (B) SEs in the derived  $P(r)$ . (D) Structural model of the Ub:ubiB:Ub complex obtained by rigid-body docking. Ubs are shown as ribbons colored blue and red, ubiB is in stick representation. (E) Schematic illustration of the Ub:ubiB:Ub complex and possible structural arrangements of K48-Ub<sub>2</sub> and K63-Ub<sub>2</sub> in complex with ubiB. See also Figures S8 and S9.



(legend on next page)



illustrated in Figure 5E). Based on the observed dimerization of Ub upon ubiB binding, one might expect that Ub<sub>2</sub> would follow a similar trend and form a dimer in the ubiB-bound state. In fact, both the  $R_g$  and the pair distribution function measured for Ub<sub>2</sub> + ubiB match predictions for the closed conformation of K48-linked tetra-Ub (PDB: 2O6V) (Figure S9A). These results suggest that the ubiB:K48-Ub<sub>2</sub> interaction involves two Ub<sub>2</sub> molecules and that the shape of the complex is similar to the closed form of K48-Ub<sub>4</sub> (packed as a dimer of Ub<sub>2</sub>s). The actual structural arrangement in such complexes is unclear at this point, but several possible scenarios are sketched in Figure 5E (also Figure S9B).

The architecture of polyUb complexes with ubistatins would depend on the ability of the Ub chain to adopt specific conformations, and could be linkage dependent. The compact packing of K48-Ub<sub>2</sub>s in the ubiB-bound state might reflect the ability of K48-linked chains to adopt conformations that enable close contacts between the hydrophobic patches of the interacting Ub moieties (Fushman and Walker, 2010; Pickart and Fushman, 2004; Varadan et al., 2002). By contrast, K63-linked chains, which adopt more extended conformations (Pickart and Fushman, 2004; Varadan et al., 2004), might form extended complexes with ubiB (e.g., as illustrated in Figure 5E). In fact, the stark increase in <sup>15</sup>N T<sub>1</sub> for K63-Ub<sub>2</sub> in the ubiB-bound state compared with ubiB-bound K48-Ub<sub>2</sub> and even free K48-Ub<sub>4</sub> (Table 2, Figure S2B) points to this possibility.

### Ubistatins Shield Ubiquitin Conjugates from Deubiquitinases

In light of the observation that select ubistatins impair activity of the proteasomal DUB Rpn11 (Figure 1C and Table S1), we set out to examine their effect on other DUBs. We found that both ubiB and h-ubiB exhibited a concentration-dependent effect on the ability of another proteasome-associated DUB, Ubp6, to disassemble K48-Ub<sub>2</sub> (Figure 6A) and K63-Ub<sub>2</sub> (not shown). Preservation of K48-Ub<sub>2</sub> was observed at or above stoichiometric amounts of ubistatin to Ub. With both linkage types, ubiB exhibited a stronger ability than h-ubiB to block deubiquitination by Ubp6.

These observations encouraged us to test if ubiB had a similar effect on other DUBs cleaving a variety of Ub conjugates. Highlighting the ability of ubiB to bind diverse linkage types, ubiB prevented the dimeric Ub substrates K11-Ub<sub>2</sub>, K48-Ub<sub>2</sub>, and K63-Ub<sub>2</sub> from being disassembled by their respective linkage-specific DUBs: Cezanne (OTUD7B), OTUB1, and AMSH (Figure 6B).

We then examined if ubiB affects DUB activity of the 26S proteasome, which is essential for cellular viability (Goldberg, 2012). Recent work revealed that K11-linked chains of six or more Ub units signal substrates for rapid degradation by the proteasome (Meyer and Rape, 2014) and that proteasomal DUBs rapidly

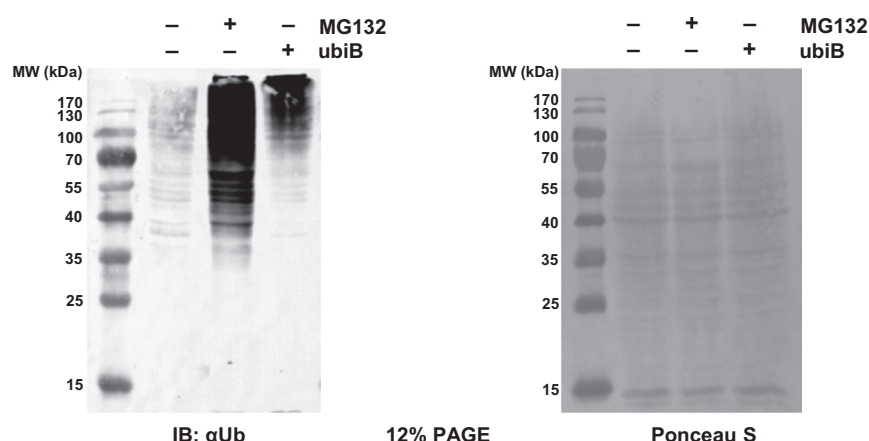
disassemble K11 linkages (Mansour et al., 2015). Using K11-Ub<sub>6+</sub> as a substrate, in the absence of ubiB, the proteasome produced numerous cleavage intermediates after just 30 min, and virtually all of the starting material was depleted at the end of the 20 hr time course (Figure 6C). However, in the presence of ubiB, the high-molecular-weight K11-Ub<sub>6+</sub> corresponding to the starting product remained constant and only trace amounts of cleavage intermediates appeared, likely in concentrations approaching the detection limit of the antibody (Figure 6C). Finally, we utilized Ubch5b-Ub<sub>n</sub>, an autoubiquitinated E2 carrying all possible Ub linkages (Mansour et al., 2015), in order to determine if and how ubiB alters the ability of the proteasome to deubiquitinate a complex substrate, independent of degradation. In the absence of ubiB, the 26S proteasome removed virtually all high-molecular-weight forms of Ubch5b-Ub<sub>n</sub> within 2 hr. By contrast, a significant amount of processing intermediates still remained in the presence of ubiB (Figure 6D). To understand which linkage types were processed, we then analyzed each time point using three linkage-specific antibodies (for K11, K48, or K63). In the absence of ubiB, the respective blots showed complete removal of K11, K48, or K63 linkages by the proteasome at the latest time points (Figure 6D). By contrast, the addition of ubiB preserved all three linkage types in the Ubch5b-Ub<sub>n</sub> substrate, and eliminated low-molecular-weight intermediates. The results of this experiment indicate that ubiB protects the K11, K48, and K63 linkages from disassembly by the proteasome when Ubch5b-Ub<sub>n</sub> is used as ubiquitinated substrate.

### Ubistatin B Penetrates Human Cancer Cells and Perturbs the Ubiquitin Landscape

The *in vivo* effects of ubistatins have been previously demonstrated in cell extracts, *Xenopus* eggs, and through micro-injection (Bellare et al., 2008; Verma et al., 2004). During our preliminary testing of ubiB on cancerous human cell lines, we consistently observed that ubiB associated with cells in the absence of chemical permeabilization. To determine if this effect was solely due to ubiB interactions with the plasma membrane or via internalization, we incubated HeLa and RCC4 cells with 10  $\mu$ M of ubiB for 6 hr. We observed distinct ubiB fluorescence at and within the boundaries of the fluorescently labeled plasma membrane in each cell type (Figure S10A). Notably, the morphology of the treated cells (Figure S10B) indicated that a 10  $\mu$ M dose of ubiB was not toxic, at least within this time period. Given the relatively high affinity of ubiB for polyUb, we further speculated that ubiB directly interacts with polyUb within cells. Next, we conducted immunofluorescence assays with anti-Ub, as well as three linkage-specific antibodies on fixed, permeabilized HeLa cells pretreated with ubiB. Signals from both ubiB and all four antibodies were distributed broadly within the cell, including the nucleus (Figure S10E).

### Figure 6. Ubistatins Shield polyUb from Disassembly by DUBs

- (A) Disassembly of K48-Ub<sub>2</sub> by Ubp6 over the indicated time course in the presence of varying concentrations of h-ubiB (left) or ubiB (right).  
 (B) Linkage-selective DUBs rapidly disassemble their respective linkage type (left), while stoichiometric amounts of ubiB prevent any cleavage within the time course (right).  
 (C) Disassembly of K11-Ub<sub>6+</sub> by yeast 26S proteasome in the absence (left) and presence (right) of ubiB.  
 (D) Processing of Ubch5b-Ub<sub>n</sub> substrate by the proteasome in the absence (top row) or presence (bottom row) of ubiB. The change in the Ub profile over time was followed with the indicated antibody.



**Figure 7. The Impact of ubiB Internalization on the Ub Landscape in Human Cancer Cells**

HeLa cells were treated with ubiB or MG132 for 8 hr prior to lysis. Anti-Ub is used to probe the Ub landscape of HeLa cells lysed in SDS. Despite the dramatic difference in high-molecular-weight Ub conjugates, Ponceau S staining (right panel) shows equal loading of proteins across all lanes. See Figure S10 for cell images and additional analysis of the Ub landscape.

## DISCUSSION

This study provides a detailed characterization of how ubistatins interact with Ub and polyUb chains, describing previously

We then set out to investigate changes to the cellular polyUb landscape following ubiB treatment. Using the standard lysis method with RIPA buffer, we observed a signal from ubiB in harvested cells, which was retained in both pelleted cell debris and supernatants following lysis (Figure S10C). Analysis by western blot indicated that treatment with ubiB increases the amount of Ub conjugates, including those with K48 and K63 linkages, in a dose-dependent manner (Figure S10D). To further examine this effect, HeLa cells treated with ubiB or proteasome inhibitor MG132 were directly lysed by boiling in Laemmli buffer and analyzed with a Ub-specific antibody (Figure 7). Both ubiB and MG132 treatments resulted in marked changes to the Ub landscape, notably, an increase in high-molecular-weight Ub conjugates. On closer examination, Ub conjugates below 70 kDa were greatly diminished in the ubiB-treated cells, in contrast with MG132 and the control. Both MG132 and ubiB impair substrate degradation by the proteasome, albeit through distinct mechanisms. While MG132 acts directly on the proteolytic sites in the 20S core particle, ubiB's binding to polyUb chains shields them from recognition by proteasomal Ub receptors and shuttling factors (Verma et al., 2004), as well as a spectrum of DUBs (Figure 6). The latter properties are a likely reason for the increase in high-molecular-weight Ub conjugates upon ubiB treatment.

Note that the observed accumulation of Ub conjugates upon ubiB treatment does not contradict the ability of ubiB to block substrate ubiquitination (Figures 1D and 1E), because the effects of ubiB on deubiquitination and ubiquitination occur at different thresholds ( $IC_{50}$  of 1  $\mu$ M and 10  $\mu$ M, respectively), with deubiquitination being the more sensitive. For example, with 1  $\mu$ M ubiB there is at most 5%–10% inhibition of ubiquitination (Figure 1E), but ~50% inhibition of deubiquitination (Table S1). Achieving complete inhibition of ubiquitination would require high ubiB concentrations ( $\geq 100$   $\mu$ M) that might not be attainable in cells. Therefore, the inhibitory effect of ubiB on deubiquitination predominates.

To summarize, we detected ubiB uptake using imaging of either live cells or after fixation. These results are consistent with the observation that ubiB perturbs the cellular pool of Ub conjugates. Further work will be required to determine if ubiB binds primarily to Ub chains within cells and whether it can serve as a fluorescent marker for cellular Ub.

unknown functional consequences of these interactions. Based on the screening of ubistatin derivatives in CFTR ubiquitination, DUB shielding, and (poly)Ub binding assays, we demonstrate that ubistatin B (ubiB) has the greatest ability to interact with Ub conjugates. By dissecting the binding interaction of a hemi-ubistatin (h-ubiB) with Ub, we found that ubiB contains two Ub-binding entities. Furthermore, confirming that ubiB can simultaneously bind multiple Ub units, we detected the formation of higher-order complexes of ubiB with Ub and with polyUb. Notably, all active ubistatin variants recognized a conserved binding surface on Ub, regardless of whether Ub was present as a monomer or polymers of different linkage types. The structure of the Ub:h-ubiB complex and SANS-derived models of the Ub:ubiB:Ub complex provide atomic-level details of how ubistatins interact with Ub. In particular, this study revealed the critical role of electrostatic/polar interactions between ubistatin's acidic groups and the basic residues surrounding the surface hydrophobic patch of Ub. The precise orientation of these critical residues on the surface of Ub narrowly restricts ubistatin binding, such that ubiB, for example, is specific for Ub yet reasonably inert to its closest homolog Rub1. That ubistatins can associate with more than one Ub simultaneously means that they can intercalate into polyUb and thus shield Ub conjugates or polyUb chains from a range of DUBs (including those associated with the proteasome). The ability of ubistatins to penetrate mammalian cells now allows for a broad range of potential applications that merit additional investigation into this class of compounds.

In the cell, diverse classes of Ub-binding domains (UBDs) are found across several protein families that enable the broad spectrum of polyUb signals to be interpreted (Komander and Rape, 2012). Our discovery of synthetic molecules capable of binding Ub as well or better than natural UBDs provides for numerous potential future applications. Thus far, two distinct nano-scale systems, gold or silver nanoparticles (Calzolari et al., 2010; Mangini et al., 2014) and fullereneol (Zanzoni et al., 2015) have been reported to form specific interactions with Ub. Several metal ions (Cd, Cu, Hg, Pt, Zn) have also been found to form adducts with Ub in which H68 coordinates at least one ion (e.g., Arena et al., 2013; Falini et al., 2008). Ongoing efforts will undoubtedly uncover other small molecules capable of binding Ub. However, ubistatins already present many advantages over other existing synthetic Ub-binding molecules: the ubistatins are highly

specific for Ub, bind Ub tightly, penetrate cells, have well characterized *in vitro* and *in vivo* outcomes, and provide a fluorescent readout.

From a historical perspective, ubistatin A was the first small molecule shown to bind K48-linked polyUb (Verma et al., 2004). Here, we demonstrate that ubistatin derivatives bind both Ub and polyUb. The Ub:ubistatin interaction is unique in that ubistatins share no resemblance to any known Ub-binding molecules; however, these compounds can now serve as a starting material to isolate or design more effective preclinical candidates. In fact, our characterization of the ubistatin derivatives suggests that substituent groups and positions can be chemically modified without dramatically reducing Ub binding. The similarities carry over to polyUb, as the changes in the NMR spectra of K48-Ub<sub>2</sub> exhibit nearly identical patterns for ubistatin B (ubiB) (Figure 2K) and ubistatin A (Verma et al., 2004). Combined NMR and SANS analyses also showed that the interaction of ubiB with polyUb chains of different linkages (e.g., K48- and K63-linked) can result in different spatial arrangements and stoichiometries. It is tempting to speculate that related small molecules can be identified or designed that specifically bind interaction surfaces on the numerous UBL domains found in the cell. For instance, minor derivatization of ubistatins may improve binding to Rub1 (Nedd8 in humans), which contains the same hydrophobic patch as Ub but differs in the surrounding residues. This approach could be extended to other UBL family members (e.g., FAT10, ISG15, SUMO) to interfere with their own unique signaling pathways or to modulate the interactions of a UBL within multidomain systems, such as Parkin or USP7.

The development of agents to target various aspects of the UPS has been underway for over a decade and has already yielded important outcomes (Buckley and Crews, 2014). In this regard, we propose that Ub itself may represent an alternative and potentially more robust therapeutic target compared with the enzymatic components of the UPS. The highly conserved amino acid sequence of Ub suggests that it is unlikely that a treated cell could tolerate drug refractory mutations in Ub. To highlight this point, mutation of several ubistatin-interacting residues in Ub, especially R42 and R72, are lethal in yeast (Roscoe et al., 2013), effectively making Ub a static target. However, the chemical properties of ubistatins have raised some concerns regarding their use as drugs (Bellows and Tyers, 2004). To mitigate some of these concerns, here we show that ubistatins can penetrate cells and have measurable effects on the Ub landscape. The use of ubistatins in cell-based and animal experiments or as therapeutics would be enhanced by the development of pro-drug forms of the parent molecule that shield the sulfonate groups by reversible covalent bonds or non-covalent complexes. Even if some other factors limit ubistatins as therapeutics, the intrinsic fluorescence and the ability of ubistatins to accommodate modular substituent groups will allow them to serve as powerful reagents in biochemical studies (Bellare et al., 2008).

The mechanism of action of ubistatins is reminiscent of that for endogenously expressed UBDs (Sims et al., 2012; Yoshida et al., 2015). In each case, the cell experiences a buildup of polyUb, but notably this is not lethal over short time periods. The shielding by a given UBD likely limits DUB activity, further contributing to

polyUb buildup. More generally, molecules like ubistatins and other select proteins allow for the controlled accumulation of polyUb linkage types: ubiB has the greatest effect on K48 linkages, the triple-UIM RAP80 derivative (Rx3-A7) is highly selective for K63 linkages (Sims et al., 2012), and TR-TUBEs cause an increase in all Ub-Ub linkage types (Yoshida et al., 2015). Yet, the chemical properties of ubistatins combined with resistance to proteases offer advantages compared with the protein-based UBDs. Ubistatins may also have applications in modulating cellular signaling pathways, such as those triggered by CXCR4 receptors (Majetschak, 2011). Finally, given that ubistatins have been shown to arrest the cell cycle, producing effects similar to proteasome inhibitors (Verma et al., 2004), and combined with the data presented in this study, we believe that the ubiquitin signal is a plausible candidate for therapeutic intervention in the ubiquitin-proteasome pathway.

## STAR★METHODS

Detailed methods are provided in the online version of this paper and include the following:

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## SUPPLEMENTAL INFORMATION

Supplemental Information includes 10 figures and 4 tables and can be found with this article online at <https://doi.org/10.1016/j.str.2017.10.007>.

## AUTHOR CONTRIBUTIONS

T.A.L. synthesized ubistatin compounds. Rpn11 functional assays were performed by F.P., T.-F.C., R.J.D.; CFTR assays by J.L.B. and J.L.G.-F.; NMR measurements and analysis by M.A.N., U.K.N., O.W., D.Z., C.M.C., S.M.B.,

D.F.; SANS studies by C.A.C., S.K., D.F.; structure calculations by O.W. and M.A.N.; fluorescence measurements by M.A.N., W.M., A.T., O.H. and analyzed by D.F.; DUB assays by M.A.N., W.M., O.H., and cellular assays by M.A.N., A.T., and M.H.G. All authors contributed to writing the manuscript and revisions.

## ACKNOWLEDGMENTS

We thank George H. Lorimer and Arnon Henn for access to their fluorimeters, Ananya Majumdar for help with setting up triple-resonance NMR experiments, Rajesh Singh for the Rub1 sample, Allan Weissman for RCC4 cells, and Konstantin Berlin for useful discussions. The project has been funded in part from the NCI Initiative for Chemical Genetics, NIH, under contract no. N01-CO-12400 to Stuart L. Schreiber, by NIH grants GM065334 to D.F. and GM095755 to D.F. and M.H.G., ISF grant 909-14 to M.H.G., and by NIH grants GM075061 and DK079307 and by Cystic Fibrosis Foundation Therapeutics grant BRODSK13XXO to J.L.B., and a Fulbright postdoctoral fellowship to M.A.N., and utilized NMR instrumentation supported in part by NSF grant DBI1040158 and neutron-scattering facilities supported in part by the NSF under agreement no. DMR-0944772. R.J.D. is a founder and shareholder of Cleave Biosciences, which is developing drugs that target the ubiquitin system.

Received: July 21, 2017

Revised: September 9, 2017

Accepted: October 24, 2017

Published: November 16, 2017

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## STAR★METHODS

## KEY RESOURCES TABLE

REAGENT or RESOURCE	SOURCE	IDENTIFIER
<b>Antibodies</b>		
Rabbit polyclonal Anti-ubiquitin	Dako	Cat#z0458; RPID: AB_2315524
Rabbit monoclonal K11 Ubiquitin linkage specific, clone 2A3/2E6	Merck-Millipore	Cat# MABS107-I; RRID:AB_2713901
Rabbit monoclonal K48 Ubiquitin linkage specific, clone Apu2	Merck-Millipore	Cat# 05-1307; RRID:AB_1587578
Rabbit monoclonal K63 Ubiquitin linkage specific, clone Apu3	Merck-Millipore	Cat# 05-1308; RRID:AB_1587580
IgG goat anti-rabbit (H L)-HRP conjugate	BioRad	Cat#170-6515; RPID: AB_11125142
IgG H&L Goat anti-rabbit (Cy5®) preadsorbed	Abcam	Cat#ab97077; RPID: AB_10679461
Mouse anti-HA monoclonal unconjugated (clone 12CA5)	Roche	Cat# 11583816001; RRID: AB_514505
<b>Bacterial and Virus Strains</b>		
BL-21(DE3) Rosetta™ Chemically Competent <i>E. coli</i>	Novagen	Cat# 71397
M15 Chemically Competent Cells	Qiagen	N/A
MAX Efficiency™ DH5α™ Chemically Competent Cells	ThermoFisher	Cat#18258012
<b>Biological Samples</b>		
Human Proteasomes - erythrocyte isolated	BioMol/Enzo Life Sciences	Cat# BML-PW9310-0050
Yeast ( <i>S. cerevisiae</i> ) proteasomes	<a href="#">Mansour et al., 2015</a>	N/A
ER-derived microsomes	<a href="#">Nakatsukasa et al., 2008</a>	N/A
<b>Chemicals, Peptides, and Recombinant Proteins</b>		
2-mercaptoethanol 99%	Sigma-Aldrich	Cat#M3148
Dithiothreitol (DTT)	Gold Bio	Cat#DTT25
Isopropyl-β-D-1-thiogalactoside (IPTG)	Gold Bio	Cat#I2481C50
Complete protease inhibitor cocktail	Roche	Cat#11836145001
Potassium Phosphate dihydrate KH <sub>2</sub> PO <sub>4</sub>	Merck	Cat#529568
Potassium Phosphate dibasic K <sub>2</sub> HPO <sub>4</sub>	Merck	Cat#105104
Sodium Phosphate dihydrate NaH <sub>2</sub> PO <sub>4</sub>	Merck	Cat#567545
Sodium Phosphate dibasic Na <sub>2</sub> HPO <sub>4</sub>	Merck	Cat#567550
Ampicillin sodium salt	Sigma	Cat#A9518
Kanamycin	Sigma	Cat#K4000
Chloroamphenicol	Sigma	Cat#C0378
KOD Hot Start Master Mix	Novagen	Cat#71842
Phenylmethanesulfonyl fluoride (PMSF)	Sigma-Aldrich	Cat#P7626
IGEPAL CA-630	Sigma-Aldrich	Cat#18896
Adenosine 5'-triphosphate, disodium, trihydrate (ATP)	Fisher Scientific	Cat#10326943
Ammonium Chloride ( <sup>15</sup> N, 99%)	Cambridge Isotope Laboratories, Inc	Cat#NLM-467
D-Glucose (U-13C6, 99%)	Cambridge Isotope Laboratories, Inc	CLM-1396-5
Albumin, from Bovine Serum	Sigma-Aldrich	Cat#A3294
Coomassie R-250 Brilliant Blue	ThermoFisher	Cat#20278
Tris(2-carboxyethyl)phosphine-HCl (TCEP-HCl)	Gold Bio	Cat# TCEP25
MG132	Calbiochem Merck Millipore	Cat#474790

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**Continued**

REAGENT or RESOURCE	SOURCE	IDENTIFIER
1-oxyl-2,2,5,5-tetramethyl-3-pyrroline-3-methyl methanesulfonate (MTSL)	Toronto Research Chemicals	Cat#O875000
Wheat Germ Agglutinin, Texas Red® conjugate	Life Technologies	Cat#W7024
Oregon Green® 488 maleimide	ThermoFisher	Cat#O6034
I <sup>125</sup> -Ubiquitin	This work	
Ub4pep~Oregon Green® conjugate	This work	
Ubistatin B (ubiB)	<a href="#">Verma et al., 2004</a>	N/A
Ubistatin compound 1	This work	N/A
Ubistatin compound 2	This work	N/A
Ubistatin compound 3	This work	N/A
Ubistatin compound 4	This work	N/A
Ubistatin compound 5	This work	N/A
Ubistatin compound 6	This work	N/A
Ubistatin compound 7	This work	N/A
Ubistatin compound 8	This work	N/A
Ubistatin compound 9 / Hemi-ubistatin B (h-ubiB)	This work	N/A
Hemi-Ubistatin compound 10	This work	N/A
Related ubistatin compound 11	This work	N/A
8-Anilino-1-naphthalenesulfonic acid (ANS)	Sigma	Cat#A1028-5G
<sup>15</sup> N-labeled (distal) dimeric Ub (K48-Ub <sub>2</sub> )	This work	N/A
<sup>15</sup> N-labeled (distal) dimeric Ub (K63-Ub <sub>2</sub> )	This work	N/A
<sup>15</sup> N-labeled monomeric Ub	This work	N/A
<sup>13</sup> C/ <sup>15</sup> N-labeled monomeric Ub	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R42A</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R42E</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R72A</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R72E</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R42E,R72E</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R74A</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>R74E</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>K63D</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>72</sup> (Δ73-76)	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>T12C</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>I36C</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>K48C</sup>	This work	N/A
<sup>15</sup> N-labeled monomeric Ub <sup>K63C</sup>	This work	N/A
<sup>15</sup> N-labeled RUB1	<a href="#">Singh et al., 2012</a>	N/A
<sup>15</sup> N-labeled UBA of UBQLN1	This work	N/A
K11-linked dimeric Ub (K11-Ub <sub>2</sub> )	This work	N/A
K11-linked >hexameric Ub (K11-Ub <sub>6+</sub> )	This work	N/A
K48-linked dimeric Ub (K48-Ub <sub>2</sub> )	This work	N/A
K63-linked dimeric Ub (K63-Ub <sub>2</sub> )	This work	N/A
6xHis-Ubp6 full length	This work	N/A
GST-Cezanne OTUD7B (OTU domain)	This work	N/A
6xHis-OTUB1 (OTU domain)	This work	N/A
GST-AMSH (JAMM/MPN <sup>+</sup> domain)	This work	N/A
Ubch5B-Ub <sub>n</sub>	This work	N/A

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REAGENT or RESOURCE	SOURCE	IDENTIFIER
Experimental Models: Cell Lines		
HeLa	ATCC	ATCC® CCL-2™
RCC4	Gift from Allan Weissman (NCI)	N/A
Recombinant DNA		
pJexpress401 Ub4pep	This work	N/A
pET3a untagged Human wild-type Ubiquitin	<a href="#">Varadan et al., 2002</a>	N/A
pET3a untagged Ub <sup>R42A</sup>	This work	N/A
pET3a untagged Ub <sup>R42E</sup>	This work	N/A
pET3a untagged Ub <sup>R72A</sup>	This work	N/A
pET3a untagged Ub <sup>R72E</sup>	This work	N/A
pET3a untagged Ub <sup>R42E/R72E</sup>	This work	N/A
pET3a untagged Ub <sup>R74A</sup>	This work	N/A
pET3a untagged Ub <sup>R74E</sup>	This work	N/A
pET3a untagged Ub <sup>K63D</sup>	This work	N/A
pET3a untagged Ub <sup>72</sup>	This work	N/A
pET3a untagged Ub <sup>T12C</sup>	This work	N/A
pET3a untagged Ub <sup>I36C</sup>	This work	N/A
pET3a untagged Ub <sup>K48C</sup>	This work	N/A
pET3a untagged Ub <sup>K63C</sup>	This work	N/A
pTXB1-RUB1	<a href="#">Singh et al., 2012</a>	N/A
pGEX2T-1 UBQLN1 UBA domain	<a href="#">Raasi et al., 2005</a>	N/A
pQE30 Ubp6 full length	<a href="#">Mansour et al., 2015</a>	N/A
pOPINK Cezanne/OTUD7B 53-446 (OTU domain)	<a href="#">Mevisen et al., 2013</a>	Addgene:61581
pProEX-OTUB1 (OTU domain)	<a href="#">Wang et al., 2009</a>	Addgene:26959
AMSH	Fushman Lab	N/A
pET3a human UbcH5B	<a href="#">Mansour et al., 2015</a>	N/A
pSM1152 (2 $\mu$ URA3 PPGKCFTR::HA)	<a href="#">Nakatsukasa et al., 2008</a>	N/A
Software and Algorithms		
ImageGauge version 3.45	Fuji Life Sciences	<a href="http://www.fujifilm.com">www.fujifilm.com</a>
Marvinsketch	ChemAxon	<a href="http://www.chemaxon.com">http://www.chemaxon.com</a>
ORCA package	Max-Planck Institut für Chemische Energiekonversion	<a href="https://orcaforum.cec.mpg.de/">https://orcaforum.cec.mpg.de/</a>
PRODRG server	GlycoBioChem Ltd.	<a href="http://davapc1.bioch.dundee.ac.uk/cgi-bin/prodrng">http://davapc1.bioch.dundee.ac.uk/cgi-bin/prodrng</a>
LIGPLOT program	EMBL-EBI	<a href="https://www.ebi.ac.uk/thornton-srv/software/LIGPLOT/">https://www.ebi.ac.uk/thornton-srv/software/LIGPLOT/</a>
Vinci software	Vinci Inc.	<a href="http://www.iss.com/fluorescence/software/vinci.html">http://www.iss.com/fluorescence/software/vinci.html</a>
Zen Lite	Carl Zeiss	<a href="http://www.zeiss.com">www.zeiss.com</a>
Fiji	LOCI - University of Wisconsin-Madison.	<a href="https://fiji.sc">https://fiji.sc</a>
Bruker Topspin version 3.5	Bruker Inc.	<a href="http://www.bruker.com">www.bruker.com</a>
Computer aided resonance assignment (CARA)	Wüthrich lab, ETH	<a href="http://cara.nmr.ch">Cara.nmr.ch</a>
SPARKY 3	Goddard and Kneller, UCSF	<a href="https://www.cgl.ucsf.edu/home/sparky/">https://www.cgl.ucsf.edu/home/sparky/</a>
HADDOCK v2.1	Bonvin Lab, University of Utrecht	<a href="http://haddock.science.uu.nl/">haddock.science.uu.nl/</a>
SASSIE	NCNR, NIST	<a href="https://sassie-web.chem.utk.edu/sassie2/">https://sassie-web.chem.utk.edu/sassie2/</a>
Pymol	Schrödinger, LLC	<a href="http://www.pymol.org">http://www.pymol.org</a>
KdFit, Matlab program for analysis of titration data	Fushman Lab	<a href="http://gandalf.umd.edu/FushmanLab/pdsw">http://gandalf.umd.edu/FushmanLab/pdsw</a>
SLfit, Matlab program for analysis of spin-labeling data	Fushman Lab	<a href="http://gandalf.umd.edu/FushmanLab/pdsw">http://gandalf.umd.edu/FushmanLab/pdsw</a>

(Continued on next page)

**Continued**

REAGENT or RESOURCE	SOURCE	IDENTIFIER
Other		
5 mL GST Trap	GE Life Sciences	Cat#17513001
5 mL His Trap	GE Life Sciences	Cat#17-5248-01
5 mL SP HP	GE Life Sciences	Cat#17-5054-01
16/60 Superdex 75	GE Life Sciences	Cat#28989333
Protein A Sepharose CL4B	GE Life Sciences	Cat#17-0780-01
4.6 x 30 mm XBridge C18 3.5 $\mu$ m column	Waters	N/A

**CONTACT FOR REAGENT AND RESOURCE SHARING**

Further information and requests for reagents should be directed to the leading author: David Fushman ([fushman@umd.edu](mailto:fushman@umd.edu)).

**EXPERIMENTAL MODEL AND SUBJECT DETAILS**

HeLa (female) and RCC4 (male) human cancer cells were cultured in Dulbecco's Modified Eagle Medium (DMEM) supplemented with 10% fetal bovine serum (FBS) in a humidified incubator set at 37°C with 5% CO<sub>2</sub>.

**METHOD DETAILS**

Certain commercial equipment, instruments, or materials are identified in this paper to foster understanding. Such identification does not imply recommendation or endorsement by the National Institute of Standards and Technology, nor does it imply that the materials or equipment identified are necessarily the best available for the purpose.

**Molecular Biology**

Human wild-type Ub in pET3a was used as a template for all Ub mutants. Mutagenesis was carried out using KOD Hot Start Master Mix and corresponding primer pairs. Followed by DpnI digestion, PCR reactions were transformed into DH5 $\alpha$  cells and selected against LB-ampicillin agar. Individual colonies were verified by DNA sequencing from the T7 promoter.

**Preparation of Proteins**

Unlabeled and <sup>15</sup>N-enriched recombinant Ub and Ub variants were obtained using bacterial expression (in *E. coli*) and purification as described (Varadan et al., 2002) and verified using SDS-PAGE and <sup>1</sup>H-<sup>15</sup>N NMR spectra. Ub dimers were assembled using linkage-specific E2 enzymes as detailed elsewhere (Varadan et al., 2002, 2004, 2005). Ub variants Ub<sup>K48R</sup>, Ub<sup>K63R</sup>, Ub<sup>D77</sup>, or Ub<sup>74</sup> containing chain-terminating mutations were used to provide with full control of chain length and the location of the isotope-labeled Ub unit. DUBs were expressed and purified following established protocols. Yeast proteasome and recombinant Ubp6 were purified as described (Mansour et al., 2015).

**NMR Experiments**

All NMR experiments were performed at 23°C on Bruker Avance III 600 MHz and 800 MHz NMR spectrometers equipped with cryo-probes. The protein samples were prepared in 20 mM (1 M = 1 mol/L) phosphate buffer (pH 6.8) containing 0.02% (w/v) NaN<sub>3</sub> and 5% D<sub>2</sub>O. <sup>15</sup>N longitudinal relaxation time (T<sub>1</sub>) was measured using standard pulse sequences. Each experiment was acquired as a pseudo-3D data set with 1024 points in <sup>1</sup>H, 80 increments in <sup>15</sup>N, and typically up to five T<sub>1</sub>-relaxation delays. The data were processed using TopSpin (Bruker Biospin Inc) and analyzed using in-house Matlab software. Residues having overlapping NMR signals were excluded from the T<sub>1</sub> analyses.

**Chemical Shift Perturbations Mapping**

Differences in chemical shifts of amide-group nuclei (<sup>1</sup>H<sub>N</sub>, <sup>15</sup>N) between two samples (A and B) were quantified as amide chemical shift perturbations (CSPs), defined as follows:

$$\text{CSP} = [(\delta_{\text{HA}} - \delta_{\text{HB}})^2 + ((\delta_{\text{NA}} - \delta_{\text{NB}})/5)^2]^{1/2}. \quad (\text{Equation 1})$$

Here  $\delta_{\text{H}}$  and  $\delta_{\text{N}}$  are chemical shifts of <sup>1</sup>H and <sup>15</sup>N, respectively, for a given backbone N-H group. The same equation was used to quantify spectral perturbations upon titration; in this case, A refers to the unbound species, and B corresponds to various steps in the titration.

### NMR-based Titration Assays

NMR titration assays were performed by adding increasing amounts of unlabeled ligand (ubistatin) to  $^{15}\text{N}$ -labeled protein (e.g., Ub) and monitoring changes (shifts, attenuations) in protein signals in  $^1\text{H}$ - $^{15}\text{N}$  HSQC or SOFAST-HMQC spectra. Signal shifts upon titration were quantified as CSPs (see Equation 1, where in this case A corresponds to the free protein and B to the current step in the titration) and fitted to the following equation:

$$\text{CSP} = \text{CSP}_{\max} \times f_B, \quad (\text{Equation 2})$$

where  $f_B$  is the bound fraction of the protein, and  $\text{CSP}_{\max}$  is the maximum possible value of the signal shift, i.e. in the fully ligand-bound state. The analysis was performed using in-house Matlab program Kdfit (Varadan et al., 2004). Various binding stoichiometry models were considered, as detailed in (Varadan et al., 2004). Specifically, for a 1:1 binding model:

$$f_B = \frac{([L_t] + [P_t] + K_d) - \sqrt{([L_t] + [P_t] + K_d)^2 - 4[P_t][L_t]}}{2[P_t]}, \quad (\text{Equation 3})$$

where  $[P_t]$  and  $[L_t]$  are the total molar concentrations of the protein and the ligand, respectively.

A buffer containing 20 mM sodium phosphate at pH 7.5 greatly improved the solubility of the ubistatins, and there was no detectable effect on Ub, which was in 20 mM sodium phosphate buffer at pH 6.8. Control spectra (not shown) revealed negligible CSPs in Ub, even when the pH changed from 6.8 to 7.5. The ability to dissolve ubistatins at higher concentrations allowed us to add smaller volumes of ubistatin stock solution to the  $^{15}\text{N}$  Ub sample.

### NMR Signal Assignment of Hemi-ubistatin and Ub

NMR signals of the nine protons in h-ubiB were assigned using 2D  $^1\text{H}$ - $^1\text{H}$  TOCSY and NOESY spectra. Proton signals in unbound h-ubiB are separated except for H4, H5, and H9, which overlap. Because of the large CSPs observed in Ub upon binding of h-ubiB, NMR signal assignment of Ub in the bound state was confirmed through 3D H<sub>1</sub>-CCCONH and C<sub>1</sub>-CCCONH TOCSY spectra collected on  $^{13}\text{C}/^{15}\text{N}$ -enriched mono-Ub in complex with h-ubiB. The assignment of Ub amide signals from the  $^1\text{H}$ - $^{15}\text{N}$  NMR spectra in the course of the titration was used as a starting point. The nine proton signals from Ub-bound h-ubiB were assigned using 2D  $^1\text{H}$ ,  $^1\text{H}$ -TOCSY and NOESY spectra with  $^{13}\text{C}$ ,  $^{15}\text{N}$  filtering to eliminate signals from Ub. Having determined the assignments of h-ubiB and Ub in the bound state we then performed 3D  $^{13}\text{C}$ -edited,  $^{13}\text{C}/^{15}\text{N}$ -filtered NOESY-HSQC experiment to detect intermolecular NOEs. To obtain high resolution and greater sensitivity with more scans, 2D versions of the  $^{13}\text{C}$ -edited/filtered NOESY-HSQC with either  $^1\text{H}$  or  $^{13}\text{C}$  in the indirect dimension were collected as well, the representative spectra are shown in (Figure S3B). Interproton distances for the eighteen assigned intermolecular NOEs were calibrated using a 2D  $^1\text{H}$ - $^1\text{H}$  NOESY experiment that allows for both inter- and intra-molecular NOEs to appear in the same spectrum (Table S3).

### Site-directed Spin Labeling and Paramagnetic Relaxation Enhancement (PRE) Measurements

Site-directed spin labeling was used to obtain long-range intermolecular distance constraints for determining the structure of Ub:h-ubiB complex. The nitroxide paramagnetic spin label, 1-oxyl-2,2,5,5-tetramethyl-3-pyrroline-3-methyl methanesulfonate (MTSL), was attached to a Cys at position 12, 36, 48, or 63 in Ub, introduced via site-directed mutagenesis, as described (Varadan et al., 2004, 2005). The paramagnetic relaxation enhancement (PRE) effects caused by MTSL were quantified as the ratio of the signal intensities in the NMR spectra of h-ubiB or Ub recorded with MTSL in the oxidized and reduced states. The location of the unpaired electron of MTSL was reconstructed using in-house Matlab program SLfit (Ryabov and Fushman, 2006) based on the ratio of intensities of Ub signals in  $^1\text{H}$ - $^{15}\text{N}$  HSQC spectra recorded with MTSL in oxidized and reduced states, while distances from the electron to individual protons in h-ubiB were determined from signal intensities in 1D  $^1\text{H}$ - $^1\text{H}$  NOESY spectra with  $^{15}\text{N}$  filtering.

### Structure Modeling for Ub:h-ubiB Complex

The structure of the Ub:h-ubiB complex was obtained using the biomolecular docking program HADDOCK v2.1 (de Vries et al., 2010). As no atomic-resolution structure of h-ubiB exists, the compound was first built in "Marvinsketch" (<http://www.chemaxon.com>) and then energy minimized to determine the representative conformation; see details in Figure S4C. Among the two possible conformations of the h-ubiB compound, due to electrostatic repulsion of sulfonate groups the *trans* conformation exhibits lower energy and was chosen for the docking process.

For the starting Ub structure we used the solution NMR optimized coordinates from PDB ID 1D3Z. Ambiguous and unambiguous constraints used to drive and restrain structure calculations of the complex are reported in Table S3. Ambiguous constraints were defined on the basis of the CSPs data collected on  $^{15}\text{N}$  Ub or h-ubiB upon addition of its binding partner. To avoid treating the whole ligand as active, we defined ambiguous constraints (AIR) between residues of the Ub protein and a set of H atoms of h-ubiB that show some contacts in the 3D  $^{13}\text{C}$ -edited,  $^{13}\text{C}/^{15}\text{N}$ -filtered NOESY-HSQC experiment only. A total of 18 unambiguous distance constraints arising from NOEs were introduced as well as 18 other unambiguous distances derived from PRE data. To accurately describe unambiguous distances arising from PRE data, cysteine residues and MTSL atoms were introduced *in silico* at positions 36 and 48 on Ub and distances were defined between the oxygen of MTSL and any other proton of h-ubiB (Table S3). During the docking process, the number of steps for the rigid body stage was set to zero and the temperature for the simulated annealing



was set to 500 K. A total of 2000 structures were generated at the rigid body stage, the best 200 of them were selected for semi-flexible refinement, and 100 were finely retained for the final refinement in water.

The final solvent-refined structures were subjected to clustering and the ten best structures in the lowest energy cluster were retained for analysis (see Table S4 for results). The resulting structures were clustered with a 1.5 Å cut off and the Ub:h-ubiB interaction contacts were analyzed within the LIGPLOT program (Wallace et al., 1995).

The HADDOCK score used during the final stage was calculated as a weighted contribution of different energetic terms:  $1.0 \times E_{vdW} + 0.2 \times E_{elec} + 0.1 \times E_{AIR} + 1.0 \times E_{desolv}$ . The electrostatic surface of Ub was computed using the adaptive Poisson-Boltzmann solver (APBS).

### Small-angle Neutron Scattering (SANS) Measurements and Analysis

Scattering measurements were performed on the 30 m SANS instruments at the NIST Center for Neutron Research (NCNR) in Gaithersburg, MD. Samples of free Ub (4.5 mg/mL) or K48-Ub<sub>2</sub> (4.29 mg/mL) and equimolar mixtures of ubiB with Ub (4.5 mg/mL) or K48-Ub<sub>2</sub> (4.24 mg/mL) were prepared in 20 mM sodium phosphate buffer (pD 6.8) in D<sub>2</sub>O. The neutron wavelength,  $\lambda$ , was 6 Å, with a wavelength spread,  $\Delta\lambda/\lambda$ , of 0.15. Scattered neutrons were detected with a 64 cm x 64 cm two-dimensional position-sensitive detector with 128 x 128 pixels at a resolution of 0.5 cm/pixel. A sample-to-detector distance of 1.5 m was used to cover the range  $0.03 \text{ Å}^{-1} \leq q \leq 0.4 \text{ Å}^{-1}$ , where  $q = 4\pi \sin(\theta)/\lambda$  and  $2\theta$  is the scattering angle. The data were reduced using the IGOR program with routines developed at the NCNR (Kline, 2006) to produce scattering intensity,  $I(q)$ . Guinier analysis was carried out using the equation  $I(q)/I(0) \approx \exp(-q^2 R_g^2/3)$  to estimate the radius of gyration ( $R_g$ ) and the forward scattering intensity,  $I(0)$ , of the samples. Atom pair distribution,  $P(r)$ , was calculated using GNOM (Semenyuk and Svergun, 1991).

### NMR & SANS-based Modeling of ubiB:2xUb Complex Structure

The ternary ubiB:2xUb system consisted of two Ub molecules and the ubiB ligand, which were randomly sampled in space. As a first step, 2000 structures (replicas) of the system were generated by means of rigid-body docking using HADDOCK (de Vries et al., 2010) software where only the first step of the HADDOCK protocol was considered. The strong similarity of both the magnitudes (CSPs, Figures 2B and 2C) and the directions of the NMR signal shifts (Figures S7A and S7B) upon Ub binding to ubistatin B and hemi-ubistatin B indicates similarity of the interatomic contacts in the Ub:ubiB and Ub:h-ubiB complexes. Therefore the intermolecular distance constraints between ubiB and each Ub molecule were the same as for Ub:h-ubiB interaction (Table S3) except that each Ub was assumed to interact with a different half of ubiB (i.e., one Ub with the left and the other Ub with the right naphthotriazole moiety). For each generated structure, the corresponding SANS profile was calculated and compared with the experimental  $I(q)$ . The  $I(q)$  and  $P(r)$  profiles were calculated for each structure using the Xtal2sas (Curtis et al., 2012; Kline, 2006) module of the SASSIE (Curtis et al., 2012) program. The agreement between the experimental and theoretical SANS profiles was quantified using the target function:

$$\chi^2 = \sum_{i=1}^{N_p} \left[ \frac{I(q_i)^{exp} - I(q_i)^{calc}}{\sigma_i} \right]^2, \quad (\text{Equation 4})$$

where  $N_p$  (=77) is the total number of SANS data points,  $I(q_i)^{exp}$  is the measured scattering intensity,  $I(q_i)^{calc}$  is the intensity calculated using Xtal2sas for scattering vector  $q_i$ , and  $\sigma_i$  is the experimental error. The whole set of structures was sorted according to their corresponding  $\chi^2$  values, ranging from 47.8 to 1020, while spanning a range of  $R_g$  values from 16 Å to 20.4 Å. The HADDOCK calculation then proceeded through the flexible docking step, after which 200 best-score final structures were selected and analyzed. The cluster of the top 10 structures is shown in Figure S8A.

In parallel, the structure that best matched the experimental SANS profile after the rigid-body docking step was subjected to molecular dynamics (MD) to account for the various degrees of freedom that were held fixed in the rigid-body docking and to adequately represent the energetics. For the MD protocol, each Ub was modeled with the AMBER99SB force field and the ligand with the GAFF force field. The model was ionized and solvated with TIP3P water molecules in a 75 Å x 75 Å x 75 Å unit cell. The resulting ~38,000 atoms system was energy minimized and equilibrated locally with ACEMD (Harvey et al., 2009) for 2 ns under NPT conditions, of 1 atm at 300 K, nonbonded cutoff of 9 Å, rigid bonds, and PME electrostatics. A time step of 4 fs was used, in conjunction with a hydrogen mass repartitioning scheme. During minimization and the first 2 ns of equilibration, the protein's C $\alpha$  and ligand's heavy atoms were restrained by a harmonic potential with  $k = 1 \text{ kcal} \cdot \text{mol}^{-1} \cdot \text{Å}^{-2}$  (1 cal = 4.184 J). No biasing potentials were imposed during MD production runs, i.e. the ligand and the protein were allowed to diffuse freely through the solvent. The complete MD simulation resulted in a 2  $\mu$ s trajectory, from which 50,000 snapshots were extracted, each snapshot representing 40 ps of the trajectory. The structures were aligned with respect to one of the Ub molecules prior to SANS profile calculation.

The ensemble of structures extracted from the MD trajectory spanned a range of  $R_g$  values from 16.1 to 19.6 Å while the range of  $\chi^2$  values was from 47.9 to 5032. The best- $\chi^2$  structure from the MD trajectory is shown in Figure S8C. As a word of caution, the mathematical problem of deriving structures from SAS data is largely underdetermined. Thus, the structure that best matches our experimental data (Figures 5C and 5D) has to be interpreted as a structure that could undergo fluctuations around its average position.

### Fluorescence-based Binding Assays

Fluorescence measurements were carried out on an ISS PC1 photon counting spectrofluorimeter at 21°C in PBS pH 7.4 buffer. The concentration of ubiB was held constant at 1  $\mu$ M for each titration (10  $\mu$ M for Ub titration) and the concentration of the ligand (in this case, Ub or Ub<sub>2</sub>) was varied. Emission and excitation spectra were acquired in the L-configuration. Anisotropy experiments were performed in T-configuration using 0.5 mm slits, excitation of 386 nm and detection at 485 nm. Data were processed using Vinci software and fit to a single-site binding model (see Equation 3) using in-house Matlab program Kdfit (Varadan et al., 2004). The measured change,  $\Delta_{\text{aniso}}$ , in fluorescence anisotropy upon titration was fit to the following equation:

$$\Delta_{\text{aniso}} = \Delta_{\text{aniso, max}} \times f_B, \quad (\text{Equation 5})$$

where  $f_B$  is the bound fraction of ubiB, see Equation 2 (where in this case  $[P_i]$  is the molar concentration of ubiB and  $[L_i]$  is the molar concentration of Ub or Ub<sub>2</sub>).

### Proteasome/Rpn11 Inhibition Assay

Proteasomes isolated from human erythrocytes were obtained from (BioMol/Enzo Life Sciences). The Rpn11 substrate, Ub4pep, comprised a His<sub>6</sub> tag for purification followed by four consecutive repeats of Ub linked in tail-to-head fashion (i.e., the C-terminal G76 of the upstream Ub fused directly to M1 of the downstream Ub), and with the 4<sup>th</sup> Ub followed by a 30-mer peptide sequence that contained a unique cysteine for labeling. A DNA sequence encoding the Ub4pep substrate was synthesized and cloned into pJexpress401 vector for *E. coli* expression. Oregon Green® 488 was covalently coupled to the unique cysteine in Ub4pep. Purified Ub4pep was incubated with the 5x molar excess of the dye in reaction buffer comprising 50 mM sodium phosphate, 200 mM NaCl, 10% Glycerol, and 100  $\mu$ M TCEP, pH 7.0. The coupling reaction was quenched by the addition of 10 mM DTT. The mixture of fluorescent dye and protein was then applied to a Superdex 75HR column (GE Healthcare) to separate the free label from the labeled protein.

Ubistatin variants were first assayed for inhibition of changes in fluorescence polarization at 10  $\mu$ M in assay buffer (50 mM Tris, pH 7.4, 0.01% NP-40, 1 mM MgCl<sub>2</sub>, 1 mM ATP, 1 mM DTT). IC<sub>50</sub> values of active analogues were obtained by assaying at a series of concentrations (0, 0.39, 0.78, 1.56, 3.13, 6.25, 12.5, 25, 50, and 100  $\mu$ M) in quadruplicate.

### In Vitro Assay of CFTR Ubiquitination

*In vitro* CFTR ubiquitination reactions were performed as described (Nakatsukasa et al., 2008). Briefly, microsomes containing HA-tagged CFTR were incubated with yeast cytosol, <sup>125</sup>I-ubiquitin, an ATP regenerating system and either 100  $\mu$ M of the indicated compound or the equivalent volume of DMSO at 26°C for 60 min. When the effects of the ubistatins were examined, the compounds were dissolved in DMSO to concentrations such that adding 1  $\mu$ L of the stock resulted in the desired concentration in a reaction. The reactions were incubated at 26°C for 60 min and were quenched by the addition of SDS to 1% and *N*-ethylmaleimide (NEM) to 10 mM in 1xTBS. After a further incubation at 37°C for 30 min, TritonX-100 was added to ~1% (and the SDS diluted to 0.1%) by the addition of 900  $\mu$ L of 1.1% TritonX-100 in 1xTBS supplemented with NEM and PIC. Immunoprecipitation of CFTR was accomplished by addition of anti-HA monoclonal antibody (clone 12CA5, Roche) and Protein A Sepharose CL4B (GE Healthcare) and incubation at 4°C for 16 hrs. Immunoprecipitates were washed 3 times in 50mM Tris-HCl pH 7.4, 150 mM NaCl, 1% Triton X-100, 0.2% SDS, 5 mM EDTA, 10 mM NEM and PIC, and subjected to SDS-PAGE. Half of each sample was used to detect HA-CFTR by western blot, and the other half was used to detect of <sup>125</sup>I-Ubiquitin-conjugated CFTR by phosphorimager analysis. The data were quantified using ImageGauge Ver. 3.45 software (Fuji Life Sciences) and normalized to the amount of HA-tagged CFTR protein present in the immunoprecipitation, as measured by western blot. The extent of CFTR ubiquitination in the DMSO control reactions was set to 100%.

### Gel-based Assay for Inhibition of Proteasome and DUBs

A total of 25  $\mu$ M K48-Ub<sub>2</sub> or K63-Ub<sub>2</sub> was incubated with 5  $\mu$ M Ubp6 in 50  $\mu$ L of PBS pH 7.4 buffer. For linkage specific DUBs, 20  $\mu$ M of K11-Ub<sub>2</sub>, K48-Ub<sub>2</sub>, or K63-Ub<sub>2</sub> was incubated with 150 nM Cezanne (OTUD7B), OTUB1, or AMSH respectively. For proteasome assay 5  $\mu$ M of K11-Ub<sub>6+</sub> or Ubch5b-Ub<sub>n</sub> was incubated with 50 nM of purified yeast 26S proteasome (containing Ubp6). All reactions were carried out at 30°C. Samples were taken at the indicated time points and stored in 5x loading dye until analysis on 15% SDS-PAGE. Reactions that included h-ubiB or ubiB were allowed to equilibrate for 20 min prior to addition of the DUB. Visualization was achieved by Coomassie staining. Western blot detection was carried out using polyclonal rabbit anti-Ub (Dako z0458) and linkage specific rabbit monoclonal anti-K11 (Millipore 2021885), anti-K48 (Millipore 2197314), and anti-K63 (Millipore 2063204), all at a 1:1,000 dilution. IgG goat anti-rabbit HRP conjugate (Bio-Rad) at a 1:50,000 dilution was used as the secondary antibody for all blots.

### Characterization of ubiB Internalization in Live Cells Using Fluorescence Microscopy

HeLa and RCC4 cells were cultured in DMEM supplemented with 10% fetal bovine serum in a glass bottom dish. Cells were treated with 10  $\mu$ M ubiB for 6 hrs and washed three times with PBS buffer. The plasma membrane was visualized by treating cells with a 1:500 dilution of wheat germ agglutinin, Texas Red® conjugate (Life Technologies) for 15 min. A Zeiss LSM700 inverted confocal microscope was used to image live cells in Z-stack configuration at 1  $\mu$ M increments with a 63X oil immersion lens. Images were acquired at 512 x 512 pixels with the 405 nm laser for excitation of ubiB and the 555 nm laser to excite the red plasma membrane probe. All images were processed using Zen software (Zeiss) and analyzed with the Fiji distribution of ImageJ.

HeLa cells grown in media containing 10  $\mu$ M ubiB for 6 hrs on a glass coverslip were used for immunofluorescence. Following treatment, cells were fixed in ice cold 100% (v/v) methanol for 5 min at room temperature, permeabilized in PBS, 15% (v/v) Triton X-100, and blocked overnight at 4°C in PBST buffer containing 1% (w/v) BSA, 22.52 mg/mL glycine, and 0.1% Tween 20. The same four Ub antibodies (see above) were used with an anti-rabbit-Cy5 as the secondary antibody. Throughout this procedure ubiB remained within the cells. Cover slips were mounted to microscope slide and sealed. Images were acquired and analyzed as above in the live cell assay.

### Detecting ubiB-induced Perturbations of the Ub Landscape

HeLa cells were treated with 10  $\mu$ M MG132 or 2  $\mu$ M ubiB for 8 hrs, washed two times in PBS pH 7.4 buffer, and harvested. Equal amounts of HeLa cells representing each condition were combined with 2x protein loading dye and immediately lysed at 95°C for 10 minutes. Samples were loaded onto a 12% SDS gel, transferred to a PVDF membrane, blotted against rabbit anti-Ub (Dako) in a 1:1,000 dilution, and detected with goat anti-rabbit HRP (BioRad) in a 1:50,000 dilution.

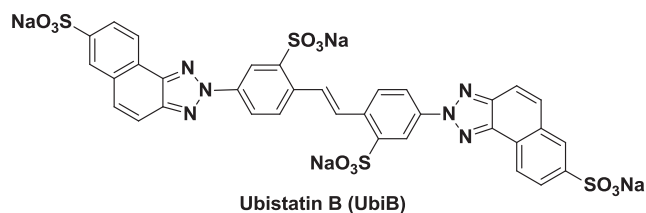
In a separate assay, following 6 hrs ubiB treatment, HeLa cells were lysed chemically in RIPA buffer (25 mM Tris pH 7.6, 150 mM NaCl, 1% NP-40, 1% sodium doxylcholate, 0.1% SDS) in the presence of protease inhibitor cocktail (Millipore 539134). At each stage of lysis, the presence of ubiB was visually detected using a 308 nm UV transilluminator and photographed (Figure S10C). Cell debris were cleared by centrifugation and the protein content of the supernatant was quantified using Bradford assay. Equal amounts of protein were loaded and subjected to Western blot with rabbit anti-Ub as described above.

### Synthesis of Ubistatin Compounds

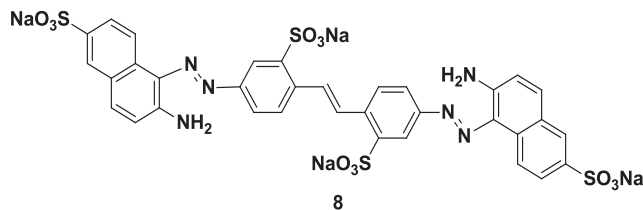
All ubistatin compounds used for study were designed and synthesized as detailed below.

#### General Procedures

Starting materials and reagents were purchased from commercial suppliers and used without further purification. All products were characterized by LC/MS and NMR analysis. Analytical LC/MS chromatography was performed on Waters Alliance HT Liquid Chromatograph coupled with Waters SQ Mass Spectrometer featuring electro spray ionization (ESI+ and ESI-) with MS scans from 100 to 1500 m/z and cone voltage 50 V; mobile phase: 0.01%  $\text{NH}_4\text{OH}$  in water/acetonitrile; column: 4.6 x 30 mm XBridge C18 3.5  $\mu$ m column, 3 mL/min, 2.5 min, Gradient: 5–95% acetonitrile.  $^1\text{H}$  NMR spectra were recorded on a Varian 500 MHz spectrometer or a Bruker 300 MHz spectrometer and are reported in ppm and referenced to residual protons in the NMR solvent.  $^{13}\text{C}$  NMR spectra were recorded at 125 MHz on a Varian spectrometer;  $^{13}\text{C}$  shifts are reported in ppm and referenced to carbon resonances in the NMR solvent. Data below are reported as chemical shift ( $\delta$ , in ppm), splitting (s = singlet, d = doublet, t = triplet, m = multiplet; br = broad), coupling constant in Hz; integration. Masses are reported in daltons.

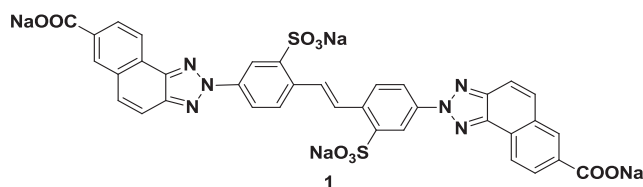


**Ubistatin B (ubiB).** To 925 mg (2.50 mmol) of 4,4'-diamino-2,2'-stilbenedisulfonic acid was added 10 mL of water and 1 mL of conc. HCl and the mixture was cooled in an ice bath (Dobas et al., 1957). A solution of 345 mg (5.00 mmol)  $\text{NaNO}_2$  in 2 mL of water was added over 30 min. After stirring an additional 10 min, the paste was added to a solution of 1.11 g (5.00 mmol) of 6-amino-2-naphthalenesulfonic acid, 200 mg of NaOH (5.00 mmol), and 1.23 g (15.0 mmol) of NaOAc in 150 mL of water cooled in an ice bath. A deep red color appears quickly and the solution was stirred overnight, warming to room temperature. The next day  $\text{Na}_2\text{CO}_3$  (800 mg, 7.55 mmol) was added and the mixture was stirred 1 h before the addition of 7.5 mL of  $\text{NH}_4\text{OH}_{(\text{aq})}$  and 2.50 g of  $\text{CuSO}_4 \cdot 5 \text{H}_2\text{O}$  (10 mmol) dissolved in 20 mL of water. The mixture was heated at 90°C for 1 h, the mixture turning green. The reaction was cooled to 70°C and 7 mL of conc. HCl was added to give a dark solution which was removed from the oil bath. Precipitates form with the addition of 30 g of NaCl. The solids were collected by filtration and dissolved in ca. 40 mL hot water before addition of ca. 200 mL of acetone and cooling. Filtration and rinsing with acetone yielded 916 mg of yellow solid (42%).  $^1\text{H}$  NMR (500 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$ : 9.11 (s, 2H), 8.73 (d,  $J$  = 8.23 Hz, 2H), 8.53 (d,  $J$  = 8.21 Hz, 2H), 8.49 (s, 2H), 8.45 (s, 2H), 8.32 (d,  $J$  = 8.43 Hz, 2H), 8.19 (d,  $J$  = 8.20 Hz, 2H), 7.91–7.95 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 143.15, 141.82, 141.69, 140.98, 137.24, 135.05, 131.45, 130.09, 129.25, 128.03, 126.57, 124.76, 124.27, 124.02, 121.45, 118.34, 115.87; MS: M-H = 833 ( $\text{C}_{34}\text{H}_{22}\text{N}_6\text{O}_{12}\text{S}_4$  = 834). IR Spectrum matched that reported by Bio-Rad Laboratories (BR182452).

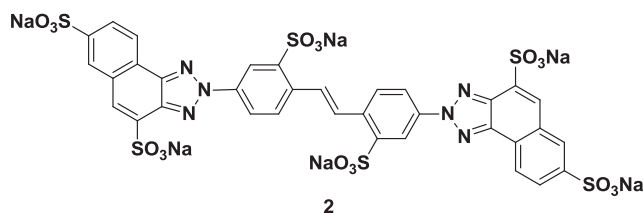


A small amount of the intermediate dye (**8**) was isolated after the  $\text{Na}_2\text{CO}_3$  addition from an aliquot of the reaction mixture by adding  $\text{NaCl}_{(\text{s})}$  and filtering off the red precipitate.  $^1\text{H}$  NMR (300 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.66 (d,  $J = 8.90$  Hz, 2H), 8.24 (d,  $J = 1.88$  Hz, 2H), 8.04 (s, 2H), 7.96 (s, 2H), 7.81–7.84 (m, 4H), 7.72 (dd,  $J = 1.19, 7.93$  Hz, 2H), 7.24 (d,  $J = 8.37$  Hz, 2H), 6.38 (d,  $J = 8.94$  Hz, 2H); MS: M-H = 837 ( $\text{C}_{34}\text{H}_{26}\text{N}_6\text{O}_{12}\text{S}_4 = 838$ ).

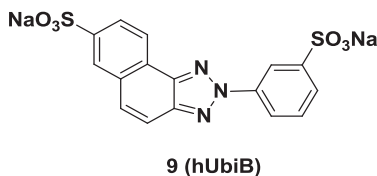
The following compounds were made using the above procedure on the same scale:



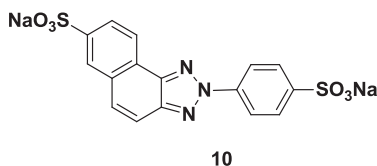
**Compound 1.** Synthesized from 4,4'-diamino-2,2'-stilbenedisulfonic acid and 2-amino-naphthoic acid, reddish brown solid (45%):  $^1\text{H}$  NMR (300 MHz,  $\text{D}_2\text{O}/\text{DMSO}-d_6$ )  $\delta$ : 8.79 (d,  $J = 2.22$  Hz, 2H), 8.50–8.53 (m, 4H), 8.35–8.38 (m, 2H), 8.25 (s, 2H), 8.22–8.25 (m, 2H), 8.03 (d,  $J = 9.29$  Hz, 2H), 7.88–7.97 (m, 4H); MS: M-H = 761 ( $\text{C}_{36}\text{H}_{22}\text{N}_6\text{O}_{10}\text{S}_2 = 762$ ).



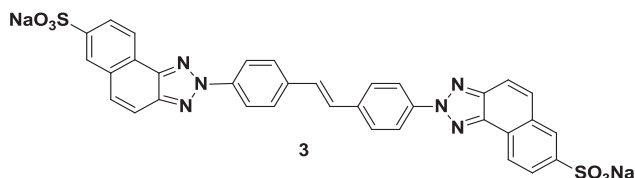
**Compound 2.** Synthesized from 4,4'-diamino-2,2'-stilbenedisulfonic acid and 2-amino-3,6-naphthalenedisulfonic acid, yellow solid (40%):  $^1\text{H}$  NMR (500 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$ : 9.11 (d,  $J = 2.27$  Hz, 2H), 8.73 (d,  $J = 8.35$  Hz, 2H), 8.64 (dd,  $J = 2.26, 8.58$  Hz, 2H), 8.54 (s, 2H), 8.42 (s, 2H), 8.36 (s, 2H), 8.29 (d,  $J = 8.70$  Hz, 2H), 8.23 (dd,  $J = 1.48, 8.35$  Hz, 2H); MS: M-H = 993 ( $\text{C}_{34}\text{H}_{22}\text{N}_6\text{O}_{18}\text{S}_6 = 994$ ).



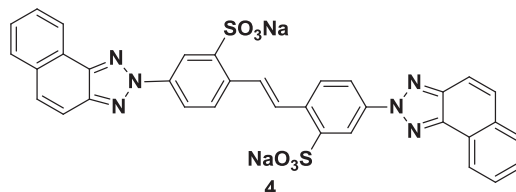
**Compound 9 (h-ubiB).** Synthesized from 3-aminobenzenesulfonic acid and 6-amino-2-naphthalene sulfonic acid, off-white solid (44%):  $^1\text{H}$  NMR (300 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.21 (s, 1H), 8.14 (d,  $J = 7.70$  Hz, 1H), 8.00 (s, 1H), 7.88 (d,  $J = 7.61$  Hz, 1H), 7.71 (d,  $J = 7.57$  Hz, 1H), 7.57 (d,  $J = 7.73$  Hz, 1H), 7.30–7.37 (m, 3H); MS: M-H = 404 ( $\text{C}_{16}\text{H}_{11}\text{N}_3\text{O}_6\text{S}_2 = 405$ ).



**Compound 10.** Synthesized from 4-aminobenzenesulfonic acid and 6-amino-2-naphthalene sulfonic acid, off-white solid (68%):  $^1\text{H}$  NMR (300 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.09 (s, 1H), 8.05 (d,  $J$  = 8.24 Hz, 1H), 7.86-7.95 (m, 4H), 7.80 (d,  $J$  = 7.80 Hz, 1H), 7.40 (d,  $J$  = 9.19 Hz, 1H), 7.27 (d,  $J$  = 9.21 Hz, 1H); MS: M-H = 404 ( $\text{C}_{16}\text{H}_{11}\text{N}_3\text{O}_6\text{S}_2$  = 405).



**Compound 3.** To 354 mg (125 mmol) of 4,4'-diaminostilbene was added 5 mL of water and 0.5 mL of conc. HCl and the mixture was cooled in an ice bath. A solution of 172 mg (2.50 mmol)  $\text{NaNO}_2$  in 1 mL of water was added over 30 min. After stirring an additional 10 min, the paste was added to a solution of 560 mg (2.50 mmol) of 6-amino-2-naphthalenesulfonic acid, 100 mg of NaOH (2.50 mmol), and 610 g (7.5 mmol) of NaOAc in 50 mL of water cooled in an ice bath. A deep red color appeared quickly and the mixture was stirred overnight, warming to room temperature. The next day  $\text{Na}_2\text{CO}_3$ , 400 mg (3.25 mmol), was added and the mixture was stirred 1 h before filtering off the red solid. The solids were added to 50 mL of pyridine and 600 mg of copper (II) acetate and heated at  $130^\circ\text{C}$  for 5 h, the mixture turned deep green. After cooling and filtering, 80 mg of solid was dissolved in a 1:1 mixture of MeOH:DMF with a small amount of  $\text{HCl}_{(\text{aq})}$ . The solution was added to a 5 g PEAX (basic) solid phase extraction column and rinsed with MeOH. The product elutes with the addition of acidic ( $\text{HCl}_{(\text{aq})}$ ) DMF solution, concentration yielded 27 mg of dark brown solid (4%): (300 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.38-8.41 (m, 4H), 8.19 (s, 2H), 8.02 (d,  $J$  = 7.06 Hz, 2H), 7.88 (d,  $J$  = 7.09 Hz, 2H), 7.84 (d,  $J$  = 7.30 Hz, 2H), 7.54-7.57 (m, 8H); MS: M-H = 673 ( $\text{C}_{34}\text{H}_{22}\text{N}_6\text{O}_6\text{S}_2$  = 674).



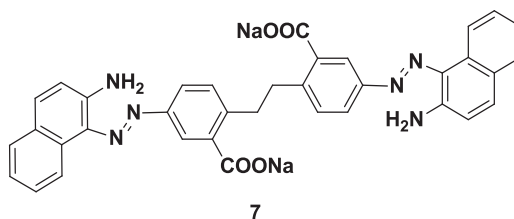
**Compound 4.** To 1.29 g (3.49 mmol) of 4,4'-diamino-2,2'-stilbenedisulfonic acid was added 20 mL of water and 1.7 mL of conc. HCl and the mixture was cooled in an ice bath. A solution of 480 mg (6.98 mmol)  $\text{NaNO}_2$  in 5 mL of water was added over 30 min. After stirring an additional 10 min, the paste was added to a solution of 1.00 g (6.98 mmol) of 2-naphthylamine and 1.71 g (20.9 mmol) of NaOAc in 100 mL of MeOH cooled in an ice bath. A red color appeared quickly and the solution was stirred 1 h before addition of  $\text{Na}_2\text{CO}_3$ , (1.0 g, 8.1 mmol) and the mixture was stirred overnight. Another 100 mL of water and 100 mL of MeOH was added before addition of 11 mL of  $\text{NH}_4\text{OH}_{(\text{aq})}$  and a solution of 3.6 g of  $\text{CuSO}_4 \cdot 5 \text{H}_2\text{O}$  (20 mmol) dissolved in 20 mL of water and the reaction was heated at  $80^\circ\text{C}$  for 24 h. After cooling and filtration, the collected green solids were stirred in acetone for 1 h and filtered. The solid was then treated with 900 mL of refluxing MeOH and filtered, the filtrate was then treated two more times with hot MeOH. The MeOH washings were combined and concentrated giving 1.26 g of a yellow solid (54%):  $^1\text{H}$  NMR (500 MHz,  $\text{DMSO}-d_6$ )  $\delta$ : 8.84 (d,  $J$  = 2.37 Hz, 2H), 8.63 (d,  $J$  = 7.68 Hz, 2H), 8.37 (dd,  $J$  = 2.32, 8.48 Hz, 2H), 8.34 (s, 2H), 8.10 (d,  $J$  = 7.73 Hz, 2H), 7.93-7.99 (m, 6H), 7.73-7.80 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{DMSO}-d_6$ )  $\delta$ : 147.86, 143.83, 142.96, 138.33, 135.65, 132.75, 130.81, 130.02, 128.80, 128.76, 128.51, 127.83, 124.59, 123.63, 120.61, 119.81, 117.02; MS: M-H = 673 ( $\text{C}_{34}\text{H}_{22}\text{N}_6\text{O}_6\text{S}_2$  = 674).

**1,2-Bis(2-carboxy-4-nitrophenyl)ethane.** A solution of 2.91 g (16.0 mmol) of 2-methyl-5-nitrobenzoic acid in 10 mL of basic water (NaOH) was warmed to  $50^\circ\text{C}$  (Müller et al., 1985). To this was added a solution of 48 mL of 30% (w/v) NaOH in water and 42 mL of 10-12% sodium hypochlorite over 1 hr. Precipitate forms and the mixture was stirred at  $50^\circ\text{C}$  overnight. The solid product was filtered and rinsed with cold water. The white solid was then dissolved in a large amount of water, 1.56 g of white solid precipitated out upon addition of 1 N HCl (54%).  $^1\text{H}$  NMR (300 MHz,  $\text{DMSO}-d_6$ )  $\delta$ : 8.64 (d,  $J$  = 3.52 Hz, 2H), 8.20 (dd,  $J$  = 2.55, 8.45 Hz, 2H), 7.50 (d,  $J$  = 8.50 Hz, 2H), 3.39 (s, 4H); MS: M-H = 359 ( $\text{C}_{16}\text{H}_{12}\text{N}_2\text{O}_8$  = 360).

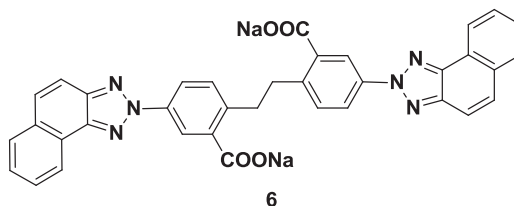
**1,2-Bis(4-amino-2-carboxyphenyl)ethane.** Iron powder (7.29 g, 130 mmol) in 50 mL of water and 1 mL conc. HCl was heated to  $120^\circ\text{C}$ , and to this mixture was added a solution of 940 mg of 1,2-bis(2-carboxy-4-nitrophenyl)ethane in 20 mL slightly basic water (Hein and Pierce, 1954). The mixture was heated 1.5 h before cooling and addition of 1 N NaOH until the mixture was basic. After



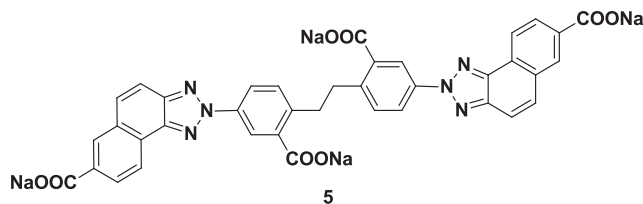
filtration, the solution was made acidic with HCl (aq) and the precipitated solids were filtered and rinsed with acetone giving 415 mg of off-white solid (57%).  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ )  $\delta$ : 7.03 (d,  $J$  = 2.5, 2H), 6.89 (d,  $J$  = 8.2, 2H), 6.63 (dd,  $J$  = 2.5, 8.2, 2H), 4.60 (br s, 4H), 2.88 (s, 4H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ )  $\delta$ : 169.31, 146.35, 131.01, 130.52, 130.05, 117.21, 115.24, 35.25. MS: M-H = 299 ( $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_4$  = 300).



**Compound 7.** To 415 mg of 1,2-bis(4-amino-2-carboxyphenyl)ethane (1.38 mmol) in 10 mL of water and 0.75 mL of conc. HCl cooled to 0°C was added a solution of 191 mg (2.76 mmol) of  $\text{NaNO}_2$  in 2 mL of water over 30 min. The solution was then added to a solution of 395 mg (2.76 mmol) of 2-naphthylamine and 679 mg of NaOAc (8.28 mmol) in 50 mL of MeOH cooled in an ice bath, the solution turning orange quickly. After 20 min, 400 mg (3.25 mmol) of  $\text{Na}_2\text{CO}_3$  was added. A small amount of the reaction mixture was filtered and rinsed with MeOH, the orange solid was pure by LC/MS and NMR analysis.  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ )  $\delta$ : 8.64 (d,  $J$  = 8.53 Hz, 2H), 8.25 (s, 2H), 7.94 (d,  $J$  = 7.23 Hz, 2H), 7.70-7.76 (m, 4H), 7.49-7.54 (m, 2H), 7.37 (d,  $J$  = 4.02 Hz, 2H), 7.29-7.35 (m, 2H), 7.08 (d,  $J$  = 8.89 Hz, 2H), 3.28 (s, 4H); MS: M-H = 607 ( $\text{C}_{36}\text{H}_{28}\text{N}_6\text{O}_4$  = 608).



**Compound 6.** To the mixture above was added 5 mL of  $\text{NH}_4\text{OH}_{(\text{aq})}$  and 1.37 g (5.52 mmol) of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  dissolved in 20 mL of water and the reaction was heated at 90°C overnight. The next day the green mixture with off-white precipitate was cooled and filtered. The filtrate was added to a separatory funnel with 1 N HCl and EtOAc. The water was extracted several times with EtOAc, the EtOAc layers were combined, dried ( $\text{MgSO}_4$ ) and concentrated to give 35 mg of dark solid product (4%) which was converted to the bis sodium salt by treatment with NaOH.  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ )  $\delta$ : 8.59 (d,  $J$  = 7.18 Hz, 2H), 8.42 (d,  $J$  = 2.43 Hz, 2H), 8.04-8.10 (m, 4H), 7.88-7.96 (m, 4H), 7.69-7.79 (m, 4H), 7.46 (d,  $J$  = 8.31 Hz, 2H), 3.29 (s, 4H); MS: M-H = 603 ( $\text{C}_{36}\text{H}_{24}\text{N}_6\text{O}_4$  = 604).



**Compound 5.** To a suspension of 260 mg (0.87 mmol) of 1,2-bis(4-amino-2-carboxyphenyl)ethane in 2 mL of water cooled in an ice bath was added 0.7 mL conc. HCl. A solution of 120 mg (1.73 mmol) of  $\text{NaNO}_2$  in 1 mL of water was added dropwise over 20 min. This was then added dropwise to a cold solution of 321 mg (1.73 mmol) of 2-amino-6-naphthoic acid, 70 mg of NaOH, and 430 mg of NaOAc in 30 mL of water and the resulting suspension was stirred overnight warming to room temperature. The next day 284 mg of  $\text{Na}_2\text{CO}_3$  was added and the mixture was stirred 1 h before addition of 2.7 mL of  $\text{NH}_4\text{OH}_{(\text{aq})}$  and 890 mg of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  in 10 mL. The mixture turns dark and was heated at 90°C for 1 h before cooling to 70°C. The addition of a few drops of 1 N HCl dissolves all solids in the basic solution which was cooled to room temperature. The addition of NaCl precipitates out pink solid which was purified by reverse phase chromatography to produce 80 mg of red solid product (12%).  $^1\text{H}$  NMR (300 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.60 (d,  $J$  = 8.3, 2H), 8.56 (d,  $J$  = 1.0, 2H), 8.53 (d,  $J$  = 2.3, 2H), 8.33 - 8.26 (m, 2H), 8.21 (dd,  $J$  = 2.4, 8.3, 2H), 7.85 (q,  $J$  = 9.2, 4H), 7.54 (d,  $J$  = 8.4, 2H), 3.38 (s, 4H). MS: M-H = 691 ( $\text{C}_{38}\text{H}_{24}\text{N}_6\text{O}_8$  = 692).