

Characterizing crystalline polymorph transitions in HfO₂ by extended x-ray absorption fine-structure spectroscopy

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Atomic layer deposited HfO₂ films on Si(100) substrates have been measured by extended x-ray absorption fine-structure (EXAFS), pre- and postanneal processing. Analysis of the second coordination shell indicates an increase in atomic order with increasing film thickness for each anneal temperature and with increasing anneal temperature for each film thickness. Fourier transformed EXAFS spectra fit with HfO₂ reference phases have identified orthorhombic to tetragonal to monoclinic transformations. Evidence for greater retention of the higher permittivity metastable tetragonal phase corresponding to thinner HfO₂ films is consistent with a surface energy effect giving rise to the critical grain size phenomenon. © 2007 American Institute of Physics. [DOI: 10.1063/1.2789180]

An industry wide effort to incorporate a thin Hf-based high- k ($k \approx 12$ – 25) gate dielectric film to replace SiO₂ ($k = 3.9$) and maintain the prescribed rate of device scaling¹ has given rise to electrical performance enhancements due to improved chemical engineering of thin dielectric films.² High resolution spectroscopic techniques capable of characterizing subtle film and interface stoichiometric variations as a function of gate stack layer deposition and subsequent anneal processing have emerged as critical requirements for the identification of microstructure driven mechanisms that limit or enhance the electrical performance of Hf-based complementary metal oxide semiconductor devices.³

HfO₂ blanket thin films were produced by atomic layer deposition (ALD) via precursor material tetrakis(ethylmethylamino)hafnium with ozone oxidation, TEMA₂Hf + O₃, on Si(100) substrates at 330 °C, while the target HfO₂ thickness was achieved, based on a growth rate of ~ 0.08 nm/cycle. Kirsch *et al.*⁴ determined the scaling limit of HfO₂ to be 15 ALD cycles ($T_{\text{phys}} = 1.2 \pm 0.2$ nm), where electrical results concur with physical evidences (x-ray reflectivity density, low energy ion scattering continuity, and x-ray photoelectron spectroscopy chemical states), indicating that 1.2 nm films begin to deviate from bulk HfO₂ properties. Transistors exhibit an improvement in both peak and high field mobility as HfO₂ thickness scales from 3.3 to 1.2 nm, which is likely due to reduced charge trapping and reduced Coulomb scattering. These results underscore the need for high resolution physical characterization to elucidate microstructure variations corresponding to performance advantages of aggres-

sively scaled HfO₂ films. To that end, extended x-ray absorption fine-structure (EXAFS) spectroscopy measurements have been performed on a series of scaled thickness, uncapped HfO₂ blanket films ($T_{\text{phys}} = 1.4, 1.8,$ and 4.0 nm), following each of the three standard device thermal processing cycles: (i) as deposited at 330 °C, (ii) postdeposition anneal (PDA) of 700 °C for 60 s in NH₃ ambient at 30 Torr, and (iii) PDA plus rapid thermal anneal (RTA) at 1000 °C for 5 s in N₂ ambient at atmospheric pressure. EXAFS is well suited for probing local structural distortions in thin HfO₂ films as a function of anneal processing since the technique is sensitive to correlations on a length scale of a few angstroms.⁵

The experiment was performed at the National Synchrotron Light Source at Brookhaven National Laboratory on beamline X23A2, operated by the National Institute of Standards and Technology. The Hf L_3 absorption edge (9561 eV) was used in the EXAFS data acquisition in the fluorescence detection mode with a 5° angle of x-ray incidence. The $\chi(k)$ functions were obtained by subtracting the atomic absorption background using the AUTOBK code.⁶ The resultant functions were Fourier transformed (FT) using a Gaussian window set for 2.0–12.5 Å⁻¹ k range. Analysis of the FT data provides information on the near-neighbor coordination, distance, and the local disorder around the absorbing Hf atom obtained using the University of Washington developed multiple scattering EXAFS calculation code FEFF8.4.⁷

The cubic (c) HfO₂ phase has a fluorite-type structure (space group F_{3m3}), in which each Hf⁴⁺ ion is coordinated by eight equidistant O²⁻ ions and each O²⁻ ion is tetrahedrally coordinated by four Hf⁴⁺ ions.^{8,9} The reference c -HfO₂ phase^{10,11} did not produce a good fit with any of the FT spectra of this sample set. Figure 1(a) illustrates a set of FEFF8.4 calculated FT scattering path functions corresponding

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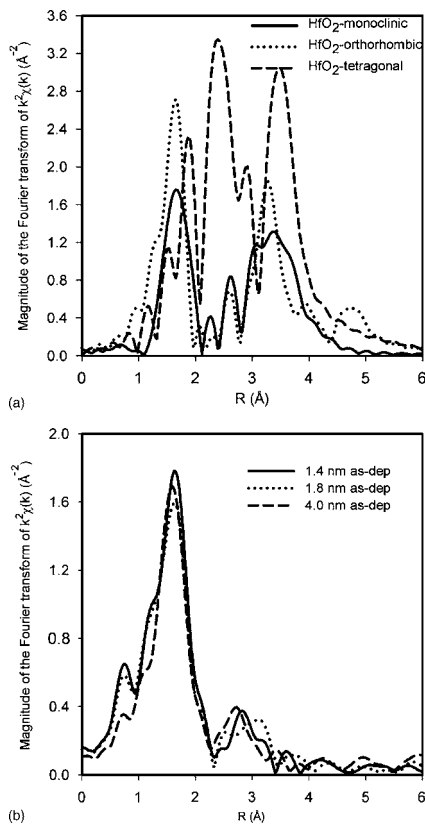


FIG. 1. (a) The reference FEFF8.4 calculated monoclinic, orthorhombic, and tetragonal phases of HfO₂ and (b) the Fourier transformed EXAFS data for 1.4, 1.8, and 4.0 nm thick HfO₂ prior to anneal processing (330 °C deposition temperature).

to monoclinic (*m*),^{12,13} tetragonal (*t*),^{13,14} and orthorhombic (*o*)^{15,16} phases of HfO₂ which were used to identify the polymorphs contributing to the FT data. These references have been rendered for pure phases with theoretical coordination numbers, without Debye-Waller factors. Although the peak intensities differ from experimental data, the location and shape offer valid phase identification comparisons. Figure 1(b) illustrates the FT EXAFS data for 1.4, 1.8, and 4.0 nm thick HfO₂ prior to anneal processing ($T_{\text{deposition}}=330$ °C). It is difficult to interpret the second order peaks conclusively since they all appear amorphous, and definitive order in these thin as-deposited films is not expected.

The temperature, pressure, and anneal ambient definitely affect the polymorphism of HfO₂. The sevenfold coordination structure of *m*-HfO₂ (space group $P2_1/c$) consists of triangularly coordinated O_I-Hf³⁺ and tetrahedrally coordinated O_{II}-Hf⁴⁺, while *t*-HfO₂ (space group $P4_2/nmc$) resembles a distorted fluorite-type cubic structure with Hf⁴⁺ surrounded by eight O²⁻ ions.¹⁴ For the first coordination shell peak of the 4.0 nm HfO₂ PDA film at ~ 1.8 \AA radial distance in Fig. 2(a), Hf-O scattering amplitude and phase were used to model the spectra which correspond to seven nearest neighbor O atoms in *m*-HfO₂. The double-peak feature at ~ 3 \AA radial distance in the calculated *m*-HfO₂ structure in Fig. 1(a) is clearly evident in the 4.0 nm HfO₂ PDA film. Fitting this sample yields a bimodal microstructure fractions of 0.67 monoclinic and 0.33 tetragonal which are consistent with previously reported x-ray diffraction (XRD) plots of 3 nm HfO₂ exposed to the N incorporating NH₃ PDA process.¹⁷ The second shell fitting data in Table I illustrate the influence of HfO₂ thickness on the resultant poly-

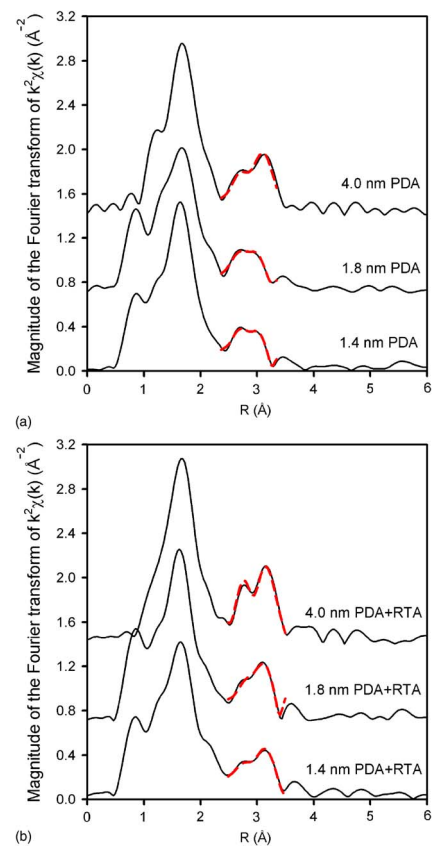


FIG. 2. (Color online) The Fourier transformed EXAFS data for 1.4, 1.8, and 4.0 nm thick HfO₂ (a) following the PDA process of 700 °C for 60 s in NH₃ ambient and (b) following the PDA (700 °C for 60 s in NH₃) + RTA (1000 °C for 10 s in N₂ ambient) process.

morph, and no match with *m*-HfO₂ was detected for the 1.4 and 1.8 nm HfO₂ PDA films which fit *t*-HfO₂ with only a slight fraction of orthorhombic. It has been reported that the high pressure (~ 2 – 15 GPa) *o*-HfO₂ polymorph (cotunnite-type) could be preserved,^{14,18} however, it does not exist in equilibrium and is rarely encountered within the process parameters associated with transistor thin film fabrication. The second shell peak at ~ 3.0 \AA is sensitive to crystal symmetry variations and is dominated by Hf-Hf backscattering.¹⁹ Multiple scattering from Hf-O-O paths was found to be negligible.

The results of the second shell fitting of FT EXAFS spectra correlated with monoclinic, tetragonal, and orthorhombic polymorphs as a function of thickness and anneal processing are indicated in Table I and overlaid on the second shell of the FT spectra shown in Figs. 2(a) and 2(b). Figure 2(a) illustrates a tetragonal to monoclinic transformation (for films annealed at the same temperature and pressure) corresponding to film thickness (grain size dependence). This transformation is martensitic in nature, whereby the change in crystal structure is diffusionless, achieved by a homogeneous deformation of the parent phase into the product phase.^{20,21} It has been shown that HfO₂ is strongly anisotropic in thermal expansion, with the *b* axis exhibiting negligible expansion, while the expansion is substantial for the *a* and *c* axes and the phase transition is athermal, occurring over a temperature range, i.e., the amount of the transformed phase varies with change in temperature but not as a function of time at a particular temperature.¹⁴ Figure 2(b) illustrates the FT EXAFS spectra of HfO₂ PDA+RTA (1000 °C for 10 s in N₂ ambient) films. The 4.0 nm HfO₂ PDA+RTA

TABLE I. Second shell fittings of Fourier transformed EXAFS spectra of HfO₂ indicates correlation with monoclinic, tetragonal, and orthorhombic phases as a function of thickness and anneal processing. The uncertainties in the near-neighbor distances and the coordination numbers are ± 0.01 and ± 0.1 Å, respectively.

Sample ID	Second shell fits					
	Monoclinic ($P2_1/c$)		Tetragonal ($P4_2/nmc$)		Orthorhombic ($Pbca$)	
	$R_{\text{Hf-Hf}}$ (Å)	$N_{\text{Hf-Hf}}$	$R_{\text{Hf-Hf}}$ (Å)	$N_{\text{Hf-Hf}}$	$R_{\text{Hf-Hf}}$ (Å)	$N_{\text{Hf-Hf}}$
1.4 nm PDA			3.38	10.3	3.44	5.4
1.8 nm PDA			3.57	10.9	3.44	5.7
4.0 nm PDA	3.51	6.5	3.7	9.8		
1.4 nm PDA+RTA	3.49	6.5	3.68	10.2		
1.8 nm PDA+RTA	3.47	6.2	3.65	9.9		
4.0 nm PDA+RTA	3.37	6.7	3.64	9.6		

sample was nearly identical to a measured *m*-HfO₂ reference powder sample (that was fitted as *m*-HfO₂ with a volume fraction of 1.0). The Hf-Hf coordination increases (greater degree of crystallization) and the monoclinic fraction increases with HfO₂ thickness. There is a critical grain size (~ 4 – 10 nm) for the tetragonal to monoclinic transformation, below which the tetragonal phase may be retained at temperatures sufficient to produce *m*-HfO₂ in slightly thicker films with slightly larger grains.²²

Although XRD (short-range order limit of ~ 4 – 5 nm) and high resolution transmission electron microscopy images in plan view and cross section suggest that the 1.4 nm HfO₂ samples of Figs. 2(a) and 2(b) remain amorphous following RTA, EXAFS data in Table II clearly identify evidence of *t*-HfO₂ for the PDA sample and coexistence of *m*-HfO₂ (57%) and *t*-HfO₂ (43%) for the PDA+RTA sample. A plausible explanation is that the short-range order identified in the EXAFS spectra corresponds to a HfO₂ grain size that is below the detection sensitivity of XRD. This indicates that the volume fraction of order and the grain size decrease with decreasing film thickness due to surface energy effects which are consistent with the retention of *t*-HfO₂ following PDA. Although the monoclinic phase is suppressed for the 1.4 and 1.8 nm HfO₂ PDA samples, each thickness exhibits evidence of *m*-HfO₂ following PDA+RTA annealing, such that the fraction of monoclinic phase increases with film thickness. The fraction of each polymorph corresponding to annealed HfO₂ films was accurately determined from the second shell fitting and is listed in Table II.

Consider the electrical performance of aggressively scaled nanocrystalline HfO₂ gate dielectric films. There is an appreciable advantage in retaining the higher permittivity tetragonal phase in terms of further equivalent oxide thickness (EOT) scaling, given the current minimum thickness limit of

TABLE II. Volume fractions of monoclinic, tetragonal, and orthorhombic HfO₂ as a function of thickness and anneal processing.

	Fraction		
	Monoclinic	Tetragonal	Orthorhombic
1.4 nm PDA	0	0.96	0.04
1.8 nm PDA	0	0.99	0.01
4.0 nm PDA	0.67	0.33	0
1.4 nm PDA+RTA	0.57	0.43	0
1.8 nm PDA+RTA	0.6	0.4	0
4.0 nm PDA+RTA	0.96	0.04	0

~ 1.2 nm for deposition of a continuous layer. Furthermore, since the ratio of *t*-HfO₂ ($k=28$ – 29) to *m*-HfO₂ ($k=16$ – 18) is shown to increase as the film is scaled thinner, more accurate assessment of the contribution from the interfacial SiO_x layer to the composite EOT is achievable. Now that characterization of capacitor and transistor devices comprised of sub-2-nm high-*k* dielectric films has become commonplace, it is critical to assume the appropriate fraction of *t*-HfO₂ that will directly lead to a more precise estimate of the permittivity contribution and corresponding quality of the SiO_x interface between the HfO₂ and the Si substrate ($k=3.9$ for stoichiometric SiO₂ and $k=11.9$ for α -Si) and avoid possible exaggeration of oxygen deficiency due to an overestimate of the SiO_x *k* value.

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