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Program and Abstracts

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Electron Attachment to Excited Molecules¹

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The interactions of slow electrons with molecules --especially the processes of electron attachment-- depend rather strongly on the internal energy content of the molecules themselves. As a rule, excited molecules interact with slow electrons with substantially larger cross sections than do ground-state molecules.¹⁻⁵ Studies of electron attachment to vibrationally/rotationally excited, "hot", molecules^{1,3,5-18} and especially to electronically excited molecules^{1,3,5,19-24} are rather recent, in contrast to the extensive studies on electron attachment to ground-state molecules which cover many decades.^{6,25,26}

Recent studies on electron attachment to vibrationally/rotationally excited (thermally by infrared laser radiation) molecules, showed that the effect of internal energy of a molecule on its electron attachment properties depends on the mode --dissociative or nondissociative-- of electron attachment. They quantified the effect of the internal energy of the molecule on the rate of destruction (by autodissociation or by autodetachment) of its parent transient anion. General increases in ro-vibrational molecular energy increase the cross section for dissociative attachment (mainly due to faster dissociation and the capturing of lower-energy electrons)^{3,6,8,9,10} and decrease the effective cross section for parent anion formation (mainly due to increased autodetachment).^{3,13-15} These findings will be discussed.

Recent studies of electron attachment to electronically excited molecules, especially photoenhanced dissociative electron attachment to short- and long-lived excited electronic states of molecules produced directly or indirectly by laser irradiation^{3,20-24} will be discussed also.

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These studies show that the cross sections for dissociative electron attachment to electronically excited molecules can be orders of magnitude larger than those for the ground-state molecules. The larger polarizabilities of the excited states are believed to be partly responsible for these large cross sections, as is indicated by the data on electron scattering from excited atoms^{1,2,4,5} and molecules.^{1,5}

The new techniques that have been developed for such studies will be briefly described and the implications^{23,26,27} of this new knowledge for new technologies (e.g., ultrasensitive analytical methods, plasma processing of materials, negative ion sources) will be indicated.

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